

Isotopic Abundances and Atomic Weights of the Elements

Cite as: Journal of Physical and Chemical Reference Data **13**, 809 (1984); <https://doi.org/10.1063/1.555720>
Published Online: 15 October 2009

Paul De Biévre, Marc Gallet, Norman E. Holden, and I. Lynus Barnes



[View Online](#)



[Export Citation](#)

ARTICLES YOU MAY BE INTERESTED IN

[Isotopic Compositions of the Elements, 2001](#)

Journal of Physical and Chemical Reference Data **34**, 57 (2005); <https://doi.org/10.1063/1.1836764>

[Atomic Weights of the Elements 1999](#)

Journal of Physical and Chemical Reference Data **30**, 701 (2001); <https://doi.org/10.1063/1.1395055>

[Representative Equations for the Thermal Conductivity of Water Substance](#)

Journal of Physical and Chemical Reference Data **13**, 893 (1984); <https://doi.org/10.1063/1.555718>

Where in the world is AIP Publishing?
Find out where we are exhibiting next

AIP Publishing

Isotopic Abundances and Atomic Weights of the Elements

Paul De Bièvre and Marc Gallet

Central Bureau for Nuclear Measurements, Joint Research Centre, Commission of the European Communities,
Geel, Belgium

3M20
1986
1984

Norman E. Holden

National Nuclear Data Center, Brookhaven National Laboratory, Upton, NY 11973, USA

and

I. Lynus Barnes

National Measurement Laboratory, National Bureau of Standards, Gaithersburg, MD 20899, USA

A large number of measurements describing the isotopic composition of the elements using a variety of analytical methods have been reported since the discovery of the first isotope in 1912. During the past several decades, however, mass spectrometric methods have been used, almost exclusively, to determine the isotopic composition, and thus the atomic weights, of the elements. This evaluated compilation reports the literature references for all complete mass spectrometric measurements published during the period 1920 through 1983. Also given are the isotopic compositions, the isotope ratios, the atomic weights calculated from the data, the appropriate nuclidic masses and an evaluation of the errors of the measurements. For each polynuclidic element, a best measurement has been selected.

Key words: atomic weights; chemical elements; elements; isotopes; isotopic composition; isotope ratios; mass spectrometry.

Contents

1. Introduction	810	Carbon	818
2. Explanation of Tables	811	Cerium	863
3. Acknowledgments	812	Chlorine	827
4. References	885	Chromium	834
4.1. Commission Reports	891	Cobalt	836
		Copper	838
		Dysprosium	868
		Erbium	869
		Europium	866
		Fluorine	822
		Gadolinium	866
		Gallium	840
		Germanium	841
		Gold	876
		Hafnium	872
		Helium	813
		Holmium	869
		Hydrogen	813
		Indium	856
		Iodine	859
		Iridium	875
		Iron	836
		Krypton	844
		Lanthanum	862
		Lead	878
		Lithium	814

List of Tables

Aluminum	825
Antimony	857
Argon	828
Arsenic	842
Barium	861
Beryllium	816
Bismuth	883
Boron	816
Bromine	844
Cadmium	855
Caesium	860
Calcium	830

©1984 by the U.S. Secretary of Commerce on behalf of the United States. This copyright is assigned to the American Institute of Physics and the American Chemical Society.
Reprints available from ACS; see Reprint List at back of issue.

Carbon	818
Cerium	863
Chlorine	827
Chromium	834
Cobalt	836
Copper	838
Dysprosium	868
Erbium	869
Europium	866
Fluorine	822
Gadolinium	866
Gallium	840
Germanium	841
Gold	876
Hafnium	872
Helium	813
Holmium	869
Hydrogen	813
Indium	856
Iodine	859
Iridium	875
Iron	836
Krypton	844
Lanthanum	862
Lead	878
Lithium	814

Lutetium	872	Selenium	843
Magnesium	823	Silicon	825
Manganese	835	Silver	853
Mercury	877	Sodium	823
Molybdenum	850	Strontium	847
Neodymium	864	Sulfur	826
Neon	822	Tantalum	873
Nickel	837	Tellurium	858
Niobium	849	Terbium	867
Nitrogen	821	Thallium	878
Osmium	875	Thorium	883
Oxygen	821	Thulium	870
Palladium	833	Tin	836
Phosphorus	826	Titanium	832
Platinum	876	Tungsten	874
Potassium	828	Uranium	884
Praseodymium	863	Vanadium	833
Rhenium	874	Xenon	859
Rhodium	852	Ytterbium	871
Rubidium	846	Yttrium	848
Ruthenium	851	Zinc	840
Samarium	865	Zirconium	848
Scandium	831		

1. Introduction

The first discovery of a stable "isotope" of an element, ^{22}Ne , by J.J. Thomson in November 1912 started an era of more refined knowledge of the nature of the elements. This period continued until the report of the discovery of ^{180}Ta by White, Collins and Rourke in 1955, the most recently reported new isotope. Tantalum had been previously believed to be mononuclidian with ^{181}Ta the only stable nuclide.

A variety of measurements describing the terrestrial isotopic composition of the elements has been published, both in the open and in more restricted literature since Thomson's discovery. However, it soon became apparent that the reported measurements frequently did not agree. In some cases, real variations in the natural isotopic composition were reported and have subsequently been verified. In other cases, unjustified high accuracy was claimed, which led to the conclusion that real isotopic variations existed. More recent work has shown that these conclusions were frequently erroneous and that no significant variations in the natural isotopic composition exist in many cases, at least at current measurement precision.

Since there is now a wide scientific interest in isotopic composition and natural isotopic measurements, the authors thought it useful to identify and assemble isotopic abundance literature for all the elements. Much of this literature is not readily available from libraries or in journals, either because it was published only as internal reports of various types, or the data were released from previously classified documents years after the mea-

surements were performed.

This document is an outgrowth of three preliminary unpublished reports prepared in 1963 and 1964 at the Central Bureau of Nuclear Measurements (Joint Research Centre of the Commission of the European Communities), Geel, Belgium. The data presented here were compiled from literature sources there, and from literature files at the General Electric Knolls Atomic Power Laboratory, Schenectady, New York; the Brookhaven National Laboratory, Upton, New York; and the National Bureau of Standards, Gaithersburg, Maryland.

The present evaluated compilation is intended to provide all published data, along with the appropriate literature references, on isotopic abundances which have been reported in the open and available literature. It is limited to data obtained by mass spectrometric methods since, in general, these methods provide measurements of higher precision and accuracy than any other methods available at this time. The literature covered was published from 1940 to the end of 1983. References to literature published prior to 1940 have, in some cases, been included but, primarily due to instrumental limitations existent in the period 1923–1940 much of that work was considered of lesser value for the purposes of this compilation. In general, literature has only been included for those cases where complete isotopic abundances (or alternatively, all isotopic ratios) have been reported. Much of the available literature in which only variations of individual isotopic ratios have been reported, has been omitted. In most of these cases, variations reported versus some reference material or ratio are not complete and make the calculation of abundances or of meaningful atomic weight values difficult if not impossible.

2. Explanation of the Tables

The elements are listed in the tables in order of increasing atomic number and using the names of the elements, in English, as recommended by the International Union of Pure and Applied Chemistry. For each element, in the line below the name and number of the element is listed the atomic weight ($A_r(E)$) recommended by the International Commission on Atomic Weights (now called the IUPAC Commission on Atomic Weights and Isotopic Abundances) in 1961 when these values were recalculated from the oxygen to the carbon scale. If the recommended value or the uncertainty ($U_r(E)$) has changed in the period 1961 to 1983, the changed value and the year of the recommended change are also listed. The references to each of the complete Reports of the Commission during that time are given at the end of this compilation.

In the tables, the appropriate nuclidic masses (in unified atomic mass units, u) for each stable isotope are listed (A.H. Wapstra and K. Bos, *Atomic Data and Nuclear Data Tables*, **19**, 175 (1977).) The uncertainties for these are not repeated here since, in general, the nuclidic masses are known to a precision much greater than that of the corresponding isotopic abundance and, except for the mononuclidic elements, the uncertainty on the nuclidic mass is of lesser importance. The reader is cautioned however, that the levels of precision reached in the past few years in the determination of isotopic abundance values have approached the point where the nuclidic mass uncertainties may no longer be neglected and the most recent nuclidic mass evaluation should be consulted.

The references to the published data are given in chronological order for each element. The isotopic abundances and isotopic ratios are given for each reference. The data are given as presented in the original literature (however, see below). The user should be aware that, in some cases, the isotopic abundances given do not total to 100 percent. Where other information was not given so that corrections could be made (e.g. where isotopic ratios were not published) the data have not been corrected but are given as published. Uncertainties assigned by the original authors to the data are given in brackets and are applicable to the last digits. Where no uncertainties are indicated, none was given in the original literature and those authors are assumed to have intended an error of ± 1 on the last digit given (see discussion below.)

The authors of this compilation initially decided it would be of use to the majority of readers if both the isotopic composition (isotope abundances in atom percent) and the isotope ratios were presented. Further it was decided that, for the purpose of comparison, it would be useful if the ratios were given with a common base isotope for each element. It should be noted that data were presented in a variety of ways in the original literature. In some cases only abundances were given, in others the isotopic ratios were given but different reference isotopes were used from paper to paper. The necessary calculations have been made to present here both isotopic

abundances and isotopic ratios using a common reference isotope. The choice of the base or common isotope was, in many cases, arbitrary although some general guidelines were used. For example, for systems of two isotopes only, the heavier isotope was frequently chosen as the denominator. For systems of more than two isotopes, a moderately abundant (a few percent) isotope near the center of mass for the element was chosen, however, in many cases, there have been other reasons that the majority of authors of data have chosen a common isotope to report the data and we have accepted that choice. To calculate the atomic weight for each entry, it was most convenient to use a computer program developed by one of the authors (ILB) which requires as input the isotopic abundance and the error on the abundance, thus it was necessary to calculate the abundances and the associated errors on the abundances in each case where that was not given in the literature. A small "c" annotating the uncertainty on an abundance indicates that we have calculated that uncertainty from that given for an isotopic ratio in the literature.

Next the atomic weight and its uncertainty are given as calculated from the abundances and the nuclidic masses. In the cases mentioned above, where isotopic abundances do not total to 100 percent, the atomic weight given has been normalized using the simple factor of the difference between the totals given and 100 percent. The indicated uncertainties for the mononuclidic elements were obtained by multiplying the uncertainties given by Wapstra and Bos for the appropriate nuclide by a factor of six. We believe that this gives an uncertainty as consistent as possible with those given for the polynuclidic elements.

An indication of the type of uncertainty quoted in the original literature is given below the atomic weight value as follows:

"NS" is used to indicate that an error value was not stated in the original literature.

"SD", "2XSD", and "3XSD" indicates that the author stated the errors to be 1, 2, or 3 standard deviations.

"P" indicates probable error (as defined by the author).

"SE" indicates standard error.

In the final row of the table a "C" has been placed if the measurement is known to be one which is calibrated with the use of separated isotopes thus becoming an "absolute" measurement. In a few cases the "C" has been appended where the effects of measurement fractionation have been removed by the use of the "double spike" technique, and where this was judged to have been done with particular care in the preparation and calibration of the spike solutions. A "B" has been added to the measurement which has been judged to be the best measurement from a natural source. The designation of "best measurement" was generally done on the basis of a calibrated measurement or a double spiked measurement. If

neither were available for an element, the most precise measurement was chosen from the group of published literature which gave sufficient detail of the measurement process for the authors to judge that reasonable care had been taken to eliminate the more common sources of error. As a result, the reader should note that the best measurement is not necessarily a good one.

Finally, in the last column (Ref. 83ICA1) the isotopic abundances are given as recommended by the IUPAC Subcommittee on the Assessment of Isotopic Composition of the Commission on Atomic Weights and Isotopic Abundances as "representative" for the element. In some cases, where it is known that the isotopic abundances for an element do not vary in nature and a calibrated measurement is available, the representative composition is the same as the best measurement. In other cases, the representative compositions are given with less precision, either because no calibrated measurements are available, or the possibility of small changes in isotopic composition exists. In general, however, these may be taken as the composition which might be expected in an average bottle of reagent chemicals. Nevertheless, the user should be aware that materials are available in commerce where the isotopic composition may vary considerably from these values (this is particularly true of the elements B, Li, and U). If more accurate values are needed, the reader is urged to obtain and use specially analyzed samples. The reader is referred to a recent publication by Peiser et al. for additional information on representative isotopic compositions and the effects of these on atomic weights (H.S. Peiser, I.L. Barnes, P.J. De Bièvre, J.R. De Laeter, R. Hagemann, N.E. Holden, T.J. Murphy, E. Roth, M. Shima, and H.G. Thode, "Element by Element Review of their Atomic Weights", *Pure Appl. Chem.* **56**, 696 (1984).

The last section of this publication contains the complete reference to the data. The reference system used was originally developed by one of the authors (NEH) for the General Electric Wall Chart of the Nuclides and is

based on using the last two digits of the year of publication, followed by the first three letters of the first author's last (family) name and is followed by a single digit serial number.

The authors hope that this evaluated compilation will be of help to the scientist interested in the isotopic abundance of the elements and will help save the tedious effort of searching the literature, particularly since many of the original citations are no longer readily available. Additional information from those references may be obtained from the authors.

As mentioned above, this compilation was developed over a number of years. Every effort has been made to make it as complete as possible but it is inevitable that the authors have, inadvertently, missed some important references. We would be most grateful if readers would draw these to our attention. We would also appreciate receiving copies of reprints of papers that might be included in a future update of this compilation.

3. Acknowledgments

Many helpful discussions with the author's colleagues on the Commission on Atomic Weights and Isotopic Abundances and, especially, those who served on the Subcommission for Isotopic Abundances have added to the completeness of this compilation and their assistance and encouragement is gratefully acknowledged. Much appreciation is due to H.S. Peiser, T.J. Murphy, N.N. Greenwood, and E.R. Cohen whose careful reading of the manuscript and thoughtful suggestions have helped to make it more readable and useful. We owe a special debt of gratitude to the late A.E. "Gus" Cameron who was always ready to help and who offered freely from his vast knowledge of the field. We also thank Ann Lawrence, Teresa Sperow, Gelene Hensley, and Joy Shoemaker whose skills and patience in the preparation of the manuscript made it possible.

Element $_1\text{H}$ Hydrogen

1961 1.00797 1969 1.0080 (3) 1971 1.0079 1981 1.00794 (7)

Mass no.	Nuclidic mass	36HAL1	38MOR1	38VOS1
1	1.007825037	99.9844 (5) ^c	99.9839 (5) ^c	99.98508 (1) ^c
2	2.014101787	0.0156 (5) ^c	0.0161 (5) ^c	0.01492 (11) ^c
Isotope ratio 2/1		0.0001563 (42)	0.0001613 (52)	0.00014922
Atomic weight		1.007982 (5)	1.007987 (5)	1.0079752 (1)
Error		NS	NS	NS
Annotation				

Mass no.	Nuclidic mass	39SWA1	51KIR1	51KIR1	54CLA1
1	1.007825037	99.9855	99.9861 (1)	99.9847 (1)	99.9848 (3)
2	2.014101787	0.0145	0.0139 (1)	0.0153 (1)	0.0152 (3)
Isotope ratio 2/1		0.0001449	0.00013902	0.00015302	0.0001520
Atomic weight		1.007971 (1)	1.007965 (1)	1.007979 (1)	1.007978 (3)
Error		NS	SD	SD	SD
Annotation					

Mass no.	Nuclidic mass	60HOR1	60HOR1	70HAG1	83ICA1
1	1.007825037	99.98511 (5)	99.98531 (5)	99.984426 (5) ^c	99.985 (1)
2	2.014101787	0.01489 (5)	0.01469 (5)	0.015574 (5) ^c	0.015 (1) (for water only)
Isotope ratio 2/1		0.00014892	0.00014692	0.00015576 (5)	0.000150
Atomic weight		1.0079749 (5)	1.0079729 (5)	1.00798176 (5)	1.00798 (1)
Error		P	P	2XSD	
Annotation				C, B	

Element $_2\text{He}$ Helium

1961 4.0026 1969 4.00260 1983 4.002602 (2)

Mass no.	Nuclidic mass	34VAU1	46ALD1	46ALD1	47FAI1
3	3.016029297	0.00286	0.000016 (4) ^c	0.00013 (13) ^c	0.00012
4	4.00260325	99.997	99.999984 (4) ^c	99.999987 (13) ^c	99.99988
Isotope ratio 3/4		2.9×10^{-5}	1.6×10^{-6} (4)	1.3×10^{-6} (3)	1.2×10^{-6}
Atomic weight		4.0025694 (1)	4.00260310 (4)	4.0026020 (3)	4.0026021 (1)
Error		NS	SD	SD	NS
Annotation					

Element $_2\text{He}$ Helium--continued

Mass no.	Nuclidic mass	47FAI1	48ALD1	70MAM1	76CLA1
3	3.016029297	0.00013	0.000120 (12) ^c	0.0001399 (13)	0.0001384 (6)
4	4.00260325	99.99987	99.999880 (12) ^c	99.9998601 (13)	99.9998616 (6)
Isotope ratio 3/4		1.3×10^{-6}	1.2×10^{-6} (2)	1.399×10^{-6} (13)	1.384×10^{-6} (6)
Atomic weight		4.0026020 (1)	4.0026021 (1)	4.00260188 (1)	4.002601895 (6)
Error		NS	SD	SD	SD
Annotation					B

Mass no.	Nuclidic mass	84ICA1
3	3.016029297	0.000138 (3)
4	4.00260325	99.999862 (3) (for air only)
Isotope ratio 3/4		1.380×10^{-6}
Atomic weight		4.00260190 (3)
Error		
Annotation		

Element $_3\text{Li}$ Lithium

	1961	6.939	1969	6.941 (3)	1983	6.941 (2)
Mass no.	Nuclidic mass		32AST1		47ING3	
6	6.0151232		8.33		7.386 (22) ^c	7.407 (33) ^c
7	7.0160045		91.67		92.614 (22) ^c	92.593 (33) ^c
Isotope ratio 7/6			11		12.54 (4)	12.50 (6)
Atomic weight			6.9326 (1)		6.9421 (2)	6.9419 (3)
Error			NS		NS	SD
Annotation						

Mass no.	Nuclidic mass	56ORD1	56ORD1	56PAC1	56WHI1
6	6.0151232	7.413 (11) ^c	7.418 (11) ^c	7.519 (57)	8.12
7	7.0160045	92.587 (11) ^c	92.582 (11) ^c	92.481 (57)	91.88
Isotope ratio 7/6		12.49 (2)	12.48 (2)	12.3 (1)	11.32
Atomic weight		6.9418 (1)	6.9418 (1)	6.9407 (6)	6.9347 (1)
Error		NS	NS	NS	NS
Annotation					

Element ${}^3\text{Li}$ Lithium--continued

Mass no.	Nuclidic mass	580MU1	58PAL1	58PAL1	600MU1
6	6.0151232	7.418 (28) ^c	7.4239 (55) ^c	7.479 (28) ^c	7.418 (22) ^c
7	7.0160045	92.582 (28) ^c	92.5761 (55) ^c	92.521 (28) ^c	92.582 (22) ^c
Isotope ratio 7/6		12.48 (5)	12.47 (1)	12.37 (5)	12.48 (5)
Atomic weight		6.9418 (3)	6.94180 (5)	6.9411 (3)	6.9418 (2)
Error		NS	NS	3XSD	NS
Annotation					

Mass No.	Nuclidic mass	600MU1	600MU1	60PAL1	62PUP1
6	6.0151232	7.605 (10) ^c	7.629 (10) ^c	7.4239 (55)	7.490 (17) ^c
7	7.0160045	92.395 (10) ^c	92.371 (10) ^c	92.5761 (55)	92.510 (17) ^c
Isotope ratio 7/6		12.15 (1)	12.108 (10)	12.47	12.35 (3)
Atomic weight		6.9399 (1)	6.9396 (1)	6.94170 (5)	6.9410 (2)
Error		NS	NS	3XSD	SE
Annotation					

Mass no.	Nuclidic mass	62TAN2	62TAN2	62TAN2	63SHI4
6	6.0151232	7.474 (10) ^c	7.474 (17) ^c	7.468 (10) ^c	7.69 (12) ^c
7	7.0160045	92.526 (10) ^c	92.526 (17) ^c	92.532 (10) ^c	92.31 (12) ^c
Isotope ratio 7/6		12.38 (1)	12.38 (3)	12.39 (1)	12.0 (2)
Atomic weight		6.9412 (1)	6.9412 (2)	6.9413 (1)	6.939 (1)
Error		SE	SE	SE	SD
Annotation					

Mass no.	Nuclidic mass	64KRA1	65SVE1	66SHI2	66SHI2
6	6.0151232	7.587 (46) ^c	7.563 (28) ^c	7.63 (10) ^c	7.46 (17) ^c
7	7.0160045	92.413 (46) ^c	92.437 (28) ^c	92.37 (10) ^c	92.54 (17) ^c
Isotope ratio 7/6		12.18 (6)	12.22 (5)	12.1 (1)	12.4 (3)
Atomic weight		6.9401 (5)	6.9403 (3)	6.940 (1)	6.941 (2)
Error		3XSD	SD	SD	SD
Annotation		C			

Element ${}_3\text{Li}$ Lithium--continued

Mass no.	Nuclidic mass	73FLE1	77BR01	83MIC1	84ICAI
6	6.0151232	7.68 (2) ^c	7.602 (37)	7.525 (29)	7.5 (2)
7	7.0160045	92.32 (2) ^c	92.398 (37)	92.475 (29)	92.5 (2)
Isotope ratio 7/6		12.02 (3)	12.15	12.29 (5)	12.3
Atomic weight		6.9391 (2)	6.9399 (4)	6.9407 (3)	6.941 (2)
Error		SD	2XSD	2XSD	
Annotation		C		C, B	

Element ${}_4\text{Be}$ Beryllium

1961	9.0122	1969	9.01218		
Mass no.	Nuclidic mass		3/NIE2	63LEI1	84ICAI
9	9.0121825	100	100	100	100
Isotope ratio		—	—	—	—
Atomic weight		9.012182 (2)	9.012182 (2)	9.012182 (2)	9.012182 (2)
Error		NS	NS		
Annotation			B		

Element ${}_5\text{B}$ Boron

1961	10.811 (3)	1969	10.81	1983	10.811 (5)
Mass no.	Nuclidic mass		46ING1	48TH01	48TH01
10	10.0129380	18.83 (2)	18.98	18.45	19.57
11	11.0093053	81.17 (2)	81.02	81.55	80.43
Isotope ratio 11/10		4.31	4.27	4.42	4.11
Atomic weight		10.8217 (2)	10.8202 (1)	10.8255 (1)	10.8143 (1)
Error		NS	NS	NS	NS
Annotation					

Mass no.	Nuclidic Mass	56SHI1	56SHI1	58BEN1	58PAL1
10	10.0129380	19.61	19.05	19.3 (1)	19.65
11	11.0093053	80.39	80.95	80.7 (1)	80.35
Isotope ratio 11/10		4.10	4.25	4.18	4.09
Atomic weight		10.8139 (1)	10.8195 (1)	10.817 (1)	10.8135 (1)
Error		NS	NS	NS	NS
Annotation					

Element ${}^5\text{B}$ Boron--continued

Mass no.	Nuclidic Mass	60BEN1	61FIN1	61FIN1	61GOR1
10	10.0129380	19.27 (13)	19.92	19.72	19.83
11	11.0093053	80.73 (13)	80.08	80.28	80.17
Isotope ratio 11/10		4.09	4.019	4.070	4.044
Atomic weight		10.8117 (1)	10.8108 (1)	10.8128 (1)	10.8117 (1)
Error		3XSD	SD	SD	2XSD
<u>Annotation</u>					
Mass no.	Nuclidic mass	61GOR1	61GOR1	61MCM1	61MCM1
10	10.0129380	19.93	20.00	19.84	19.84
11	11.0093053	80.07	80.00	80.16	80.16
Isotope ratio 11/10		4.017	4.000	4.040	4.040
Atomic weight		10.8107 (1)	10.8100 (1)	10.8116 (1)	10.8116 (1)
Error		2XSD	2XSD	NS	NS
<u>Annotation</u>					
Mass no.	Nuclidic mass	61MCM1	61MCM1	63SHI3	63BIE1
10	10.0129380	19.73	19.72	19.80	19.82 (3)
11	11.0093053	80.27	80.28	80.20	80.18 (3)
Isotope ratio 11/10		4.068	4.072	4.05	4.045
Atomic weight		10.8127 (1)	10.8128 (1)	10.8120 (1)	10.8118 (3)
Error		NS	NS	NS	SD
<u>Annotation</u>					
Mass no.	Nuclidic mass	63BIE1	68AGY1	68AGY1	69BIE1
10	10.0129380	20.14 (2)	19.58 (3)	20.05 (3)	19.824 (20)
11	11.0093053	79.86 (2)	80.42 (3)	79.95 (3)	80.176 (20)
Isotope ratio 11/10		3.965	4.108 (8)	3.987 (8)	4.0444 (52)
Atomic weight		10.8086 (2)	10.8142 (3)	10.8095 (3)	10.8118 (2)
Error		SD	2XSE	2XSE	3XSD
<u>Annotation</u>					

Element $^{10}_5\text{B}$ Boron--continued

Mass no.	Nuclidic mass	70CAT1	71GEN1	73NOM1	73TAM1
10	10.0129380	19.827 (13)	19.9 (2) ^c	19.85 (1) ^c	19.83 (3) ^c
11	11.0093053	80.173 (13)	80.1 (2) ^c	80.15 (1) ^c	80.17 (3) ^c
Isotope ratio 11/10		4.0436 (33)	4.03 (4)	4.039 (2)	4.042 (5)
Atomic weight		10.8118 (2)	10.811 (2)	10.8115 (1)	10.8117 (3)
Error		3XSD	SD	SD	SD
Annotation		C,B			

Mass no.	Nuclidic mass	79KAN1	83ICA1
10	10.0129380	19.74 (2)	19.9 (2)
11	11.0093053	80.26 (2)	80.1 (2)
Isotope ratio 11/10		4.066 (4)	4.025
Atomic weight		10.8126 (2)	10.810 (2)
Error		SD	
Annotation			

Element $^{12}_6\text{C}$ Carbon

Mass no.	Nuclidic mass	1961 12.01115 (5)	1969 12.011	37BR01	39NIE3	39NIE3	41MUR2
12	12.	98.927 (43) ^c		98.866 (23) ^c	98.900 (22) ^c	98.891 (22) ^c	
13	13.003354839	1.073 (43) ^c		1.134 (23) ^c	1.100 (22) ^c	1.109 (22) ^c	
Isotope ratio 12/13		92.2 (3.7)		87.2 (1.8)	89.9 (1.8)	89.2 (1.8)	
Atomic weight		12.0108 (4)		12.0114 (2)	12.0110 (2)	12.0111 (2)	
Error		NS		SD	SD	SD	NS
Annotation							

Mass no.	Nuclidic mass	41MUR2	48RAN1	48RAN1	48RAN2
12	12.	98.937 (21) ^c	98.9350 (52)	98.8814 (55)	98.881
13	13.003354839	1.063 (21) ^c	1.0650 (52)	1.1186 (55)	1.119
Isotope ratio 12/13		93.7 (1.8)	92.9 (5)	88.4 (5)	88.4
Atomic weight		12.0107 (2)	12.01069 (5)	12.01122 (6)	12.01123 (1)
Error		NS	NS	NS	NS
Annotation					

Element ^{12}C Carbon--continued

Mass no.	Nuclidic mass	48RAN2	50TR01	50BEC1	50NIE1
12	12.	98.937	98.891	98.876 (5)	98.892 (4)
13	13.003354839	1.063	1.109	1.124 (5)	1.108 (4)
Isotope ratio 12/13		93.1	89.17	87.97	89.25
Atomic weight		12.01067 (1)	12.01113 (1)	12.01128 (5)	12.01112 (4)
Error		NS	NS	NS	P
Annotation					

Mass no.	Nuclidic mass	51MAR1	51MAR1	51WIC1	51WIC1
12	12.	98.930 (11) ^c	98.878 (11) ^c	98.8787 (13) ^c	98.881 (12) ^c
13	13.003354839	1.070 (11) ^c	1.122 (11) ^c	1.1213 (13) ^c	1.119 (12) ^c
Isotope ratio 12/13		92.5 (1.0)	88.1 (1.0)	88.18 (10)	88.36 (10)
Atomic weight		12.0107 (1)	12.01113 (1)	12.01125 (11)	12.0112 (1)
Error		NS	NS	NS	NS
Annotation					

Mass no.	Nuclidic mass	51WIC1	51WIC2	51WIC2	52DIB1
12	12.	98.8824 (12) ^c	98.9095 (36) ^c	98.8777 (38) ^c	98.9130 (59) ^c
13	13.003354839	1.1176 (12) ^c	1.0905 (36) ^c	1.1223 (38) ^c	1.0870 (59) ^c
Isotope ratio 12/13		88.48 (10)	90.7 (3)	88.1 (3)	91.0 (5)
Atomic weight		12.01121 (1)	12.01094 (4)	12.01126 (4)	12.01091 (6)
Error		NS	NS	NS	2XSD
Annotation					

Mass no.	Nuclidic mass	52WIC1	52WIC1	53CRA1	53CRA1
12	12.	98.9181 (12) ^c	98.8912 (12) ^c	98.87678 (25) ^c	98.92658 (23) ^c
13	13.003354839	1.0819 (12) ^c	1.1088 (12) ^c	1.12322 (25) ^c	1.07342 (23) ^c
Isotope ratio 12/13		91.43 (10)	89.19 (10)	88.03 (2)	92.16 (2)
Atomic weight		12.01086 (1)	12.01112 (1)	12.011270 (2)	12.010770 (2)
Error		NS	NS	P	P
Annotation					

Element $^{12}_{\text{C}}$ Carbon--continued

Mass no.	Nuclidic Mass	53DAN1	53DAN1	53DAN1	53DAN1
12	12.	98.88691 (74) ^c	98.9116 (71) ^c	98.8982 (74) ^c	98.8885 (76) ^c
13	13.003354839	1.11309 (74) ^c	1.0884 (71) ^c	1.1018 (74) ^c	1.1115 (76) ^c
Isotope ratio 12/13		88.84 (6)	90.88 (6)	88.76 (6)	87.97 (6)
Atomic weight		12.011168 (7)	12.0109201 (7)	12.01105 (7)	12.01115 (8)
Error		NS	NS	NS	NS
Annotation					

Mass no.	Nuclidic mass	53WIC1	53WIC1	55LAN1	55LAN1
12	12.	98.9128 (12) ^c	98.8746 (13) ^c	98.9143 (14) ^c	98.8953 (20) ^c
13	13.003354839	1.0872 (12) ^c	1.1254 (13) ^c	1.0857 (14) ^c	1.1047 (20) ^c
Isotope ratio 12/13		90.98 (10)	87.86 (10)	91.11 (12)	89.52 (16)
Atomic weight		12.01091 (1)	12.01129 (1)	12.01089 (1)	12.01108 (2)
Error		2XSD	2XSD	SD	SD
Annotation					

Mass no.	Nuclidic mass	57CRA1	57GAV1	57GAV1
12	12.	98.889 (3) ^c	98.91892 (58) ^c	98.90074 (50) ^c
13	13.003354839	1.111 (3) ^c	1.08108 (58) ^c	1.09926 (60) ^c
Isotope ratio 12/13		89.05 (27)	91.50 (5)	89.97 (5)
Atomic weight		12.01115 (3)	12.010847 (6)	12.011029 (6)
Error		P	NS	NS
Annotation		B		

Mass No.	Nuclidic Mass	83ICA1
12	12.	98.90 (3)
13	13.003354839	1.10 (3)
Isotope Ratio 12/13		89.91
Atomic Weight		12.0110 (3)
Error		
Annotation		

Element $_{\gamma}N$ Nitrogen

1961 14.0067

Mass no.	Nuclidic mass	50NIE1	58JUN1	63PIL1	63PIL1
14	14.003074008	99.6350 (13) ^c	99.63370 (40) ^c	99.6366 (53) ^c	99.6361
15	15.000108978	0.3650 (13) ^c	0.36630 (40) ^c	0.3634 (53) ^c	0.3639
Isotope ratio 14/15		273 (1)	272.0 (3)	274.1 (4)	273.8 (4)
Atomic weight		14.00671 (1)	14.006726 (4)	14.006697 (5)	14.00670 (5)
Error		P	NS	NS	NS
Annotation			C, B		

Mass no.	Nuclidic mass	63PIL1	83ICA1
14	14.003074008	99.6353 (5) ^c	99.634 (9)
15	15.000108978	0.3647 (5) ^c	0.366 (9)
Isotope ratio 14/15		273.2 (4)	272.2
Atomic weight		14.006710 (5)	14.00672 (9)
Error		NS	
Annotation			

Element $_{\delta}O$ Oxygen

1961 15.9994 1969 15.9994 (3)

Mass no.	Nuclidic mass	41MUR1	44TH01	49HIB1	49HIB2
16	15.99491464	99.7598 (74) ^c	99.7574 (37) ^c	99.770 (5)	99.775 (6)
17	16.9991306	0.0407 (20) ^c	0.03920 (80) ^c	0.035 (9)	0.035 (8)
18	17.99915939	0.1995 (60) ^c	0.2034 (6) ^c	0.196 (10)	0.190 (10)
Isotope ratio 16/18, 17/18		500. (15) 0.204 (8)	490.4 (8.6) 0.193 (5)	509 0.179	525 0.184
Atomic weight		15.9993 (1)	15.99938 (5)	15.99919 (3)	15.99907 (3)
Error		NS	NS	SD	SD
Annotation					

Element ${}_{\text{8}}^{\text{O}}$ OXYGEN--continued

Mass no.	Nuclidic mass	50NIE1	50NIE1	76BAE1	83ICA1
16	15.99491464	99.75769 (50) ^c	99.75873 (70) ^c	99.7628 (5) ^c	99.762 (15)
17	16.9991306	0.03766 (10) ^c	0.03736 (50) ^c	0.0372 (4) ^c	0.038 (3)
18	17.99915939	0.20465 (50) ^c	0.20391 (50) ^c	0.200045 (5)	0.200 (12)
Isotope ratio 16/18, 17/18		487.44 (58) 0.1840 (3)	489.24 (59) 0.1832 (12)	498.7 (2.2) 0.1860	498.8 0.190
Atomic weight		15.99939 (1)	15.99938 (1)	15.999296 (6)	15.9993 (3)
Error		P	P	SD	
Annotation				C, B	

Element ${}_{\text{9}}^{\text{F}}$ Fluorine

	1961 18.9984	1971 18.99840	1975 18.998403	
Mass no.	Nuclidic mass	20AST1	83ICA1	
19	18.99840325	100	100	
Isotope ratio		—	—	
Atomic weight		18.9984032 (8)	18.9984032 (8)	
Error		NS		
Annotation				

Element ${}_{\text{10}}^{\text{Ne}}$ Neon

	1961 20.183	1967 20.179 (3)	1979 20.179	
Mass no.	Nuclidic mass	47DIB1	49HIB2	50NIE2
20	19.9924391	90.51 (15)	89.99 (3)	90.92 (4)
21	20.9938453	0.29 (?)	0.30 (1)	0.257 (1)
22	21.9913837	9.21 (18)	9.72 (1)	8.82 (14)
Isotope ratio 21/20, 22/20		0.0032 0.1018	0.0033 0.1080	0.002827 (6) 0.09703 (40)
Atomic weight		20.179 (3)	20.1897 (4)	20.1713 (8)
Error		SD	SD	3XP
Annotation				SD

Element ${}_{10}^{20}\text{Ne}$ Neon--continued

Mass no.	Nuclidic mass	65EBE1	66WAL1	71MEL1	83ICA1
20	19.9924391	90.50 (7)	90.514 (31)	90.512 (8)	90.51 (9)
21	20.9938453	0.268 (2)	0.266 (5)	0.267 (1)	0.27 (2)
22	21.9913837	9.23 (7)	9.220 (29)	9.221 (6)	9.22 (9)
Isotope ratio 21/20, 22/20		0.002959 (22) 0.10204 (83)	0.002935 (58) 0.10187 (38)	0.00295 0.10188	0.00298 0.1019 (for air only)
Atomic weights		20.179 (1)	20.1794 (6)	20.1794 (6)	20.179 (2)
Error		3XSE	NS	NS	
Annotation		C	C, B		

Element ${}_{11}^{22}\text{Na}$ Sodium

1961 22.9898 1971 22.98977

Mass no.	Nuclidic mass	56WHI1	83TCA1
23	22.9897697	100	100
Isotope ratio			
Atomic weight		22.989770 (5)	22.989770 (5)
Error		NS	
Annotation			

Element ${}_{12}^{24}\text{Mg}$ Magnesium

1961 24.312 1967 24.305

Mass no.	Nuclidic mass	48HIB1	48HIB1	48WHI1	56WHI1
24	23.9850450	78.98 (4)	78.97 (3)	78.60 (13)	78.8 (2)
25	24.9858392	10.05 (2)	10.01 (2)	10.11 (5)	10.15 (1)
26	25.9825954	10.97 (4)	11.02 (4)	11.29 (8)	11.06 (1)
Isotope ratio 25/24, 26/24		0.1273 0.1389	0.1268 0.1397	0.1286 0.1436	0.1288 0.1404
Atomic weight		24.3048 (8)	24.3054 (7)	24.312 (2)	24.308 (3)
Error		SE	SE	SD	NS
Annotation					

Element ^{12}Mg Magnesium--continued

Mass no.	Nuclidic mass	600MUT	64SHI2	64SHI2	66CAT1
24	23.9850450	78.60 (4)	78.907 (36) ^c	78.945 (40) ^c	78.992 (25)
25	24.9858392	10.15 (3)	10.059 (36) ^c	10.029 (36) ^c	10.003 (9)
26	25.9825954	11.31 (4)	11.034 (22) ^c	11.026 (28) ^c	11.005 (19)
Isotope ratio 25/24, 26/24		.1291 .1439	0.12748 (5) 0.13984 (5)	0.12704 (5) 0.13966 (4)	0.12663 (13) 0.13932 (26)
Atomic weight		24.3123 (7)	24.3061 (5)	24.3057 (6)	24.3050 (4)
Error		NS	NS	NS	3XSD
Annotation					C, B

Mass no.	Nuclidic mass	67TAK1	70SCH1	74LEE1	77LEE1
24	23.9850450	79.12 (8) ^c	78.962 (20) ^c	74.963 (20) ^c	78.962 (20) ^c
25	24.9858392	9.99 (7) ^c	9.999 (10) ^c	9.999 (10) ^c	9.999 (10) ^c
26	25.9825954	10.99 (7) ^c	11.039 (10) ^c	11.038 (11) ^c	11.039 (10) ^c
Isotope ratio 25/24, 26/24		0.1261 (8) 0.1389 (6)	0.12663 0.139805 (13)	0.12663 0.139805 (13)	0.12663 0.139805 (13)
Atomic weight		24.329 (1)	24.3056 (3)	24.3056 (3)	24.3056 (3)
Error		SE	2XSD	2XSE	2XSE
Annotation					

Mass no.	Nuclidic mass	/9ESAT	83ICAT
24	23.9850450	78.962	78.99 (3)
25	24.9858392	9.999	10.00 (1)
26	25.9825954	11.039	11.01 (2)
Isotope ratio 25/24, 25/24		0.12663 0.139805	0.12660 0.13938
Atomic weight		24.3056 (3)	24.3051 (5)
Error			
Annotation		NS	

Element ^{13}Al Aluminum

1961 26.9815 1971 26.98154

Mass no.	Nuclidic mass	56WHI1	83ICA1
27	26.9815413	100	100
Isotope ratio			
Atomic weight			
Error			
Annotation			

Element ^{14}Si Silicon

1961 28.086 1969 28.086 (3) 1975 28.0855 (3)

Mass no.	Nuclidic mass	46ING1	46NEY1	46WIL1	48WHI1
28	27.9769284	92.28 (8)	92.24 (10)	92.268 (19) ^c	92.16 (6)
29	28.9764964	4.67 (5)	4.69 (5)	4.678 (18) ^c	4.71 (3)
30	29.9737717	3.05 (3)	3.07 (5)	3.0541 (90) ^c	3.13 (4)
Isotope ratio					
29/28	0.0506	0.0508	0.0507 (2)	0.0511	
30/28	0.0330	0.0333	0.0331 (1)	0.0340	
Atomic weight					
Error					
Annotation					
Mass no.	Nuclidic mass	49HIB2	52DIB1	52NOR1	53REY1
28	27.9769284	92.19 (6)	92.41 (3)	92.14 (20) ^c	92.18 (3)
29	28.9764964	4.70 (3)	4.57 (5)	4.73 (10)	4.71 (2)
30	29.9737717	3.12 (4)	3.01 (5)	3.13 (10)	3.12 (2)
Isotope ratio					
29/28	0.0510	0.0494	0.0513	0.0511	
30/28	0.0338	0.0326	0.0340	0.0338	
Atomic weight					
Error					
Annotation					

Element ^{14}Si Silicon--continued

Mass no.	Nuclidic mass	75BAR2	83ICA1
28	27.9769284	92.22933 (155)	92.23 (1)
29	28.9764964	4.66982 (124)	4.67 (1)
30	29.9737717	3.10085 (74)	3.10 (1)
Isotope ratio 29/28 30/28		0.050633 0.033621	0.05063 0.03361
Atomic weight		28.08553 (2)	28.0855 (2)
Error		3XSD	
Annotation		C, B	

Element ^{15}P Phosphorus

1961	30.9738	1971	30.97376
Mass no.	Nuclidic mass	63LEI1	83ICA1
31	30.9737634	100	100
Isotope ratio			
Atomic weight		30.973763 (4)	30.973763 (4)
Error		NS	
Annotation			

Element ^{16}S Sulfur

1961	32.064 (3)	1969	32.06	1983	32.066 (6)
Mass no.	Nuclidic mass	38NIE2	49HER1	49TR01	
32	31.9720718	95.061 (83) ^c	94.84	94.89	
33	32.9714591	0.741 (15) ^c	0.84	0.759	
34	33.96786774	4.183 (82) ^c	4.26	4.34	
36	35.9670790	0.01521 (15) ^c	0.05	0.0136	
Isotope ratio 33/32 34/32 36/32		0.0078 (2) 0.044 (1) 0.000166 (16)	0.009 0.045 0.0005	0.008 0.0456 0.0001	
Atomic weight		32.064 (2)	32.0675 (4)	32.0668 (2)	
Error		NS	NS	NS	
Annotation					

Element ^{16}S Sulfur--continued

Mass no.	Nuclidic Mass	50HER1	50MAC1	56BRA1	83ICA1
32	31.9720718	95.00 (3)	95.018	95.0	95.02 (9)
33	32.9714591	0.74 (2)	0.750	0.760 (4)	0.75 (1)
34	33.96786774	4.24 (2)	4.215	4.22 (1)	4.21 (8)
36	35.9670790	0.017 (3)	0.017	0.014	0.02 (1)
Isotope ratio 33/32	0.0078	0.0079	0.008	0.008	
34/32	0.0446	0.0444	0.044	0.044	
36/32	0.00018	0.00018	0.0001	0.0002	
Atomic weight	32.0648 (4)	32.06437 (4)	32.0663 (2)	32.064 (2)	
Error	SD	P	NS		
Annotation		B			

Element ^{17}Cl Chlorine

1961 35.453

Mass no.	Nuclidic mass	36NIE1	55BOY1	550WE1	61MEY1
35	34.968852729	75.4 (2) ^c	75.529 (16)	75.79 (18) ^c	75.80 (6)
37	36.965902624	24.6 (2) ^c	24.471 (16)	24.21 (18) ^c	24.20 (6)
Isotope ratio 35/37	3.07 (3)	3.0865 (27)	3.13 (3)	3.132	
Atomic weight	35.460 (4)	35.4575 (3)	35.452 (4)	35.452 (1)	
Error	NS	SD	NS	SE	
Annotation					

Mass no.	Nuclidic mass	62SHI2	83ICA1
35	34.968852729	75.7705 (450)	75.77 (5)
37	36.965902624	24.2295 (450)	24.23 (5)
Isotope ratio 35/37	$3.1272^{+0.0079}_{-0.0082}$	3.1271	
Atomic weight	35.4527 (9)	35.453 (1)	
Error	3XSD		
Annotation	C,B		

Element ^{18}Ar Argon

	1961	39.948	1969	39.948	1979	39.948	
Mass no.	Nuclidic mass	47DIB1		50NIE1		50NIE1	71MEL1
36	35.967545605	0.35 (1)		0.33645 (60) ^c		0.33327 (60) ^c	0.339 (1)
38	37.9627322	0.08 (1)		0.06325 (10) ^c		0.06275 (10) ^c	0.064 (1)
40	39.9623831	99.57 (3)		99.60030 (60) ^c		99.60397 (60) ^c	99.597 (1)
Isotope ratio 36/40 38/40		0.0035 0.0008		0.003378 (6) 0.000635 (1)		0.003346 (6) 0.000630 (1)	0.00340 0.00064
Atomic Weight		39.9468 (9)		39.94768 (2)		39.94781 (2)	39.94756 (3)
Error		SD		P		P	NS
Annotation				C, B		C	

Mass no.	Nuclidic mass	83ICA1
36	35.967545605	0.337 (3)
38	37.9627322	0.063 (1)
40	39.9623831	99.600 (3) (for air only)
Isotope ratio 36/40 38/40		
Atomic weight		39.9477 (1)
Error		
Annotation		

Element ^{19}K Potassium

	1961	39.102	1969	39.102 (3)	1921	39.098 (3)	1975	39.0983 (3)	1979	39.0983
Mass no.	Nuclidic mass		35BRE1		35NIE1		48HIB1		50NIE1	
39	38.9637079		93.43 (2) ^c		93.31 (1) ^c		93.25 (7)		93.081 (4)	
40	39.9639988		0.0112 (2) ^c		0.0108 (11) ^c		0.010 (2)		0.0119 (1)	
41	40.9618254		6.56 (2) ^c		6.68 (1) ^c		6.75 (7)		6.91 (4)	
Isotope ratio 39/41 40/41			14.25 (3) 0.00171 (2)		13.96 (1) 0.00161 (16)		13.81 0.0015		13.48 (7) 0.001725 (17)	
Atomic weight			39.0949 (4)		39.0973 (2)		39.099 (1)		39.1019 (6)	
Error			NS		NS		SE		P	
Annotation										

Element ^{19}K Potassium--continued

Mass no.	Nuclidic mass	52REU1	56REU1	56WHI1	60KEN1
39	38.9637079	93.462 (21) ^c	93.126 (42) ^c	93.23 (5)	93.219 (14) ^c
40	39.9639988	0.0118 (1)	0.011730 (47) ^c	0.0118 (2)	0.011750 (65) ^c
41	40.9618254	6.526 (21) ^c	6.862 (43) ^c	6.76 (5)	6.770 (14) ^c
Isotope ratio 39/41 40/41		14.32 (5) 0.00181 (1)	13.57 (9) 0.001710 (2)	13.79 0.00175	13.77 (3) 0.00174 (1)
Atomic weight		39.0939 (4)	39.1009 (8)	39.099 (1)	39.0991 (3)
Error		SE	P	NS	P

Annotation

Mass no.	Nuclidic mass	600MU1	62HAR1	62STA1	68SHI1
39	38.9637079	93.19 (2)	93.46 (5)	93.423 (43) ^c	93.28 (10)
40	39.9639988	0.011 (1)	0.0115 (1)	0.01162 (6) ^c	0.0117 (2)
41	40.9618254	6.78 (2)	6.52 (6)	6.565 (43) ^c	6.70 (8)
Isotope ratio 39/41 40/41		13.74 0.001622	14.33 0.00176	14.23 (10) 0.00177 (2)	13.92 0.001746
Atomic weight		39.0993 (4)	39.094 (1)	39.0949 (8)	39.097 (2)
Error		NS	NS	NS	SD

Annotation

Mass no.	Nuclidic mass	75GARI1	75IMA1	80IMA1	83ICA1
39	38.9637079	93.2581 (29)	93.29 (10) ^c	93.27 (30) ^c	93.2581 (30)
40	39.9639988	0.01167 (4)	0.0131 (2) ^c	0.01157 (6) ^c	0.0117 (1)
41	40.9618254	6.7302 (29)	6.69 (8) ^c	6.72 (3) ^c	6.7302 (30)
Isotope ratio 39/41 40/41		13.8566 (63) 0.0017343 (61)	13.94 (2) 0.001955 (27)	13.877 (45) 0.001722 (9)	13.857 0.001738
Atomic weight		39.09829 (6)	39.095 (2)	39.098 (4)	39.09830 (6)
Error		3xSD	SD	SD	
Annotation		C, B			

Element $^{20}_{\text{Ca}}$ Calcium

1961 40.08 1983 40.078 (4)

Mass no.	Nuclidic mass	38NIE2	48WHI1	600MUL	62STA1
40	39.9625907	96.961 (60)	96.92 (3)	96.89 (4)	96.959 (13) ^c
42	41.9586218	0.640 (19)	0.64 (1)	0.66 (1)	0.6485 (41) ^c
43	42.9587704	0.1454 (48)	0.132 (4)	0.18 (1)	0.13280 (69) ^c
44	43.9554848	2.065 (57)	2.13 (4)	2.03 (1)	2.0718 (86) ^c
46	45.953689	0.00330 (48)	0.0032	0.0023 (2)	0.003150 (34) ^c
48	47.952532	0.1852 (58)	0.179 (1)	0.23 (2)	0.1850 (11) ^c
Isotope ratio 40/44	46.954	45.50	47.73	46.8 (2)	
42/44	0.310	0.30	0.325	0.313 (1.5)	
43/44	0.070	0.06	0.0887	0.0641 (2)	
45/44	0.0016	0.0015	0.00113	0.00152 (1.5)	
48/44	0.090	0.08	0.113	0.0893 (4)	
Atomic weight	40.077 (2)	40.079 (2)	40.081 (2)	40.0773 (3)	
Error	NS	SD	NS	NS	
Annotation					
Mass no.	Nuclidic mass	64BAC1	68SHI1	71COL1	72M001
40	39.9625907	96.88 (5)	96.87 (99) ^c	96.8918 (200) ^c	96.941 (1) ^c
42	41.9586218	0.655 (6)	0.660 (8) ^c	0.6562 (19) ^c	0.647 (1) ^c
43	42.9587704	0.138 (2)	0.136 (2) ^c	0.1312 (13) ^c	0.135 (1) ^c
44	43.9554848	2.12 (4)	2.133 (4) ^c	2.1202 (5) ^c	2.086 (1) ^c
46	45.953689	0.0046 (10)	0.0034 (1) ^c	0.0034 (4) ^c	0.004 (1) ^c
48	47.952532	0.200 (6)	0.199 (3) ^c	0.1972 (40) ^c	0.187 (1) ^c
Isotope ratio 40/44	45.70	45.41 (60)	45.70	46.480 (87)	
42/44	0.309	0.3096 (37)	0.3095 (9)	0.3104 (11)	
43/44	0.065	0.0637 (9)	0.0619 (6)	0.0648 (9)	
46/44	0.002	0.00160 (2)	0.0016 (2)	0.0017 (5)	
48/44	0.094	0.0935 (12)	0.093 (2)	0.0898 (6)	
Atomic Weight	40.080 (2)	40.081 (1)	40.0802 (3)	40.0780 (1)	
Error	SD	SD	NS	2XSD	
Annotation				B	

Element $_{20}^{40}$ Ca Calcium--continued

Mass no.	Nuclidic mass	78RUS1	80ROS1	83ICA1
40	39.9625907	96.98213 (617) ^c	96.980 (1)	96.941 (13)
42	41.9586218	0.64214 (4) ^c	0.648 (1)	0.647 (3)
43	42.9587704	0.13340 (2) ^c	0.135 (1)	0.135 (3)
44	43.9554848	2.05675 (13) ^c	2.095 (1)	2.086 (5)
46	45.953689	0.00313 (2) ^c	0.003 (1)	0.004 (3)
48	47.952532	0.18245 (4) ^c	0.189 (1)	0.187 (3)
Isotope ratio 40/44	47.153 (3)	46.266 (28)	46.472	
42/44	0.31221 (2)	0.3094 (4)	0.31016	
43/44	0.06486 (1)	0.0644 (5)	0.06472	
46/44	0.00152 (1)	0.0015 (4)	0.00192	
48/44	0.08871 (2)	0.0901 (5)	0.08965	
Atomic weight	40.07629 (1)	40.0785 (1)	40.0780 (4)	
Error	2XSE	2XSD		
Annotation				

Element $_{21}^{45}$ Sc Scandium

1961	44.956	1969	44.9559	1983	44.95591
Mass no.	Nuclidic mass		50LEL1		83IAC1
45	44.9559136		100		100
Isotope ratio					
Atomic weight		44.955914 (9)		44.955914 (9)	
Error			NS		
Annotation					

Element ^{22}Ti Titanium

	1961	47.90	1969	47.90 (3)	1979	47.88 (3)	
Mass no.	Nuclidic mass		38NIE2		49HER1		52MAT1
46	45.9526327		7.95 (15) ^c		8.22 (13)		7.92 (7)
47	46.9517649		7.76 (14) ^c		7.42 (5)		7.50 (7)
48	47.9479467		73.45 (20) ^c		73.38 (11)		73.09 (13)
49	48.9478705		5.51 (11) ^c		5.56 (4)		5.90 (13)
50	49.9447858		5.34 (11) ^c		5.41 (5)		5.59 (11)
Isotope ratio 46/48		0.1082 (22)		0.1120		0.1084	0.1065
47/48		0.1056 (21)		0.1011		0.1026	0.0981
49/48		0.0750 (15)		0.0758		0.0807	0.0752
50/48		0.0727 (14)		0.0737		0.0765	0.0735
Atomic weight		47.873 (4)		47.874 (3)		47.886 (3)	47.8827 (3)
Error		P		2XSD		2XSD	NS
Annotation							
Mass no.	Nuclidic mass		54HOG1		58DRA1		68BEL1
46	45.9526327		7.99 (2)		8.00 (5)		8.24 (45)
47	46.9517649		7.32 (2)		7.29 (4)		7.44 (22)
48	47.9479467		73.99 (7)		73.98 (8)		73.71 (18)
49	48.9478705		5.46 (2)		5.38 (5)		5.43 (16)
50	49.9447858		5.25 (5)		5.35 (4)		5.18 (31)
Isotope ratio 46/48		0.1080		0.1081		0.1118	0.10858
47/48		0.0989		0.0985		0.1009	0.09943
49/48		0.0738		0.0727		0.0737	0.07442
50/48		0.0710		0.0723		0.0703	0.07258 (38)
Atomic weight		47.875 (1)		47.876 (1)		47.87 (1)	47.87633 (9)
Error		SD		SE		SD	3XSE
Annotation						C, B	

Element ^{22}Ti Titanium--continued

Mass no.	Nuclidic mass	80NIE1	81NIE1	83ICA1
46	45.9526327	8.0124	7.9957	8.0 (1)
47	46.9517649	7.3309	7.3159	7.3 (1)
48	47.9479467	73.8145	73.6765	73.8 (1)
49	48.9478705	5.4964	5.5228	5.5 (1)
50	49.9447858	5.3458	5.4891	5.4 (1)
Isotope ratio 46/48	0.108548	0.10952	0.1111	
47/48	0.099315 (5)	0.09930	0.1003	
49/48	0.074463 (4)	0.07496	0.0732	
50/48	0.072422 (4)	0.07450	0.0705	
Atomic weight	47.87675 (1)	47.880360 (3)	47.878 (3)	
Error	2XSD	NS		
Annotation				

Element ^{23}V Vanadium

	1961	50.942	1969	50.9414 (3)	1977	50.9415	
Mass no.		Nuclidic mass		49HES1	49HES1	49HES1	49HES1
50		49.9471613		0.273 (4)	0.27 (3) ^c	0.27 (3) ^c	0.255 (4)
51		50.9439625		99.727 (4)	99.73 (3) ^c	99.73 (3) ^c	99.745 (4)
Isotope ratio 50/51				0.00274 (4)	0.00271 (27)	0.00268 (26)	0.00256 (4)
Atomic weight				50.94124 (4)	50.9413 (3)	50.9413 (3)	50.94142 (4)
Error				SE	SE	SE	SE
Annotation							

Mass no.	Nuclidic mass	49HES1	49LEL2	50HER1	56WHI1
50	49.9471613	0.253 (50) ^c	0.23 (1)	0.28 (2)	0.25 (1)
51	50.9439625	99.747 (50) ^c	99.77 (1)	99.72 (2)	99.75 (1)
Isotope ratio 50/51		0.00254 (5)	0.00230	0.00281	0.00251
Atomic weight		50.94209 (5)	50.9417 (1)	50.9412 (2)	50.9415 (1)
Error		SE	NS	2XSD	NS
Annotation					

Element ^{23}V Vanadium--continued

Mass no.	Nuclidic mass	63SVE1	64FLE1	66FLE1	68IMA1	69BAL1
50	49.9471613	0.2497 (12) ^c	0.2497 (12) ^c	0.2497 (6) ^c	.250 (2) ^c	0.2419 (30) ^c
51	50.9439625	99.7503 (12) ^c	99.7503 (12) ^c	99.7503 (6) ^c	99.750 (2) ^c	99.7581 (30) ^c
Isotope ratio 50/51	0.002503 (12)	0.002503 (10)	0.002503 (6)	0.002506 (38)	0.002425 (30)	
Atomic weight	50.94147 (1)	50.94147 (1)	50.941473 (6)	50.94147 (2)	50.94155 (3)	
Error		NS	SE	SE	SD	NS
Annotation		C, B	C			

Mass no.	Nuclidic mass	70PEL1	80IMA1	83ICA1
50	49.9471613	0.2444 (17) ^c	.2500 (3) ^c	0.250 (2)
51	50.9439625	99.756 (17) ^c	99.7500 (3) ^c	99.750 (2)
Isotope ratio 50/51	0.002450 (17)	0.002506 (12)	0.002506	
Atomic weight	50.94173 (2)	50.941470 (3)	50.94147 (2)	
Error		NS	SD	
Annotation				

Element ^{24}Cr Chromium

Mass no.	Nuclidic mass	1961 51.996	1983 51.9961 (6)	39NIE2	48ING4	48WH11	49HIB2	60FLE1
50	49.9460463	4.49	4.51 (8)	4.31 (4)	4.41 (6)	4.352 (24)		
52	51.9405097	83.78	83.52 (30)	83.76 (14)	83.46 (11)	83.764 (36)		
53	52.9406510	9.43	9.55 (20)	9.55 (9)	9.54 (6)	9.509 (27)		
54	53.9388822	2.30	2.42 (4)	2.38 (2)	2.61 (9)	2.375 (18)		
Isotope ratio 50/52	0.0536	0.0540	0.0515	0.0528	0.0520			
53/52	0.1126	0.1143	0.1140	0.1143	0.1135			
54/52	0.0275	0.0290	0.0284	0.0313	0.0283			
Atomic weight	51.9912 (3)	51.994 (3)	51.998 (1)	52.000 (2)	51.9959 (3)			
Error		NS	NS	SD	SD	3XSD		
Annotation								

Element $^{24}_{\text{Cr}}$ Chromium--continued

Mass no.	Nuclidic Mass	600MU1	62SVE1	66SHI1	66SHI3	73BAR1
50	49.9460463	4.37 (3)	4.357 (5)	4.3452 (85)	4.354 (15)	4.34 (1) ^c
52	51.9405097	83.86 (8)	83.760 (9)	83.7895 (117)	83.803 (32)	83.80 (1) ^c
53	52.9406510	9.44 (9)	9.508 (7)	9.5006 (110)	9.507 (27)	9.50 (1) ^c
54	53.9388822	2.36 (3)	2.375 (5)	2.3647 (48)	2.336 (8)	2.36 (1) ^c
Isotope ratio 50/52	0.0521		0.0520	0.051859 (100)	0.05196	0.05186 (13)
53/52	0.1126		0.1135	0.113386 (145)	0.1134	0.11339 (14)
54/52	0.0281		0.0284	0.028222 (48)	0.02787	0.01822 (7)
Atomic weight		51.995 (1)	51.9962 (2)	51.9961 (2)	51.9954 (4)	51.9961 (3)
Error		NS	SE	3XSD	2XSD	2XSD
Annotation			C	C, B		

Mass no.	Nuclidic mass	75TAM1	83ICA1
50	49.9460463	4.353 (4) ^c	4.345 (9)
52	51.9405097	83.815 (35) ^c	83.789 (12)
53	52.9406510	9.479 (33) ^c	9.501 (11)
54	53.9388822	2.353 (16) ^c	2.365 (5)
Isotope ratio 50/52	0.05194 (5)	0.0519	
53/52	0.1131 (4)	0.1134	
54/52	0.02807 (19)	0.0282	
Atomic weight		51.9955 (4)	51.9961 (2)
Error		SD	
Annotation			

Element $^{25}_{\text{Mn}}$ Manganese

1961 54.9380

Mass no.	Nuclidic mass	63LEI1	83ICA1
55	54.9380463	100	100
Isotope ratio			
Atomic weight		54.93805 (1)	54.93805 (1)
Error		NS	
Annotation			

Element ^{26}Fe Iron

1961 55.847 (3)

Mass no.	Nuclidic mass	39NIE2	41VAL1	47VAL1	48WHI1
54	53.9396121	6.04	5.84	5.81 (1) ^c	5.81 (1)
56	55.9349393	91.57	91.68	91.75 (1) ^c	91.64 (2)
57	56.9353957	2.11	2.17	2.15 (1) ^c	2.21 (1)
58	57.9332778	0.28	0.31	0.29 (1) ^c	0.34 (1)
Isotope ratio 54/56	0.066	0.0631	0.0634	0.0634	0.0634
57/56	0.023	0.0237	0.0234	0.0241	0.0241
58/56	0.003	0.0034	0.0032	0.0037	0.0037
Atomic weight	55.8411 (3)	55.8463 (3)	55.8463 (3)	55.8479 (3)	55.8479 (3)
Error	NS	NS	SE	SE	SE
Annotation			B		

Mass no.	Nuclidic mass	49HIB2	64CHE2	65SHI2	65SHI2	83ICA1
54	53.9396121	5.903 (15)	5.83 (5)	5.773 (61) ^c	5.818 (52) ^c	5.8 (1)
56	55.9349393	91.52 (2)	91.75 (5)	91.785 (65) ^c	91.760 (54) ^c	91.72 (30)
57	56.9353957	2.245 (11)	2.14 (2)	2.139 (27) ^c	2.129 (18) ^c	2.2 (1)
58	57.9332778	0.335 (3)	0.28 (1)	0.3029 (92) ^c	0.2936 (92) ^c	0.28 (1)
Isotope ratio 54/56	0.0645	0.0635	0.0629 (7)	0.0634 (6)	0.0632	
57/56	0.0245	0.0233	0.0233 (3)	0.0232 (2)	0.0240	
58/56	0.0037	0.0035	0.0033 (1)	0.0032 (1)	0.0033	
Atomic weight	55.8463 (3)	55.846 (1)	55.847 (1)	55.846 (1)	55.847 (3)	
Error	SD	NS	NS	NS	NS	
Annotation						

Element ^{27}Co Cobalt

1961 58.9332

Mass no.	Nuclidic mass	63LEI1	83ICA1
59	58.9331978	100	100
Isotope ratio			
Atomic weight	58.93320 (1)	58.93320 (1)	
Error	NS		
Annotation			

Element ^{28}Ni Nickel								
	1961	58.71	1969	58.71 (3)	1973	58.70	1979	58.69
Mass no.	Nuclidic mass	41STR1		41VAL2		44EWA2		48ING4
58	57.9353471	62.85 (81)		67.4		69.18 (69)		67.92 (15)
60	59.9307890	29.51 (88)		26.7		25.82 (52)		26.22 (12)
61	60.9310586	1.70 (13)		1.2		0.97 (3)		1.16 (2)
62	61.9283464	4.66 (28)		3.8		3.28 (6)		3.71 (3)
64	63.9279680	1.27 (10)		0.88		0.75 (3)		0.98 (2)
Isotope ratio 58/60		2.130		2.52		2.679		2.590
61/60		0.058		0.045		0.038		0.044
62/60		0.158		0.142		0.127		0.141
64/60		0.043		0.033		0.029		0.037
Atomic weight		58.84 (2)		58.708 (4)		58.656 (9)		58.700 (2)
Error		NS		NS		P		NS
Annotation								
Mass no.	Nuclidic mass	48WH1		52MAT1		73BAR1		80MOR1
58	57.9353471	67.76 (22)		68.0		68.274 (1) ^c		68.2812
60	59.9307890	26.16 (66)		26.3		26.095 (1) ^c		26.0974
61	60.9310586	1.25 (3)		1.13		1.134 (1) ^c		1.1295 (8) ^c
62	61.9283464	3.66 (1)		3.66		3.593 (1) ^c		3.5892 (10) ^c
64	63.9279680	1.16 (20)		1.01		0.904 (1) ^c		0.9027 (8) ^c
Isotope ratio 58/60		2.590		2.586		2.6164 (26)		2.6164
61/60		0.048		0.043		0.04346 (43)		0.04328 (3)
62/60		0.140		0.139		0.13769 (34)		0.13753 (4)
64/60		0.044		0.038		0.03464 (35)		0.03459 (4)
Atomic weight		58.71 (1)		58.759 (2)		58.68767 (1)		58.68735 (6)
Error		SD		NS		2XSD		NS
Annotation								
B								

Element ^{28}Ni Nickel--continued

Mass no.	Nuclidic mass	82SHI1	83ICA1
58	57.9353471	68.2803	68.27 (1)
60	59.9307890	26.0967	26.10 (1)
61	60.9310586	1.1295	1.13 (1)
62	61.9283464	3.5878	3.59 (1)
64	63.9279680	0.9057	0.91 (1)
Isotope ratio 58/60	2.6164	2.616	
61/60	0.04328	0.043	
62/60	0.13748	0.137	
64/60	0.03471	0.035	
Atomic weight	56.687464 (7)	58.688 (1)	
Error	NS		
Annotation			

Element ^{29}Cu Copper

	1961	63.54	1965	63.546	1969	63.546 (3)		
Mass no.		Nuclidic mass		44EWA1		47BR01	47DUC1	47ING1
63	62.9295992	69.97 (29)		69.09		69.48 (16)	69.078 (38) ^c	
65	64.9277924	30.03 (29)		30.91		30.52 (16)	30.921 (38) ^c	
Isotope ratio 63/65		2.330 (32)		2.235		2.277 (17)	2.234 (4)	
Atomic weight		63.530 (6)		63.5472 (2)		63.539 (3)	63.5468 (8)	
Error		P		NS		P	NS	
Annotation								
Mass no.		Nuclidic mass		48HES2		48ING4	48ING4	48WHI1
63	62.9295992	69.089 (95) ^c		69.43 (10) ^c		69.41 (10)	68.94 (19)	
65	64.9277924	30.911 (95) ^c		30.57 (10) ^c		30.59 (10)	31.06 (19)	
Isotope ratio 63/65		2.235 (10)		2.271 (11)		2.269	2.220	
Atomic weight		63.547 (2)		63.540 (2)		63.541 (2)	63.550 (4)	
Error		NS		NS		NS	SD	
Annotation								

Element $_{29}^{\text{Cu}}$ Copper--continued

Mass no.	Nuclidic mass	50HIB1	51SOM1	58WAL1	58WAL1
63	62.9295992	68.98 (4)	68.94 (26)	68.992 (19) ^c	68.799 (19) ^c
65	64.9277924	31.02 (4)	31.06 (26)	31.008 (19) ^c	31.201 (19) ^c
Isotope ratio 63/65		2.224	2.220	2.225 (13)	2.205 (13)
Atomic weight		63.5494 (8)	63.550 (2)	63.5492 (4)	63.5531 (4)
Error		2XSD	NS	2XSD	2XSD
Annotation					
Mass no.	Nuclidic mass	62WHT1	64SHT1	65SHT1	65SHI1
63	62.9295992	69.23 (19) ^c	69.174 (20)	68.981 (32)	69.244 (31)
65	64.9277924	30.77 (19) ^c	30.826 (20)	31.019 (32)	30.756 (31)
Isotope ratio 63/65		2.25 (2)	2.2440 (2)	2.2238	2.2514
Atomic weight		63.544 (4)	63.5456 (4)	63.5494 (6)	63.5442 (6)
Error		NS	3XSD	3XSD	3XSD
Annotation			C, B		
Mass no.	Nuclidic mass	67KAN1	75MUR1	76MUR1	83ICA1
63	62.9295992	69.325 (98) ^c	69.51 (20) ^c	69.164 (40) ^c	69.17 (2)
65	64.9277924	30.675 (98) ^c	30.49 (20) ^c	30.836 (40) ^c	30.83 (2)
Isotope ratio 63/65		2.260 (8)	2.28 (1)	2.243 (5)	2.244
Atomic weight		63.543 (2)	63.539 (4)	63.5458 (8)	63.5456 (4)
Error		SD	SD	SD	
Annotation					

	Element ^{30}Zn Zinc							
	1961	65.37	1969	65.37 (3)	1971	65.38	1983	65.39 (2)
Mass no.	Nuclidic mass	48HES2		48HES2		48BLE1		49HIB2
64	63.9291454	48.89		48.90		48.89 (17) ^c		48.87 (10)
66	65.9260352	27.82		27.82		27.81 (21) ^c		27.62 (10)
67	66.9271289	4.14		4.17		4.070 (45) ^c		4.12 (9)
68	67.9248458	18.54		18.48		18.61 (16) ^c		18.71 (10)
70	69.9253249	0.617		0.623		0.620 (7) ^c		0.69 (2)
Isotope ratio 64/67	11.81		11.73		12.01 (12)		11.86	
66/67	6.72		6.67		6.83 (7)		6.70	
68/67	4.48		4.43		4.57 (5)		4.54	
70/67	0.15		0.15		0.1500 (15)		0.17	
Atomic weight		65.3865 (3)		65.38156 (3)		65.387 (5)		65.393 (3)
Error		NS		NS		P		SD
Annotation								

Mass no.	Nuclidic mass	590KA1	72ROS1	83ICA1
64	63.9291454	49.77	48.63 (13)	48.6 (3)
66	65.9260352	27.19	27.90 (8)	27.9 (2)
67	66.9271289	4.07	4.10 (3)	4.1 (1)
68	67.9248458	18.54	18.75 (16)	18.8 (4)
70	69.9253249	0.41	0.62 (1)	0.6 (1)
Isotope ratio 64/67	12.23		11.86	11.85
66/67	6.68		6.80	6.80
68/67	4.56		4.57	4.58
70/67	0.10		0.15	0.15
Atomic weight		65.3598 (6)	65.396 (5)	65.40 (1)
Error		NS	2XSD	
Annotation			B	

	Element ^{31}Ga Gallium				
	1961	69.72	1983	69.723 (4)	
Mass no.	Nuclidic mass	48ING3	48ING4	49HIB2	53ANT1
69	68.9255809	60.16 (18) ^c	60.317 (94) ^c	60.00 (7)	60.5 (2)
71	70.9247006	39.84 (18) ^c	39.683 (94) ^c	40.00 (5)	39.5 (2)
Isotope ratio 71/69	0.6622 (50)		0.6579 (26)	0.6667	0.6537 (65)
Atomic weight		69.722 (4)	69.719 (2)	69.725 (1)	69.715 (4)
Error		P	P	SD	P
Annotation					

Element ^{31}Ga Gallium--continued

Mass no.	Nuclidic mass	72LAE1	76LAE1	83ICA1
69	68.9255809	60.093 (18) ^c	60.078 (108) ^c	60.1 (2)
71	70.9247006	39.907 (18) ^c	39.922 (108) ^c	39.9 (2)
Isotope ratio 71/69		0.6641 (5)	0.6645 (30)	0.6639
Atomic weight		69.7234 (4)	69.724 (2)	69.723 (4)
Error		NS	2XSD	
Annotation		C, D		

Element ^{32}Ge Germanium

Mass no.	1961	72.59	1969	72.59 (3)	47HFS1	49HTB1	49HTB1	51GRA1
70	69.9242498		20.55 (17)		20.65 (4)		20.60 (6)	20.45 (2)
72	71.9220800		27.37 (15)		27.43 (2)		27.38 (8)	27.41 (2)
73	72.9234639		7.67 (4)		7.86 (4)		7.83 (6)	7.77 (1)
74	73.9211788		36.74 (27)		36.34 (5)		36.40 (10)	36.58 (2)
76	75.9214027		7.67 (4)		7.72 (1)		7.78 (5)	7.79 (1)
Isotope ratio 70/73		2.679		2.627		2.631		2.632
72/73		3.568		3.490		3.497		3.528
74/73		4.790		4.623		4.649		4.708
76/73		1.000		0.982		0.994		1.003
Atomic weight			72.630 (6)		72.623 (1)		72.628 (3)	72.6341 (7)
Error			NS		SE		SE	SD
Annotation								

Mass no.	Nuclidic mass	51CRA1	52DIB1	53REY1	63SHI2
70	69.9242498	20.38 (2)	20.64 (20)	20.52 (17)	20.807 (34) ^c
72	71.9220800	27.37 (2)	27.50 (30)	27.43 (21)	27.591 (64) ^c
73	72.9234639	7.78 (1)	7.72 (8)	7.76 (8)	7.615 (35) ^c
74	73.9211788	36.65 (2)	36.43 (30)	36.54 (23)	36.538 (95) ^c
76	75.9214027	7.82 (1)	7.71 (8)	7.76 (8)	7.449 (38) ^c
Isotope ratio 70/73		2.619	2.674	2.644	2.732
72/73		3.518	3.562	3.535	3.623
74/73		4.711	4.719	4.709	4.798
76/73		1.005	0.999	1.000	0.978
Atomic weight		72.6382 (7)	72.624 (7)	72.631 (6)	72.611 (2)
Error		SD	P	P	SD
Annotation				B	

Element ^{32}Ge Germanium--continued

Mass no.	Nuclidic mass	63SHI2	63SHI2	83ICA1
70	69.9242498	20.92	21.106 (12) ^c	20.5 (5)
72	71.9220800	27.58	27.670 (27) ^c	27.4 (6)
73	72.9234639	7.78	7.683 (12) ^c	7.8 (2)
74	73.9211788	36.20	36.091 (29) ^c	36.5 (7)
76	75.9214027	7.51	7.450 (11) ^c	7.8 (2)
Isotope ratio	70/73	2.689	2.747	2.63
	72/73	3.545	3.601	3.51
	74/73	4.653	4.698	4.68
	76/73	0.965	0.970	1.00
Atomic weight		72.6062 (5)	72.5968 (6)	72.63 (2)
Error		SD	SD	
Annotation				

Element ^{33}As Arsenic

Mass no.	Nuclidic mass	63LEI1	83ICA1
75	74.9215955	100	100
Isotope ratio			
Atomic weight		74.92160 (2)	74.92160 (2)
Error		NS	
Annotation			

Element ^{34}Se Selenium

1961 78.96 1969 78.96 (3)

Mass no.	Nuclidic mass	48WHI1	48WHI1	48WHI1	49HIB2
74	73.9224771	0.86 (13)	0.88 (1)	0.87 (1)	0.96 (3)
76	75.9192066	9.08 (9)	8.95 (3)	9.02 (7)	9.12 (3)
77	76.9199077	7.51 (75)	7.65 (3)	7.58 (7)	7.50 (14)
78	77.9173040	23.54 (24)	23.51 (11)	23.52 (2)	23.61 (5)
80	79.9165205	50.02 (50)	49.62 (14)	49.82 (20)	49.96 (21)
82	81.916709	8.99 (9)	9.39 (9)	9.19 (20)	8.84 (8)
Isotope ratio	74/78 76/78 77/78 80/78 82/78	0.037 0.386 0.319 2.125 0.382	0.037 0.381 0.325 2.111 0.399	0.037 0.384 0.322 2.118 0.391	0.041 0.386 0.318 2.116 0.374
Atomic weight		78.99 (2)	78.995 (3)	78.990 (7)	78.974 (6)
Error		SE	SE	SE	SD
Annotation		B			

Mass no.	Nuclidic mass	83ICA1
74	73.9224771	0.9 (1)
76	75.9192066	9.0 (2)
77	76.9199077	7.6 (2)
78	77.9173040	23.6 (6)
80	79.9165205	49.7 (7)
82	81.916709	9.2 (5)
Isotope ratio	74/78 76/78 77/78 80/78 82/78	0.038 0.383 0.323 2.111 0.400
Atomic weight		78.99 (2)
Error		
Annotation		

Element ^{35}Br Bromine

1961 79.909 (2) 1965 79.904

Mass no.	Nuclidic mass	36BLE1	46WIL1	48WHI1	49HIB2
79	78.9183361	50.6 (6)	50.53 (10) ^c	50.51 (50)	50.57 (7)
81	80.916290	49.4 (6)	49.47 (10) ^c	49.49 (50)	49.43 (6)
Isotope ratio 79/81		1.026 (26)	1.021 (4)	1.021	1.023
Atomic weight		79.91 (1)	79.907 (2)	79.91 (1)	79.906 (1)
Error		NS	NS	SD	SD
Annotation					

Mass no.	Nuclidic mass	55CAM2	64CAT1	83ICA1
79	78.9183361	50.5367 (49)	50.686 (47)	50.69 (5)
81	80.916290	49.4633 (49)	49.314 (47)	49.31 (5)
Isotope ratio 79/81		1.0217 (2)	1.02784 (190)	1.028
Atomic weight		79.9066 (1)	79.9036 (9)	79.904 (1)
Error		2XSD	3XSD	
Annotation		C, B		

Element ^{36}Kr Krypton

1961 83.80

Mass no.	Nuclidic mass	47DIB1	47LOU1	47LOU1	50NIE2
78	77.920397	0.36 (1)	0.343 (3)	0.341 (3)	0.354 (2)
80	79.916375	2.25 (2)	2.233 (9)	2.223 (2)	2.27 (1)
82	81.913483	11.57 (4)	11.510 (40)	11.490 (10)	11.56 (2)
83	82.914134	11.44 (3)	11.490 (30)	11.470 (20)	11.55 (2)
84	83.9115064	57.14 (3)	57.000 (90)	57.040 (40)	56.90 (10)
86	85.910614	17.24 (5)	17.420 (30)	17.440 (30)	17.37 (2)
Isotope ratio 78/84	0.0063	0.0060	0.0060	0.0062	
80/84	0.039	0.039	0.039	0.040	
82/84	0.202	0.202	0.201	0.203	
83/84	0.200	0.202	0.201	0.203	
86/84	0.302	0.306	0.306	0.305	
Atomic weight	83.799 (2)	83.805 (1)	83.8069 (7)	83.8006 (7)	
Error	SD	SE	SE	3XP	
Annotation					

Element ^{36}Kr Krypton--continued

Mass no.	Nuclidic mass	50SCH1	64CLA1	71MEL1
78	77.920397	0.353 (1)	0.353 (5) ^c	0.355 (1)
80	79.916375	2.29 (1)	2.264 (19) ^c	2.256 (7)
82	81.913483	11.58 (1)	11.590 (52) ^c	11.553 (9)
83	82.914134	11.51 (2)	11.538 (50) ^c	11.536 (9)
84	83.9115064	56.95 (4)	56.929 (60) ^c	56.982 (10)
86	85.910614	17.31 (3)	17.325 (50) ^c	17.318 (9)
Isotope ratio	78/84	0.0062	0.0062	0.0062
	80/84	0.040	0.040	0.0396
	82/84	0.203	0.204	0.2027
	83/84	0.202	0.203	0.2024
	86/84	0.304	0.304	0.3039
Atomic weight		83.7988 (8)	83.800 (2)	83.8004 (4)
Error		NS	SD	NS
Annotation				
Mass no.	Nuclidic mass	73WAL1	83ICA1	
78	77.920397	0.360 (4)	0.35 (2)	
80	79.916375	2.277 (4)	2.25 (2)	
82	81.913483	11.58 (1)	11.6 (1)	
83	82.914134	11.52 (1)	11.5 (1)	
84	83.9115064	56.96 (1)	57.0 (3)	
86	85.910614	17.30 (1)	17.3 (2)	
Isotope ratio	78/84	0.0063	0.0061	
	80/84	0.0400	0.039	
	82/84	0.2033	0.203	
	83/84	0.2022	0.202	
	86/84	0.3037	0.304	
Atomic weight		83.7985 (4)	83.800 (5)	
Error		P		
Annotation		B		

Element ^{37}Rb Rubidium

1961 85.47 1969 85.4678 (3)

Mass no.	Nuclidic mass	38BRE1	48PAU1	50NIE2	50NIE2
85	84.9117996	72.299 (77) ^c	72.53 (23) ^c	72.18 (14) ^c	72.137 (93) ^c
87	86.9091836	27.701 (77) ^c	27.47 (23) ^c	27.82 (14) ^c	27.863 (93) ^c
Isotope ratio 85/87		2.61 (1)	2.64 (3)	2.595 (6)	2.589 (4)
Atomic weight		85.465 (2)	85.460 (5)	85.467 (3)	85.468 (2)
Error		NS	3XSD	P	P
Annotation					
Mass no.	Nuclidic mass	53HER1	56PAC1	600MU1	63SHI1
85	84.9117996	72.153 (23) ^c	72.603 (75)	72.1 (1) ^c	72.218 (12) ^c
87	86.9091836	27.847 (23) ^c	27.397 (75)	27.9 (1) ^c	27.782 (12) ^c
Isotope ratio 85/87		2.591 (3)	2.65 (1)	2.58 (1)	2.5995 (15)
Atomic weight		85.4680 (5)	85.459 (1)	85.469 (2)	85.4667 (2)
Error		NS	NS	NS	3XSD
Annotation					
Mass no.	Nuclidic mass	63SHI1	63SHI1	69CAT1	83ICAT1
85	84.9117996	72.243 (38) ^c	72.183 (38) ^c	72.1654 (132)	72.165 (13)
87	86.9091836	27.757 (38) ^c	27.817 (38) ^c	27.8346 (132)	27.835 (13)
Isotope ratio 85/87		2.6027 (49)	2.5949 (49)	2.59265 (170)	2.5926
Atomic weight		85.4662 (8)	85.4674 (8)	85.4678 (3)	85.4678 (3)
Error		SD	SD	3XSD	
Annotation					
C, B					

Element ^{38}Sr Strontium

1961 87.62

Mass no.	Nuclidic mass	38NIE1	48WHI1	53ALD1	54HER1
84	83.913428	0.561 (11) ^c	0.55 (8)	0.553 (41)	0.471 (58)
86	85.9092732	9.858 (90) ^c	9.75 (10)	9.866 (23)	9.892 (53)
87	86.9088902	7.018 (66) ^c	6.96 (7)	7.018 (31)	6.923 (54)
88	87.9056249	82.56 (10) ^c	82.74 (83)	82.563 (48)	82.713 (83)
Isotope ratio 84/86		0.057	0.056	0.056	0.048
87/86		0.712	0.714	0.711	0.700
88/86		8.375	8.486	8.368	8.362
Atomic weight		87.617 (2)	87.620 (4)	87.617 (2)	87.620 (2)
Error		P	P	SE	SD

Annotation

Mass no.	Nuclidic mass	54HER1	54HER1	54HER1	54HER1
84	83.913428	0.55	0.56	0.5608 (82)	0.55
86	85.9092732	9.76	9.81	9.946 (23)	9.91
87	86.9088902	6.94	6.94	7.019 (31)	7.01
88	87.9056249	82.75	82.69	82.472 (35)	82.52
Isotope ratio 84/86		0.056	0.057	0.0564	0.055
87/86		0.711	0.707	0.7057	0.707
88/86		8.478	8.429	8.292	8.327
Atomic weight		87.6197 (4)	87.6183 (4)	87.6147 (6)	87.6160 (4)
Error		SD	SD	SD	SD

Annotation

Mass no.	Nuclidic mass	54HER1	56AKI1	58PIN1	58PIN1
84	83.913428	0.57	0.58 (6)	0.561 (16)	0.561 (16)
86	85.9092732	9.87	9.99 (2)	9.872 (23)	9.850 (23)
87	86.9088902	7.01	7.14 (25)	7.029 (25)	7.023 (25)
88	87.9056249	82.56	82.29 (62)	82.538 (33)	82.565 (33)
Isotope ratio 84/86		0.058	0.058	0.0568	0.0570
87/86		0.710	0.715	0.712	0.713
88/86		8.365	8.237	8.361	8.382
Atomic weight		87.6159 (4)	87.612 (3)	87.6161 (7)	87.6166 (7)
Error		SD	NS	SD	SD

Annotation

Element ^{38}Sr Strontium--continued

Mass no.	Nuclidic mass	58PIN1	600MU1	82M001	83ICA1
84	83.913428	0.56	0.60 (1)	0.55738 (155)	0.56 (1)
86	85.9092732	9.87	9.75 (1)	9.85659 (337)	9.86 (1)
87	86.9088902	7.01	7.07 (1)	7.00152 (263)	7.00 (1)
88	87.9056249	82.56	82.58 (3)	82.58451 (657)	82.58 (1)
Isotope ratio 84/86	0.057	0.0061	0.056549 (143)	0.0568	
87/86	0.7012	0.7241	0.710339 (261)	0.7099	
88/86	8.365	8.470	8.378612 (3248)	8.375	
Atomic weight	87.6164 (4)	87.6166 (4)	87.616814 (117)	87.6167 (4)	
Error	SD	NS	3XSD		
Annotation			C, B		

Element ^{39}Y Yttrium

Mass no.	1961 88.905	1969 88.9059	56HES1	57COL1	83ICA1
89	88.9058560	100	100	100	
Isotope ratio					
Atomic weight		88.90586 (2)	88.90586 (2)	88.90586 (2)	
Error		NS	NS		
Annotation					

Element ^{40}Zr Zirconium

Mass no.	1961 91.22	1983 91.224 (2)	48WH11	49HIN1	58DRA1	63MUR1
90	89.9047080	51.46 (51)	51.7	51.12 (11)	51.50 (17) ^c	
91	90.9056442	11.23 (11)	10.8	11.22 (5)	11.223 (71) ^c	
92	91.9050392	17.11 (17)	17.1	17.40 (4)	17.10 (10) ^c	
94	93.9063191	17.40 (17)	17.5	17.57 (4)	17.38 (10) ^c	
96	95.908272	2.80 (3)	2.9	2.79 (10)	2.799 (19) ^c	
Isotope ratio 90/92	3.008	3.023	2.938	3.011		
91/92	0.656	0.632	0.645	0.656		
94/92	1.017	1.023	1.010	1.016		
96/92	0.164	0.170	0.160	0.164		
Atomic weight	91.224 (8)	91.229 (6)	91.326 (5)	91.222 (4)		
Error	SD	NS	SD	2XSD		
Annotation						

Element $_{40}^{Zr}$ Zirconium--continued

Mass no.	Nuclidic mass	78SHI2	81MIN1	83NOM1	83ICA1
90	89.9047080	51.449 (59)	51.444 $^{+0.085}_{-0.025}$	51.452 (9) ^c	51.45 (2)
91	90.9056442	11.320 (15)	11.214 $^{+0.008}_{-0.004}$	11.223 (12) ^c	11.22 (2)
92	91.9050392	17.189 (21)	17.150 $^{+0.008}_{-0.017}$	17.146 (7) ^c	17.15 (1)
94	93.9063191	17.283 (21)	17.393 $^{+0.016}_{-0.057}$	17.380 (12) ^c	17.38 (2)
96	95.908272	2.759 (4)	2.798 $^{+0.008}_{-0.019}$	2.799 (5) ^c	2.80 (1)
Isotope ratio 90/92	2.993		3.000	3.0008 (12)	2.9965
91/92	0.659		0.654	0.6546 (7)	0.6564
94/92	1.005		1.014	1.0137 (12)	1.0093
96/92	0.161		0.163	0.1632 (8)	0.1619
Atomic weight		91.219 (1)	91.224 (2)	91.2235 (4)	91.221 (6)
Error		SD	P	2XSD	
Annotation			C	C, B	

Element $_{41}^{Nb}$ Niobium

1961	92.906	1969	92.9064
Mass no.	Nuclidic mass	56WHI1	83ICA1
93	92.9063780	100	100
Isotope ratio			
Atomic weight		92.90638 (2)	92.90638 (2)
Error		NS	
Annotation			

Element $_{42}^{\text{Mo}}$ Molybdenum

	1961	95.94	1969	95.94 (3)	1975	95.94	
Mass no.	Nuclidic mass		39MAT1		40VAL1		46WIL1
							46WIL1
92	91.906809		15.51 (14)		14.9 (2)		15.83 (16)
94	93.9050862		8.69 (12)		9.40 (9)		9.102 (91)
95	94.9058379		16.27 (12)		16.1 (2)		15.76 (16)
96	95.9046755		16.83 (12)		16.6 (2)		16.54 (16)
97	96.9060179		8.73 (12)		9.65 (10)		9.481 (94)
98	97.9054050		25.423 (82)		24.1 (2)		23.70 (24)
100	99.907473		8.54 (12)		9.25 (10)		9.576 (91)
Isotope ratio 92/96		0.922		0.898		0.957	0.959
94/96		0.516		0.566		0.550	0.543
95/96		0.967		0.970		0.953	0.950
97/96		0.519		0.581		0.573	0.574
98/96		1.411		1.452		1.433	1.446
100/96		0.507		0.557		0.579	0.585
Atomic weight		95.887 (8)		95.91 (1)		95.885 (9)	95.894 (9)
Error		NS		P		P	P
Annotation							
Mass no.	Nuclidic Mass		49HIB2	600MU1	63MUR1	64CR01	64WET1
92	91.906809		15.05 (1)	15.75 (20)	14.779 (81)	14.739 (27)	14.826 (48) ^c
94	93.9050862		9.35 (1)	9.21 (9)	9.184 (56)	9.107 (43)	9.243 (32) ^c
95	94.9058379		15.78 (4)	15.91 (7)	15.931 (88)	15.918 (43)	15.908 (48) ^c
96	95.9046755		16.56 (4)	16.38 (11)	16.699 (88)	16.713 (61)	16.720 (48) ^c
97	96.9060179		9.50 (1)	9.44 (2)	9.530 (57)	9.476 (63)	9.561 (32) ^c
98	97.9054050		24.00 (3)	23.67 (6)	24.28 (12)	24.421 (99)	24.165 (79) ^c
100	99.907473		9.68 (1)	9.56 (5)	9.597 (57)	9.625 (71)	9.577 (32) ^c
Isotope ratio 92/96		0.909	0.961	0.885	0.882	0.887	
94/96		0.565	0.562	0.550	0.545	0.553	
95/96		0.953	0.971	0.954	0.952	0.951	
97/96		0.580	0.576	0.571	0.567	0.572	
98/96		1.449	1.445	1.454	1.461	1.445	
100/96		0.585	0.584	0.575	0.576	0.573	
Atomic weight		95.9118 (9)	95.883 (8)	95.936 (5)	95.943 (4)	95.931 (3)	
Error		SD	SD	2XSD	SD	SE	
Annotation							

Element $_{42}\text{Mo}$ Molybdenum--continued

Mass no.	Nuclidic mass	72STE1	74M001	75TAM1	81MIN1	83ICA1
92	91.906809	14.83	14.8362 (148)	14.827 (12) ^c	14.83 (3) ^c	14.84 (4)
94	93.9050862	9.25	9.2466 (92)	9.239 (17) ^c	9.27 (5) ^c	9.25 (2)
95	94.9058379	15.92	15.9201 (159)	15.922 (22) ^c	15.90 (3) ^c	15.92 (4)
96	95.9046755	16.67	16.6756 (167)	16.683 (22) ^c	16.67 (3) ^c	16.68 (4)
97	96.9060179	9.55	9.5551 (96)	9.557 (15) ^c	9.53 (2) ^c	9.55 (2)
98	97.9054050	24.14	24.1329 (241)	24.143 (43) ^c	24.13 (5) ^c	24.13 (6)
100	99.907473	9.63	9.6335 (96)	9.629 (8) ^c	9.63 (2) ^c	9.63 (2)
Isotope ratio 92/96	0.890	0.8897	0.8888 (8)	0.8900 (18)	0.890	
94/96	0.555	0.5543	0.5538 (10)	0.5561 (38)	0.555	
95/96	0.955	0.9545	0.9544 (13)	0.9538 (24)	0.954	
97/96	0.573	0.5730	0.5729 (9)	0.5740 (19)	0.573	
98/96	1.448	1.4472	1.4472 (26)	0.4486 (44)	1.447	
100/96	0.578	0.5775	0.5772 (5)	0.5779 (11)	0.577	
Atomic weight	95.9319 (6)	95.9318 (9)	95.932 (1)	95.932 (2)	95.931 (2)	
Error	2XSD	2XSD	SD	NS		
Annotation	B					

Element $_{44}\text{Ru}$ Ruthenium

	1961	101.07	1969	101.07 (3)	1983	101.07 (2)			
Mass no.	Nuclidic mass		44EWA1		53FRI1		54BAL1		56WHI1
96	95.907596		5.68 (17)		5.50		5.47 (3)		5.57 (8)
98	97.905287		2.22 (9)		1.91		1.84 (2)		1.86 (4)
99	98.9059371		12.81 (19)		12.70		12.77 (5)		12.7 (1)
100	99.9042175		12.70 (19)		12.69		12.56 (5)		12.6 (1)
101	100.9055808		16.98 (25)		17.01		17.10 (6)		17.1 (1)
102	101.9043475		31.34 (47)		31.52		31.70 (6)		31.6 (2)
104	103.905422		18.27 (27)		18.67		18.56 (5)		18.5 (1)
Isotope ratio 96/100	0.447		0.443		0.436		0.442		
98/100	0.175		0.150		0.146		0.148		
99/100	1.009		1.001		1.017		1.008		
101/100	1.337		1.340		1.361		1.357		
102/100	2.468		2.484		2.524		2.508		
104/100	1.439		1.471		1.478		1.468		
Atomic weight		101.03 (1)		101.0672 (7)		101.069 (3)		101.062 (6)	
Error		P		NS		NS		NS	
Annotation									

Element $_{44}\text{Ru}$ Ruthenium--continued

Mass no.	Nuclidic mass	70TAK1	76DEV1	83ICA1
96	95.907596	5.52	5.52 (1)	5.52 (5)
98	97.905287	1.87	1.86 (1)	1.88 (5)
99	98.9059371	12.67	12.74 (2)	12.7 (1)
100	99.9042175	12.60	12.60 (2)	12.6 (1)
101	100.9055808	17.09	17.05 (1)	17.0 (1)
102	101.9043475	31.55	31.57 (3)	31.6 (2)
104	103.905422	18.71	18.66 (3)	18.7 (2)
Isotope ratio 96/100	0.438	0.438	0.438	
98/100	0.148	0.148	0.149	
99/100	1.006	1.011	1.008	
101/100	1.356	1.353	1.349	
102/100	2.503	2.506	2.508	
104/100	1.485	1.481	1.484	
Atomic weight	101.0805 (7)	101.068 (1)	101.070 (7)	
Error	NS	SD		
Annotation		B		

Element $_{45}\text{Rh}$ Rhodium

1961	102.905	1967	102.9055
Mass no.	Nuclidic mass	63LEI1	83ICA1
103	102.905503	100	100
Isotope ratio			
Atomic weight		102.90550 (2)	102.90550 (2)
Error	NS		
Annotation			

Element ^{46}Pd Palladium

	1961	106.4	1979	106.42			
Mass no.	Nuclidic mass	53SIT1	78KEL1	78SH11	81MER1	83ICA1	
102	101.905609	0.96	1.013 (1) ^c	1.020 (8)	1.040 (2) ^c	1.020 (12)	
104	103.904026	10.97	11.089 (2) ^c	11.14 (5)	11.261 (8) ^c	11.14 (8)	
105	104.905075	22.23	22.317 (3) ^c	22.33 (5)	22.518 (16) ^c	22.33 (8)	
106	105.903475	27.33	27.288 (5) ^c	27.33 (2)	27.374 (14) ^c	27.33 (5)	
108	107.903894	26.71	26.534 (6) ^c	26.46 (6)	26.292 (14) ^c	26.46 (9)	
110	109.905169	11.81	11.759 (2) ^c	11.72 (6)	11.515 (8) ^c	11.72 (9)	
Isotope ratio 102/104		0.0875	0.09132 (5)	0.0916	0.092381 (82)	0.0916	
105/104		2.0264	2.01245 (29)	2.0045	1.99967 (30)	2.0045	
106/104		2.4913	2.46077 (46)	2.4533	2.43085 (61)	2.4533	
108/104		2.4348	2.39279 (57)	2.3752	2.58997 (53)	2.3752	
110/104		1.0766	1.06040	1.0521	1.02255 (29)	1.0521	
Atomic weight		106.4307 (7)	106.4198 (1)	106.415 (3)	106.3987 (5)	106.415 (4)	
Error		NS	2XSE	2XSD	2XSE		
Annotation				C, B			

Element ^{47}Ag Silver

	1961	107.870 (3)	1965	107.868	1981	107.8682 (3)	
Mass no.	Nuclidic mass	48PAU1	48WH11	57HES1	57HES1		
107	106.905095	51.92 (14) ^c	51.35 (7)	51.99 (46) ^c	51.90 (35) ^c		
109	108.904754	48.08 (14) ^c	48.65 (7)	48.01 (46) ^c	48.10 (35) ^c		
Isotope ratio 107/109		1.080 (6)	1.055	1.083 (20)	1.079 (15)		
Atomic weight		107.867 (3)	107.878 (1)	107.865 (9)	107.867 (7)		
Error		3XSD	SE	SE	SE		
Annotation							
Mass no.	Nuclidic mass	57HES1	59CR01	59CR01	60MUR1		
107	106.905095	52.02 (35) ^c	52.015 (46) ^c	52.036 (16) ^c	51.830 (46) ^c		
109	108.904754	47.98 (35) ^c	47.985 (46) ^c	47.964 (16) ^c	48.170 (46) ^c		
Isotope ratio 107/109		1.084 (15)	1.084 (2)	1.0849 (7)	1.076 (2)		
Atomic weight		107.865 (7)	107.8646 (9)	107.8642 (3)	107.8683 (9)		
Error		NS	SD	SD	P		
Annotation							

Element ^{47}Ag Silver--continued

Mass no.	Nuclidic mass	60MUR1	60SHI1	62CR01	62MUR1
107	106.905095	51.737 (46) ^c	51.818 (30) ^c	51.77 (10) ^c	51.597 (46) ^c
109	108.904754	48.263 (46) ^c	48.182 (30) ^c	48.23 (10) ^c	48.403 (46) ^c
Isotope ratio 107/109		1.072 (2)	1.07547 (130)	1.0733 (43)	1.066 (2)
Atomic weight		107.8702 (9)	107.8686 (6)	107.870 (2)	107.8730 (9)
Error		P	3XSD	3XSD	SD
Annotation			c	c	

Mass no.	Nuclidic mass	62MUR1	62MUR1	62SHI1	81MUR1
107	106.905095	51.503 (49) ^c	51.527 (49) ^c	51.830 (26) ^c	52.07 (2)
109	108.904754	48.496 (49) ^c	48.473 (49) ^c	48.170 (26) ^c	47.93 (2)
Isotope ratio 107/109		1.062 (2)	1.063 (2)	1.07597 (135)	1.0864
Atomic weight		107.87 (1)	107.874 (1)	107.8683 (5)	107.8635 (4)
Error		SD	SD	3XSD	NS
Annotation					

Mass no.	Nuclidic mass	82POW1	83CHE1	83ICA1
107	106.905095	51.839170 (5057)	51.948 (9) ^c	51.839 (5)
109	108.904754	48.160830 (5057)	48.052 (9) ^c	48.161 (5)
Isotope ratio 107/109		1.07638 (135)	1.0811 (17)	1.07637
Atomic weight		107.8681 (1)	107.8660 (2)	107.8682 (1)
Error		3XSD	2XSD	
Annotation		c, B		

Element ^{48}Cd Cadmium

1961 112.40 1975 112.41

Mass no.	Nuclidic mass	48EWA1	48LEL1	48WH11	49HIB2	
106	105.906461	1.2	1.215 (12) ^c	1.22 (1)	1.22 (1)	
108	107.904186	0.7	0.875 (10) ^c	0.98 (1)	0.89 (2)	
110	109.903007	12.8	12.39 (12) ^c	12.35 (4)	12.43 (4)	
111	110.904182	13.0	12.75 (12) ^c	12.78 (4)	12.86 (4)	
112	111.9027614	24.2	24.07 (19) ^c	24.00 (10)	23.79 (6)	
113	112.9044013	12.2	12.26 (12) ^c	12.30 (4)	12.34 (5)	
114	113.9033607	28.9	28.86 (10) ^c	28.75 (20)	28.81 (7)	
116	115.904758	7.0	7.58 (3) ^c	7.63 (4)	7.66 (3)	
Isotope ratio	106/112	0.050	0.0505 (5)	0.0508	0.0513	
	108/112	0.029	0.0364 (4)	0.0408	0.0374	
	110/112	0.529	0.5147 (51)	0.5146	0.5225	
	111/112	0.537	0.5297 (53)	0.5325	0.5406	
	113/112	0.504	0.5093 (51)	0.5125	0.5187	
	114/112	1.194	1.1990 (120)	1.1979	1.2110	
	116/112	0.289	0.3182 (32)	0.3179	0.3220	
Atomic weight		112.398 (9)	112.423 (4)	112.420 (4)	112.424 (2)	
Error		NS	P	P	SD	
Annotation						
Mass no.	Nuclidic mass	58PAL1	75ROS1	70MUR1	80ROS1	83ICA1
106	105.906461	1.240 (15)	1.25 (1)	1.27 (2)	1.25 (2)	1.25 (3)
108	107.904186	0.871 (9)	0.894 (6)	0.90 (2)	0.89 (1)	0.89 (1)
110	109.903007	12.32 (7)	12.51 (5)	12.48 (15)	12.49 (6)	12.49 (9)
111	110.904182	12.67 (7)	12.81 (4)	12.85 (15)	12.80 (4)	12.80 (6)
112	111.9027614	24.15 (12)	24.13 (7)	24.07 (8)	24.13 (7)	24.13 (11)
113	112.9044013	12.21 (3)	12.22 (4)	12.26 (7)	12.22 (4)	12.22 (6)
114	113.9033607	28.93 (20)	28.71 (10)	28.70 (8)	28.73 (21)	28.73 (21)
116	115.904758	7.61 (8)	7.47 (5)	7.47 (8)	7.49 (6)	7.49 (9)
Isotope ratio	106/112	0.0513	0.0518	0.0528	0.0518	0.0518
	108/112	0.0361	0.0370	0.0372	0.0369	0.0369
	110/112	0.5101	0.5184	0.5186	0.5176	0.5176
	111/112	0.5246	0.5309	0.5342	0.5305	0.5305
	113/112	0.5056	0.5064	0.5093	0.5305	0.5064
	114/112	1.1979	1.1898	1.1926	0.5064	1.1906
	116/112	0.3151	0.3096	0.3106	1.1906	0.3104
Atomic Weight		112.426 (5)	112.410 (4)	112.408 (6)	112.412 (4)	112.412 (5)
Error		SD	2XSD	NS		
Annotation		C		B		

Element $_{49}\text{In}$ Indium

1961 114.82

Mass no.	Nuclidic mass	48WH11	49HIB2	56WH11	83ICA1
113	112.904056	4.23 (3)	4.16 (1)	4.33 (4)	4.3 (2)
115	114.903875	95.77 (3)	95.84 (1)	95.67 (4)	95.7 (2)
Isotope ratio 113/115	0.04417		0.04341	0.04526	0.04493
Atomic weight	114.8193 (6)		114.8207 (2)	114.8173 (8)	114.818 (4)
Error		SE	SD	NS	
Annotation				B	

Element $_{50}\text{Sn}$ Tin

1961 118.69 1969 118.69 (3) 1983 118.710 (7)

Mass no.	Nuclidic mass	48HIN1	48WH11	49HIB2
112	111.904823	0.94 (2)	0.900 (3)	1.01 (3)
114	113.902781	0.65 (3)	0.61 (1)	0.68 (1)
115	114.9033441	0.33 (2)	0.350 (6)	0.35 (3)
116	115.9017435	14.36 (4)	14.07 (8)	14.28 (1)
117	116.9029536	7.51 (4)	7.54 (3)	7.67 (5)
118	117.9016066	24.21 (7)	23.98 (3)	23.84 (8)
119	118.9033102	8.45 (4)	8.620 (3)	8.68 (1)
120	119.9021990	33.11 (10)	33.03 (12)	32.75 (3)
122	121.903440	4.61 (5)	4.78 (1)	4.74 (4)
124	123.905271	5.83 (5)	6.110 (6)	6.01 (9)
Isotope ratio 112/120	0.0284	0.0272	0.0308	
114/120	0.0196	0.0185	0.0208	
115/120	0.0100	0.0106	0.0107	
116/120	0.4337	0.4260	0.4360	
117/120	0.2268	0.2283	0.2342	
118/120	0.7312	0.7260	0.7278	
119/120	0.2552	0.2610	0.2650	
122/120	0.1392	0.1447	0.1447	
124/120	0.1761	0.1850	0.1835	
Atomic weight	118.729 (4)	118.762 (3)	118.746 (6)	
Error	P	SE	SD	
Annotation				

Element ^{50}Sn Tin--continued

Mass no.	Nuclidic Mass	52DIB1	65LAE1	83DEV1	83ICA1
112	111.904823	0.99 (5)	1.01 (3)	0.973 (3)	0.97 (1)
114	113.902781	0.68 (3)	0.67 (3)	0.652 (3)	0.65 (1)
115	114.9033441	0.36 (3)	0.38 (3)	0.359 (3)	0.36 (1)
116	115.9017435	14.49 (10)	14.76 (5)	14.532 (36)	14.53 (11)
117	116.9029536	7.71 (8)	7.75 (3)	7.675 (23)	7.68 (7)
118	117.9016066	24.09 (20)	24.30 (8)	24.218 (36)	24.22 (11)
119	118.9033102	8.59 (8)	8.55 (3)	8.583 (13)	8.58 (4)
120	119.9021990	32.52 (30)	32.38 (8)	32.590 (33)	32.59 (10)
122	121.903440	4.77 (5)	4.56 (3)	4.629 (9)	4.63 (3)
124	123.905271	5.81 (6)	5.64 (3)	5.789 (18)	5.79 (5)
Isotope ratio 112/120	0.0304	0.0312	0.02984	0.0298	
114/120	0.0209	0.0207	0.0200	0.0199	
115/120	0.0111	0.0116	0.0110	0.0111	
116/120	0.4456	0.4558	0.4459	0.4458	
117/120	0.2371	0.2392	0.2355	0.2357	
118/120	0.7408	0.7504	0.7432	0.7432	
119/120	0.2641	0.2642	0.2634	0.2633	
122/120	0.1467	0.1408	0.1420	0.1421	
124/120	0.1787	0.1742	0.1776	0.1777	
Atomic weight	118.726 (7)	118.685 (4)	118.710 (7)	118.710 (5)	
Error	P	SD	3XSD		
Annotation		B			

Element ^{51}Sb Antimony

1961 121.75 1969 121.75 (3)

Mass no.	Nuclidic mass	48WH11	83ICA1
121	120.9038237	57.25 (57)	57.3 (9)
123	122.904222	42.75 (57)	42.7 (9)
Isotope ratio 121/123	1.339		
Atomic weight	121.76 (1)	121.76 (2)	
Error	P		
Annotation	B		

Element ^{52}Te Tellurium

1961 127.60 1969 127.60 (3)

Mass no.	Nuclidic mass	46WIL1	48WHI1	48WHI1	49HIB2
120	119.904021	0.090 (6) ^c	0.091 (1)	0.092	0.090 (10)
122	121.903055	2.43 (6) ^c	2.49 (2)	2.32	2.47 (1)
123	122.904278	0.85 (2) ^c	0.89 (2)	0.88	0.89 (1)
124	123.902825	4.59 (13) ^c	4.63 (5)	4.51	4.74 (2)
125	124.904435	6.98 (7) ^c	7.01 (1)	6.99	7.03 (3)
126	125.903310	18.70 (7) ^c	18.72 (4)	18.53	18.72 (5)
128	127.904464	31.85 (9) ^c	31.72 (1)	32.57	31.75 (6)
130	129.906229	34.51 (7) ^c	34.46 (9)	34.11	34.27 (5)
Isotope ratio	120/130	0.00256 (17)	0.0026	0.0027	0.0026
	122/130	0.0705 (18)	0.0722	0.0680	0.0721
	123/130	0.0247 (5)	0.0230	0.0258	0.0260
	124/130	0.133 (4)	0.134	0.132	0.138
	125/130	0.202 (2)	0.203	0.205	0.205
	126/130	0.542 (2)	0.543	0.543	0.546
	128/130	0.923 (3)	0.920	0.955	0.926
Atomic weight		127.632 (7)	127.623 (3)	127.6356 (9)	127.615 (2)
Error		NS	SE	NS	SD
Annotation					

Mass no.	Nuclidic mass	78SMI1	83ICA1
120	119.904021	0.0960 (7) ^c	0.096 (2)
122	121.903055	2.603 (3) ^c	2.60 (1)
123	122.904278	0.908 (1) ^c	0.908 (3)
124	123.902825	4.816 (3) ^c	4.816 (8)
125	124.904435	7.139 (3) ^c	7.14 (1)
126	125.903310	18.952 (5) ^c	18.95 (1)
128	127.904464	31.687 (7) ^c	31.69 (2)
130	129.906229	33.799 (7) ^c	33.80 (2)
Isotope ratio	120/130	0.00284 (2)	0.0028
	122/130	0.07701 (4)	0.0769
	123/130	0.02687 (2)	0.0269
	124/130	0.14250 (6)	0.142
	125/130	0.21122 (6)	0.211
	126/130	0.56078	0.561
	128/130	0.93753 (16)	0.937
Atomic weight		127.5856 (3)	127.5858 (9)
Error		2XSD	
Annotation		B	

Element ^{53}I Iodine

1961 126.9044 1969 126.9045

Mass no.	Nuclidic mass	49LEL1	83ICA1
127	126.904477	100	100
Isotope ratio			
Atomic weight 126.90448 (3) 126.90448 (3)			
Error NS			
Annotation			

Element ^{54}Xe Xenon

1961 131.30 1979 131.29 (3)

Mass no.	Nuclidic mass	47DIB1	47LOU1	50NIE2	71POD1
124	123.90612	0.102 (9)	0.095 (1)	0.096 (1)	0.0949 (5) ^c
126	125.904281	0.098 (3)	0.088 (1)	0.090 (1)	0.0884 (5) ^c
128	127.9035308	1.93 (1)	1.917 (6)	1.919 (4)	1.914 (5) ^c
129	128.9047801	26.51 (2)	26.240 (80)	26.44 (8)	26.431 (21) ^c
130	129.9035095	3.68 (4)	4.053 (5)	4.08 (1)	4.064 (8) ^c
131	130.905076	21.04 (9)	21.240 (30)	21.18 (5)	21.229 (21) ^c
132	131.904148	27.12 (7)	26.930 (20)	26.89 (7)	26.879 (21) ^c
134	133.905395	10.54 (5)	10.520 (20)	10.44 (2)	10.432 (21) ^c
136	135.907219	8.98 (3)	8.930 (30)	8.87 (1)	8.868 (16) ^c
Isotope ratio		124/130	0.0277	0.0234	0.02335 (14)
		126/130	0.0266	0.0217	0.0220
		128/130	0.5245	0.4730	0.4708 (17)
		129/130	7.204	6.474	6.488
		131/130	5.717	5.241	5.198
		132/130	7.370	6.644	6.599
		134/130	2.864	2.596	2.562
		136/130	2.440	2.203	2.176
Atomic weight		131.304 (2)	131.302 (2)	131.292 (2)	131.293 (1)
Error		SD	SE	P	2XSD
Annotation				B	

Element ^{54}Xe Xenon--continued

Mass no.	Nuclidic mass	83ICA1
124	123.90612	0.10 (1)
126	125.904281	0.09 (1)
128	127.9035308	1.91 (3)
129	128.9047801	26.4 (6)
130	129.9035095	4.1 (1)
131	130.905076	21.2 (4)
132	131.904148	26.9 (5)
134	133.905395	10.4 (2)
136	135.907219	8.9 (1)
Isotope ratio	124/130	0.0244
	126/130	0.0220
	128/130	0.4659
	129/130	6.439
	131/130	5.171
	132/130	6.561
	134/130	2.537
	136/130	2.171
Atomic weight		131.29 (2)
Error		
Annotation		

Element ^{55}Cs Caesium

	1961	132.905	1969	132.9055	1971	132.9054
Mass no.		Nuclidic mass		56WH11		83ICA1
133		132.905433		100		100
Isotope ratio						
Atomic weight				132.90543 (5)		132.90543 (5)
Error					NS	
Annotation						

Element ^{56}Ba Barium

	1961	137.34	1969	137.34 (3)	1975	134.33	
Mass no.	Nuclidic mass	38NIE1		49HER1	56AKI1	600MU1	62RID1
130	129.906277	0.1010 (43) ^c		0.103 (4)	0.13 (2)	0.109 (50)	0.098 (2)
132	131.905042	0.0975 (43) ^c		0.096 (4)	0.19 (2)	0.103 (30)	0.091 (3)
134	133.904490	2.415 (49) ^c		2.39 (5)	2.60 (5)	2.45 (5)	2.33 (2)
135	134.905668	6.59 (13) ^c		6.56 (12)	6.73 (12)	6.72 (19)	6.42 (3)
136	135.904556	7.81 (13) ^c		7.79 (10)	8.07 (10)	8.06 (13)	7.77 (2)
137	136.905816	11.32 (19) ^c		11.25 (10)	11.87 (25)	11.41 (10)	11.18 (5)
138	137.905236	71.66 (21) ^c		71.83 (29)	70.41 (35)	71.12 (13)	72.11 (6)
Isotope ratio 130/138		0.00141 (6)		0.001434	0.00185	0.00153	0.001359
132/138		0.00136 (6)		0.001336	0.0270	0.00145	0.001262
134/138		0.0337 (6)		0.03327	0.03693	0.03445	0.03231
135/138		0.092 (2)		0.0913	0.0956	0.09449	0.08903
136/138		0.109 (2)		0.1085	0.1146	0.11333	0.10775
137/138		0.158 (3)		0.1566	0.1686	0.16013	0.15504
Atomic weight		137.328 (4)		137.317 (4)	137.297 (5)	137.315 (7)	137.341 (1)
Error		P		2XSD	P	NS	SD
Annotation							

Mass no.	Nuclidic mass	62UME1	69EUG1	73LAE1	76CH01
130	129.906277	0.1022 (1) ^c	0.1058 (2)	0.1054 (4) ^c	0.10
132	131.905042	0.0986 (1) ^c	0.1012 (2)	0.1013 (4) ^c	0.10
134	133.904490	2.388 (2) ^c	2.417 (3)	2.411 (2) ^c	2.42
135	134.905668	6.531 (6) ^c	6.592 (2)	6.593 (6) ^c	6.59
136	135.904556	7.817 (6) ^c	7.853 (4)	7.863 (6) ^c	7.81
137	136.905816	11.134 (7) ^c	11.232 (4)	11.248 (9) ^c	11.32
138	137.905236	71.929 (23) ^c	71.699 (7)	71.678 (23) ^c	71.66
Isotope ratio 130/138		0.0014215 (20)	0.001473	0.001471 (6)	0.00140
132/138		0.0013712 (17)	0.001412	0.001413 (6)	0.00140
134/138		0.033197 (34)	0.03371	0.03363 (3)	0.0338
135/138		0.09080 (8)	0.09194	0.09198 (9)	0.0920
136/138		0.10868 (9)	0.10952	0.10970 (9)	0.1090
137/138		0.15479 (10)	0.15665	0.15693 (10)	0.1580
Atomic weight		137.3321 (2)	137.3269 (1)	137.3268 (2)	137.327 (1)
Error		2XSD	SE	3XSD	NS
Annotation			C, B		

Element ^{56}Ba Barium--continued

Mass no.	Nuclidic mass	83ICA1
130	129.906277	0.106 (2)
132	131.905042	0.101 (2)
134	133.904490	2.417 (27)
135	134.905668	6.592 (18)
136	135.904556	7.854 (39)
137	136.905816	11.23 (1)
138	137.905236	71.70 (7)
Isotope ratio 130/138		0.001478
132/138		0.001409
134/138		0.03371
135/138		0.09194
136/138		0.10954
137/138		0.15663
Atomic weight		137.327 (1)
Error		
Annotation		

Element ^{57}La Lanthanum

1961	138.91	1969	138.9055 (3)						
Mass no.	Nuclidic mass			47ING2	56WH1	72MAS1	75YAN1	83ICA1	
138	137.907114			0.089 (2)	0.0885 (16)	0.089 (2) ^c	0.089 (2) ^c	0.09 (2)	
139	138.906355			99.911 (2)	99.9115 (16)	99.911 (2) ^c	99.911 (2) ^c	99.91 (2)	
Isotope ratio 138/139				0.000891	0.000893 (16)	.0008873 (24)	.000890 (7)	0.000891	
Atomic weight				138.90547 (2)	138.90547 (2)	138.90547 (2)	138.90547 (2)	138.9055 (2)	
Error				SE	NS	2XSD	2XSD		
Annotation				B					

Element ^{58}Ce Cerium

1961 140.12

Mass no.	Nuclidic mass	47ING2	49HIB2	62UME1	83ICA1
136	135.90714	0.193 (5)	0.195 (5)	0.1904 (3) ^c	0.19 (1)
138	137.905996	0.250 (5)	0.265 (5)	0.2536 (4) ^c	0.25 (1)
140	139.905442	88.48 (10)	88.449 (20)	88.475 (8) ^c	88.48 (10)
142	141.909249	11.07 (10)	11.098 (33)	11.081 (7) ^c	11.08 (10)
Isotope ratio	136/140	0.002181	0.002205	0.0021526 (29)	0.002147
	138/140	0.002825	0.002996	0.0028663 (41)	0.002825
	142/140	0.1251	0.12547	0.12525 (8)	0.1252
Atomic weight		140.115 (2)	140.1147 (6)	140.1148 (1)	140.115 (2)
Error		SE	SD	2XSD	
Annotation				B	

Element ^{59}Pr Praseodymium

1961 140.907 1969 140.9077

Mass no.	Nuclidic mass	57COL1	83ICA1
141	140.907657	100	100
Isotope ratio			
Atomic weight		140.90766 (3)	140.90766 (3)
Error		NS	
Annotation			

Element ^{60}Nd Neodymium						
	1961	144.24	1969	144.24 (3)		
Mass no.	Nuclidic mass	47ING3	48ING2	48MAT1	53WAL1	
142	141.907731	27.25	27.13 (20)	26.80 (16)	27.09 (3)	
143	142.909823	12.26	12.20 (10)	12.12 (8)	12.14 (2)	
144	143.910096	23.97	23.87 (20)	23.91 (12)	23.83 (3)	
145	144.912582	8.23	8.30 (5)	8.35 (8)	8.29 (1)	
146	145.913126	17.06	17.18 (20)	17.35 (9)	17.26 (2)	
148	147.916901	5.66	5.72 (6)	5.78 (8)	5.74 (2)	
150	149.920900	5.53	5.60 (6)	5.69 (8)	5.63 (2)	
Isotope ratio 143/142	0.4499	0.4497	0.4522	0.4513		
144/142	0.8797	0.8798	0.8921	0.8718		
145/142	0.3020	0.3060	0.3116	0.3037		
146/142	0.6261	0.6333	0.6474	0.6264		
148/142	0.2077	0.2108	0.2157	0.2077		
150/142	0.2030	0.2064	0.2123	0.2037		
Atomic weight	144.2254 (8)	144.238 (7)	144.257 (7)	144.244 (2)		
Error	NS	NS	SE	SD		
Annotation						
Mass no.	Nuclidic mass	56WHIT	64KOM1	76NAK1	81HOL1	83ICA1
142	141.907731	27.3 (2)	26.81 (60)	27.157	27.16 (4)	27.13 (10)
143	142.909823	12.32 (9)	12.07 (45)	12.177	12.18 (2)	12.18 (5)
144	143.910096	23.8 (2)	23.75 (60)	23.795	23.83 (4)	23.80 (10)
145	144.912582	8.29 (6)	8.39 (23)	8.293	8.30 (2)	8.30 (5)
146	145.913126	17.10 (14)	17.31 (53)	17.188	17.17 (3)	17.19 (8)
148	147.916901	5.67 (5)	5.99 (28)	5.755	5.74 (1)	5.76 (3)
150	149.920900	5.56 (5)	5.75 (35)	5.635	5.62 (1)	5.64 (3)
Isotope ratio 143/142	0.4513	0.4502	0.4484	0.4485	0.4489	
144/142	0.8719	0.8859	0.8762	0.8774	0.8773	
145/142	0.3037	0.3129	0.3054	0.3056	0.3059	
146/142	0.6260	0.6457	0.6329	0.6322	0.6336	
148/142	0.2077	0.2234	0.2113	0.2113	0.2123	
150/142	0.2037	0.2145	0.20/5	0.2069	0.2079	
Atomic weight	144.227 (6)	144.27 (3)	144.24123 (8)	144.239 (11)	144.242 (3)	
Error	NS	NS	NS	2XSD		
Annotation			B			

Element ^{62}Sm Samarium					
	1961	150.35	1969	150.4	1979
Mass no.	Nuclidic mass	48ING1	48MAT1	50HIB1	57AIT1
144	143.912009	3.16 (10)	2.95 (7)	2.87 (15)	3.02 (2)
147	146.914907	15.07 (15)	14.62 (12)	14.94 (6)	14.87 (4)
148	147.914832	11.27 (11)	10.97 (11)	11.24 (5)	11.22 (3)
149	148.917193	13.84 (14)	13.56 (12)	13.85 (6)	13.82 (4)
150	149.917285	7.47 (7)	7.27 (10)	7.36 (7)	7.40 (2)
152	151.919741	26.63 (26)	27.34 (25)	26.90 (10)	26.80 (5)
154	153.922218	22.53 (22)	23.29 (19)	22.84 (13)	22.88 (6)
Isotope ratio 144/152	0.1187	0.1079	0.1067	0.1126	
147/152	0.5659	0.5347	0.5554	0.5549	
148/152	0.4232	0.4012	0.4178	0.4187	
149/152	0.5197	0.4960	0.5149	0.5157	
150/152	0.2805	0.2659	0.2736	0.2761	
154/152	0.8460	0.8519	0.8491	0.8537	
Atomic weight	150.34 (1)	150.43 (1)	150.39 (1)	150.379 (3)	
Error	3XSE	NS	2XSD	NS	
Annotation					
Mass no.	Nuclidic mass	57COL1	64KOM1	75LUG1	81HOL1
144	143.912009	3.15 (3)	3.15 (4)	3.076 (1) ^c	3.09 (1)
147	146.914907	15.09 (10)	14.93 (4)	14.995 (1) ^c	15.03 (3)
148	147.914832	11.35 (9)	11.21 (8)	11.242 (1) ^c	11.24 (2)
149	148.917193	13.96 (10)	13.70 (16)	13.819 (1) ^c	13.84 (3)
150	149.917285	7.47 (6)	7.44 (8)	7.380 (1) ^c	7.38 (2)
152	151.919741	26.55 (20)	26.65 (13)	26.738 (2) ^c	26.72 (5)
154	153.922218	22.43 (20)	22.62 (2)	22.750 (1) ^c	22.70 (4)
Isotope ratio 144/152	0.1186	0.1182	0.11502 (4)	0.1156	0.1161
147/152	0.5684	0.5602	0.56081	0.5625	0.5618
148/152	0.4275	0.4206	0.42045 (3)	0.4207	0.4232
149/152	0.5258	0.5141	0.51683 (4)	0.5180	0.5169
150/152	0.2814	0.2792	0.27602 (2)	0.2762	0.2772
154/152	0.8448	0.8488	0.85082 (5)	0.8496	0.8502
Atomic weight	150.338 (9)	150.359 (4)	150.36558 (9)	150.361 (2)	150.36 (1)
Error	SD	NS	2XSD	2XSD	
Annotation					

Element $_{63}\text{Eu}$ Europium

1961 151.96

Mass no.	Nuclidic mass	47ING3	48HES1	57COL1	64KOM1	72LOV1
151	150.919860	47.75	47.77 (25) ^c	47.86 (8)	47.86 (28)	47.794 (25) ^c
153	152.921243	52.25	52.23 (25) ^c	52.14 (8)	52.14 (28)	52.206 (25) ^c
Isotope ratio 151/153	0.9139		0.9145 (91)	0.9179	0.9179	0.9155 (9)
Atomic weight	151.9656 (2)	151.965 (5)	151.963 (2)	151.963 (6)	151.9647 (5)	
Error		NS	P	SD	NS	2XSD
Annotation		D				

Mass no.	Nuclidic mass	72LOV1	81HOL1	83ICA1
151	150.919860	47.808 (50)	47.81 (3)	47.8 (5)
153	152.921243	52.192 (50)	52.19 (3)	52.2 (5)
Isotope ratio 151/153	0.9160 (18)		0.9160	0.9157
Atomic weight	151.964 (1)	151.9644 (6)	151.96 (1)	
Error		2XSD	2XSD	
Annotation				

Element $_{64}\text{Gd}$ Gadolinium

1961 157.25 1969 157.25 (3)

Mass no.	Nuclidic mass	47ING3	48HES1	50LEL1	57COL1
152	151.919803	0.20	0.200 (2)	0.200 (5)	0.205 (10)
154	153.920876	2.1	2.15 (2)	2.16 (2)	2.23 (3)
155	154.922629	14.90	14.78 (15)	14.68 (15)	15.10 (15)
156	155.922130	20.67	20.59 (21)	20.36 (20)	20.6 (2)
157	156.923967	15.73	15.71 (16)	15.64 (16)	15.70 (16)
158	157.924111	24.82	24.78 (25)	24.96 (25)	24.50 (25)
160	159.927061	21.77	21.79 (22)	22.01 (22)	21.6 (2)
Isotope ratio 152/158	0.00806		0.00807	0.00801	0.00837
154/158	0.0846		0.0868	0.0865	0.0910
155/158	0.6003		0.5964	0.5881	0.6164
156/158	0.8328		0.8309	0.8157	0.8409
157/158	0.6338		0.6340	0.6256	0.6409
160/158	0.8771		0.8793	0.8818	0.8817
Atomic weight	157.247 (4)	157.250 (8)	157.262 (8)	157.232 (7)	
Error		NS	P	P	SD
Annotation					

Element $_{64}^{154}$ Gd Gadolinium--continued

Mass no.	Nuclidic mass	66KOM1	70EUG1	81HOL1	83ICA1
152	151.919803	0.19 (1)	0.2029 (5)	0.200 (5)	0.20 (1)
154	153.920876	2.14 (1)	2.1809 (6)	2.18 (3)	2.18 (3)
155	154.922629	14.66 (28)	14.800 (3)	14.80 (6)	14.80 (5)
156	155.922130	20.57 (48)	20.466 (2)	20.47 (8)	20.47 (4)
157	156.923967	15.71 (20)	15.652 (2)	15.65 (6)	15.65 (3)
158	157.924111	25.09 (47)	24.835 (4)	24.84 (9)	24.84 (12)
160	159.927061	21.67 (42)	21.863 (2)	21.86 (8)	21.86 (4)
Isotope ratio 152/158	0.0075	0.008170	0.00805	0.00805	
154/158	0.0853	0.08782	0.0878	0.0878	
155/158	0.5843	0.59593	0.5958	0.5958	
156/158	0.8198	0.82408	0.8241	0.8241	
157/158	0.6261	0.63024	0.6300	0.6300	
160/158	0.8637	0.88033	0.8800	0.8800	
Atomic weight	157.25 (1)	157.2520 (1)	157.252 (3)	157.252 (2)	
Error	NS	SD	2XSD		
Annotation		B			

Element $_{65}^{158}$ Tb Terbium

Mass no.	Nuclidic mass	1961	1969	1984
		158.924	158.9254	
159	158.925350	100	100	
Isotope ratio				
Atomic weight		158.92535 (3)	158.92535 (4)	
Error		NS		
Annotation				

Element ^{66}Dy Dysprosium

1961 162.50 1969 162.50 (3)

Mass no.	Nuclidic mass	49ING1	50LEL1	57COL1	66KOM1
156	155.924287	0.0524 (5)	0.064 ^{+0.001} _{-0.064}	0.057 (1)	0.058 (9)
158	157.924412	0.0902 (9)	0.105 ^{+0.001} _{-0.105}	0.100 (1)	0.092 (4)
160	159.925203	2.294 (11)	2.36 (10)	2.35 (2)	2.53 (8)
161	160.926939	18.88 (9)	18.73 (19)	19.0 (1)	19.04 (25)
162	161.926805	25.53 (13)	25.36 (25)	25.5 (2)	25.52 (15)
163	162.928737	24.97 (12)	24.91 (25)	24.9 (2)	24.87 (30)
164	163.929183	28.18 (14)	28.47 ^{+0.25} _{-0.56}	28.1 (2)	28.18 (30)
Isotope ratio	156/164	0.00186	0.00225	0.00203	0.0021
	158/164	0.00320	0.00367	0.00356	0.0033
	160/164	0.0814	0.0829	0.0836	0.0898
	161/164	0.6700	0.6579	0.6762	0.6757
	162/164	0.9060	0.8908	0.9075	0.9056
	163/164	0.8861	0.8750	0.8861	0.8825
Atomic weight		162.499 (3)	162.50 (1)	162.495 (4)	162.492 (7)
Error		P	P	SD	NS

Annotation

Mass no.	Nuclidic mass	81HOL1	83ICAI
156	155.924287	0.056 (1)	0.06 (1)
158	157.924412	0.096 (2)	0.10 (1)
160	159.925203	2.34 (2)	2.34 (5)
161	160.926939	18.91 (5)	18.9 (1)
162	161.926805	25.51 (7)	25.5 (2)
163	162.928737	24.90 (7)	24.9 (2)
164	163.929183	28.19 (8)	28.2 (2)
Isotope ratio	156/164	0.00199	0.00213
	158/164	0.00341	0.00355
	160/164	0.0830	0.0830
	161/164	0.6708	0.6702
	162/164	0.9049	0.9043
	163/164	0.8833	0.8830
Atomic weight		162.498 (2)	162.498 (4)
Error		2XSD	
Annotation		B	

Element ^{67}Ho Holmium

	1961	164.930	1969	164.9303	1971	164.9304	
Mass no.	Nuclidic mass	50LEL1		57COL1		83ICA1	
165	164.930332	100		100		100	
Isotope ratio							
Atomic weight		164.93033 (4)		164.93033 (4)		164.93033 (4)	
Error		NS		NS			
Annotation							

Element ^{68}Er Erbium

	1961	167.26	1969	167.26 (3)			
Mass no.	Nuclidic mass	41WAH2		50HAY1	50LEL1	66KOM1	
162	161.928787	0.1		0.136 (3)	0.154 (7)	0.258 (4)	
164	163.929211	1.5		1.56 (3)	1.60 (2)	1.91 (1)	
166	165.930305	32.9		33.41 (30)	33.36 (33)	33.23 (16)	
167	166.932061	24.4		22.94 (20)	22.82 (23)	22.80 (25)	
168	167.932383	26.9		27.07 (30)	27.02 (27)	26.82 (40)	
170	169.935476	14.2		14.88 (20)	15.04 (15)	14.96 (20)	
Isotope ratio		162/166	0.00304	0.00407	0.00462	0.0078	
		164/166	0.0456	0.0467	0.04796	0.0575	
		167/166	0.7416	0.6866	0.6841	0.6891	
		168/166	0.8176	0.8102	0.8100	0.8071	
		170/166	0.4316	0.4454	0.4508	0.4502	
Atomic weight		167.248 (1)		167.262 (1)	167.264 (6)	167.247 (7)	
Error		NS		P	P	NS	
Annotation							

Element ^{68}Er Erbium--continued

Mass no.	Nuclidic mass	81HOL1	83CICA1
162	161.928787	0.137 (1)	0.14 (1)
164	163.929211	1.609 (5)	1.61 (4)
166	165.930305	33.61 (7)	33.6 (2)
167	166.932061	22.93 (5)	22.95 (13)
168	167.932383	26.79 (7)	26.8 (2)
170	169.935476	14.93 (5)	14.9 (1)
Isotope ratio	162/166	0.00408	0.00417
	164/166	0.0389	0.0479
	167/166	0.6822	0.6830
	168/166	0.7971	0.7976
	170/166	0.4442	0.4435
Atomic weight		167.257 (2)	167.256 (4)
Error		2XSD	
Annotation		B	

Element ^{69}Tm Thulium

Mass no.	Nuclidic mass	50LAG1	57COL1	83ICA1
169	168.934225	100	100	100
Isotope ratio				
Atomic weight		168.93423 (4)	168.93423 (4)	168.93423 (4)
Error		NS	NS	
Annotation				

Element $_{70}\text{Yb}$ Ytterbium

1961 173.04 1969 173.04 (3)

Mass no.	Nuclidic mass	41WAH1	49HAY1	50LEL1	57COL1
168	167.933908	0.06	0.140 (2)	0.130 (5)	0.135 (2)
170	169.934774	4.21	3.034 (30)	3.03 (3)	3.140 (15)
171	170.936338	14.26	14.34 (14)	14.27 (14)	14.4 (2)
172	171.936393	21.49	21.88 (22)	21.77 (22)	21.90 (25)
173	172.938222	17.02	16.18 (16)	16.08 (16)	16.2 (2)
174	173.938873	29.58	31.77 (32)	31.91 (32)	31.6 (4)
176	175.942576	13.38	12.65 (13)	12.80 (13)	12.60 (15)
Isotope ratio 168/174	0.00203	0.00441	0.00407	0.00427	
170/174	0.1423	0.0955	0.0950	0.0994	
171/174	0.4820	0.4514	0.4472	0.4557	
172/174	0.7264	0.6887	0.6822	0.6930	
173/174	0.5753	0.5093	0.5039	0.5126	
176/174	0.4522	0.3982	0.4011	0.3987	
Atomic weight	173.0060 (7)	173.031 (6)	173.041 (6)	173.024 (6)	
Error	NS	P	P	SD	

Annotation

Mass no.	Nuclidic mass	66KOM1	77MC1	81HOL1	83ICA1
168	167.933908	0.22 (9)	0.136 (1) ^c	0.127 (2)	0.13 (1)
170	169.934774	3.43 (10)	3.063 (3) ^c	3.04 (2)	3.05 (5)
171	170.936338	14.06 (30)	14.334 (9) ^c	14.28 (8)	14.3 (2)
172	171.936393	21.82 (15)	21.879 (10) ^c	21.83 (10)	21.9 (3)
173	172.938222	16.05 (18)	16.122 (9) ^c	16.13 (7)	16.12 (18)
174	173.938873	31.53 (30)	31.768 (20) ^c	31.83 (14)	31.8 (4)
176	175.942576	12.86 (18)	12.698 (6) ^c	12.76 (5)	12.7 (1)
Isotope ratio 168/174	0.0070	0.00429 (5)	0.00399	0.00409	
170/174	0.1088	0.0964 (1)	0.0955	0.0959	
171/174	0.4459	0.4512	0.4486	0.4497	
172/174	0.6920	0.6887 (4)	0.6858	0.6887	
173/174	0.5090	0.5075 (3)	0.5068	0.5069	
176/174	0.4079	0.3997 (2)	0.4009	0.3994	
Atomic weight	173.02 (1)	173.0327 (3)	173.038 (3)	173.034 (7)	
Error	NS	2XSE	2XSD		
Annotation		B			

Element $_{71}\text{Lu}$ Lutetium

1961 174.97 1977 174.967 (3) 1981 174.967 (1)

Mass no.	Nuclidic mass	39MAT2	50HAY1	57COL1	76MCC1
175	174.940785	97.485 (6) ^c	97.40 (3)	97.412 (13)	97.393 (5)
176	175.942694	2.515 (67) ^c	2.60 (3)	2.588 (13)	2.607 (5)
Isotope ratio 176/175	0.0258 (7)		0.02669	0.02657	0.02677 (5)
Atomic weight	174.9660 (7)	174.9668 (3)	174.9667 (1)	174.96690 (5)	
Error		NS	P	SD	2XSD
Annotation					

Mass no.	Nuclidic mass	81HOL1	83PAT1	83ICA1
175	174.940785	97.41 (2)	97.416 (5) ^c	97.41 (2)
176	175.942694	2.59 (2)	2.584 (5) ^c	2.59 (2)
Isotope ratio 176/175	0.02659		0.026525 (20)	0.02669
Atomic weight	174.9667 (2)	174.96667 (5)	174.9667 (2)	
Error		2XSD	2XSD	
Annotation			B	

Element $_{72}\text{Hf}$ Hafnium

1961 178.49 1969 178.49 (3)

Mass no.	Nuclidic mass	43MAT1	49HIB2	53REY1	56WH11
174	173.940065	0.18 (1)	0.18 (1)	0.199 ^{+0.003} _{-0.010}	0.163 (2)
176	175.941420	5.30 (11)	5.15 (2)	5.23 (5)	5.21 (2)
177	176.943233	18.47 (9)	18.39 (1)	18.55 (17)	18.56 (6)
178	177.943710	27.13 (14)	27.08 (4)	27.23 (22)	27.10 (10)
179	178.945827	13.85 (7)	13.78 (2)	13.73 (13)	13.75 (5)
180	179.946561	35.14 (18)	35.44 (6)	35.07 (24)	35.22 (10)
Isotope ratio 174/180	0.00508	0.00508	0.00567	0.00463	
176/180	0.1509	0.1453	0.1491	0.1479	
177/180	0.5260	0.5189	0.5289	0.5270	
178/180	0.7715	0.7641	0.7765	0.7695	
179/180	0.3944	0.3888	0.3915	0.3904	
Atomic weight	178.488 (4)	178.497 (1)	178.485 (5)	178.491 (2)	
Error		P	SD	P	NS
Annotation					

Element $_{72}^{\text{Hf}}$ Hafnium--continued

Mass no.	Nuclidic mass	75TAM1	83PATT	83ICA1
174	173.940065	0.155 (1) ^c	0.1621 (9)	0.162 (2)
176	175.941420	5.193 (19) ^c	5.2056 (12)	5.206 (4)
177	176.943233	18.484 (37) ^c	18.6060 (13)	18.606 (3)
178	177.943710	27.261 (27) ^c	27.2969 (13)	27.297 (3)
179	178.945827	13.652 (35) ^c	13.6289 (19)	13.629 (5)
180	179.946561	35.255 (1) ^c	35.1005 (22)	35.100 (6)
Isotope ratio	174/180	0.00438 (1)	0.004618	0.00455
	176/180	0.1473 (4)	0.148306	0.1477
	177/180	0.5243 (15)	0.530078	0.5204
	178/180	0.7733 (23)	0.777678	0.7699
	179/180	0.3872 (1)	0.388282	0.3903
Atomic weight		178.4915 (8)	178.48643 (7)	178.490 (9)
Error		NS	NS	
Annotation		B		

Element $_{73}^{\text{Ta}}$ Tantalum

	1961	180.948	1969	180.9479 (3)	1979	180.9479	
Mass no.		Nuclidic mass		55WHI1	56WHI1	58PAL1	83ICA1
180	179.947489		0.0123 (3)	0.0123 (3)	0.0117 (6)	0.012 (2)	
181	180.948014		99.988	99.9877 (3)	99.9883 (6)	99.988 (2)	
Isotope ratio	180/181		0.0001230	0.0001233	0.0001170	0.0001200	
Atomic weight		180.947891(3)		180.947891 (3)	180.947897 (6)	180.94789 (2)	
Error		NS		NS	NS		
Annotation				B			

Element $_{74}W$ Tungsten

1961 183.85 1969 183.85 (3)

Mass no.	Nuclidic mass	46ING1	46WIL1	48MAT1	48WHI1
180	179.946727	0.122 (2)	0.130 (2)	0.160 (4)	0.126 (6)
182	181.948225	25.77 (30)	26.41 (53)	26.35 (8)	26.31 (3)
183	182.950245	14.24 (20)	14.40 (29)	14.32 (11)	14.28 (1)
184	183.950953	30.68 (30)	30.64 (61)	30.68 (12)	30.64 (3)
186	185.954377	29.17 (30)	28.41 (57)	28.49 (9)	28.64 (1)
Isotope ratio 180/183	0.00857	0.0090	0.01117	0.00882	
182/183	1.8097	1.834	1.8401	1.8424	
184/183	2.1545	2.128	2.1425	2.1457	
186/183	2.0485	1.973	1.9895	2.0056	
Atomic weight	183.872 (6)	183.84 (2)	183.844 (3)	183.8498 (7)	
Error	P	P	NS	SE	
Annotation			B		

Mass no.	Nuclidic mass	49HIB2	600MU1	83ICA1
180	179.946727	0.143 (2)	0.116 (10)	0.13 (3)
182	181.948225	26.09 (12)	26.47 (12)	26.3 (2)
183	182.950245	14.24 (2)	14.27 (7)	14.3 (1)
184	183.950953	30.68 (1)	30.62 (9)	30.67 (15)
186	185.954377	28.85 (1)	28.55 (12)	28.6 (2)
Isotope ratio 180/183	0.0100	0.00813	0.00909	
182/183	1.8322	1.8549	1.8392	
184/183	2.1545	2.1458	2.1448	
186/183	2.0260	2.0007	2.0000	
Atomic weight	183.8582 (3)	183.845 (3)	183.849 (6)	
Error	SD	NS		
Annotation				

Element $_{75}Re$ Rhenium

1961 186.2 1973 186.207

Mass no.	Nuclidic mass	47HES1	48WHI1	73GRA1	83ICA1
185	184.952977	37.244 (69)	37.07 (6)	37.398 (16)	37.40 (2)
187	186.955765	62.756 (69)	62.93 (6)	62.602 (16)	62.60 (2)
Isotope ratio 185/187	0.5935 (18)	0.5891	0.59738 (39)	0.59744	
Atomic weight	186.210 (1)	186.213 (1)	186.2068 (3)	186.2068 (3)	
Error	NS	SE	3XSD		
Annotation			C, B		

Element ^{76}Os Osmium

1961 190.2

Mass no.	Nuclidic mass	37NIE1	83ICAT
184	183.952514	0.018 (2)	0.02 (1)
186	185.953852	1.59 (5)	1.58 (10)
187	186.955762	1.64 (5)	1.6 (1)
188	187.955850	13.27 (11)	13.3 (2)
189	188.958156	16.14 (15)	16.1 (3)
190	189.958455	26.38 (20)	26.4 (4)
192	191.961487	40.96 (14)	41.0 (3)
Isotope ratio 184/188	0.009	0.0015	
186/188	0.120	0.119	
187/188	0.124	0.120	
189/188	1.216	1.210	
190/188	1.988	1.985	
192/188	3.087	3.083	
Atomic weight	190.238 (5)	190.24 (1)	
Error	P		
Annotation	B		

Element ^{77}Ir Iridium

1961 192.2 1969 192.22 (3)

Mass no.	Nuclidic mass	54BAL1	83ICAT
191	190.960603	37.3	37.3 (5)
193	192.962942	62.7	62.7 (5)
Isotope ratio 191/193	0.5949	0.5949	
Atomic weight	192.216 (2)	192.22 (1)	
Error	NS		
Annotation	B		

Element $_{78}\text{Pt}$ Platinum

1961 195.09 1969 195.09 (3)

Mass no.	Nuclidic mass	47ING3	56WH11	83ICA1
190	189.959937	0.09	0.0127 (5)	0.01 (1)
192	191.961049	0.78	0.78 (1)	0.79 (5)
194	193.962679	32.8	32.9 (1)	32.9 (5)
195	194.964785	33.7	33.8 (1)	33.8 (5)
196	195.964947	25.4	25.2 (1)	25.3 (5)
198	197.967879	7.23	7.19 (4)	7.2 (2)
Isotope ratio	190/195 192/195 194/195 196/195 198/195	0.0027 0.0231 0.9733 0.7537 0.2145	0.000376 0.02308 0.9734 0.7456 0.2127	0.00030 0.02237 0.9734 0.7486 0.2130
Atomic weight		195.079 (2)	195.079 (2)	195.080 (9)
Error		NS	NS	
Annotation			B	

Element $_{79}\text{Au}$ Gold

1961 196.967 1969 196.9665

Mass no.	Nuclidic mass	63LEI1	83ICA1
197	196.966560	100	100
Isotope ratio			
Atomic weight		196.96656 (3)	196.96656 (4)
Error		NS	
Annotation			

Element ^{80}Hg Mercury

1961 200.59 1969 200.59 (3)

Mass no.	Nuclidic mass	47ING1	49HIB2	50NIE2	50NIE2
196	195.965812	0.155	0.16 (3)	0.147 (4)	0.146 (20) ^c
198	197.966760	10.12	10.02 (6)	10.067 (101)	10.018 (9) ^c
199	198.968269	17.01	16.92 (7)	17.000 (170)	16.837 (15) ^c
200	199.968316	23.21	23.10 (8)	23.050 (230)	23.127 (15) ^c
201	200.970293	13.15	13.22 (6)	13.224 (132)	13.222 (14) ^c
202	201.970632	29.66	29.72 (9)	29.731 (297)	29.799 (18) ^c
204	203.973481	6.69	6.84 (6)	6.781 (68)	6.851 (5) ^c
Isotope ratio 196/202	0.0052	0.0054	0.00494	0.00491	
198/202	0.3412	0.3372	0.3386	0.3362 (3)	
199/202	0.5735	0.5693	0.5718	0.5650 (5)	
200/202	0.7825	0.7773	0.7753	0.7761 (5)	
201/202	0.4434	0.4448	0.4448	0.4437 (6)	
204/202	0.2256	0.2302	0.2281	0.2299 (2)	
Atomic weight	200.5830 (5)	200.594 (3)	200.590 (6)	200.597 (1)	
Error	NS	SD	P	P	
Annotation			B		

Mass no.	Nuclidic mass	55DIB1	83ICA1
196	195.965812	0.156 (1)	0.14 (10)
198	197.966760	10.12 (10)	10.02 (7)
199	198.968269	16.99 (9)	16.84 (11)
200	199.968316	23.07 (12)	23.13 (11)
201	200.970293	13.27 (7)	13.22 (11)
202	201.970632	29.64 (15)	29.80 (14)
204	203.973481	6.97 (5)	6.85 (5)
Isotope ratio 196/202	0.005263	0.00506	
198/202	0.3414	0.3406	
199/202	0.5732	0.5734	
200/202	0.7783	0.7791	
201/202	0.4477	0.4452	
204/202	0.2291	0.2293	
Atomic weight	200.588 (4)	200.59 (2)	
Error	P		
Annotation			

Element $_{81}\text{Tl}$ Thallium

1961 204.37 1969 204.37 (3) 1979 204.383

Mass no.	Nuclidic mass	38NIE1	48PAU1	48WH11	49HIB2
203	202.972336	29.08 (40) ^c	30.07 (49)	29.46 (5)	29.52 (5)
205	204.974410	10.92 (40) ^c	69.93 (49)	70.54 (5)	70.48 (7)
Isotope ratio 203/205		0.410 (8)	0.4300	0.4176	0.4188
Atomic weight		204.393 (8)	204.37 (1)	204.385 (1)	204.383 (1)
Error		NS	3XSD	P	SD
Annotation					

Mass no.	Nuclidic mass	78MUR2	80DUN1	83ICA1
203	202.972336	29.665 (45) ^c	29.524 (9)	29.524 (9)
205	204.974410	70.335 (45) ^c	70.476 (9)	70.476 (9)
Isotope ratio 203/205		0.4218 (9)	0.41891 (18)	0.41892
Atomic weight		204.3805 (9)	204.3833 (2)	204.3833 (2)
Error		SD	3XSD	
Annotation			C, B	

Element $_{82}\text{Pb}$ Lead

1961 207.19 1969 207.2

Mass no.	Nuclidic mass	38NIE3	38NIE3	41NIE1	41NIE1
204	203.973037	1.48	1.26	1.544 (19) ^c	1.230 (15) ^c
206	205.974455	23.59	27.31	22.61 (43) ^c	27.47 (50) ^c
207	206.975885	22.64	20.00	22.62 (43) ^c	19.87 (39) ^c
208	207.976641	52.29	51.43	53.23 (60) ^c	51.43 (61) ^c
Isotope ratio 204/206		0.0627	0.0461	0.0683	0.0448
207/206		0.9597	0.7323	1.0004	0.7233
208/206		2.2166	1.8832	2.3543	1.8722
Atomic weight		207.2185 (4)	207.1792 (4)	207.236 (7)	207.179 (8)
Error		P	NS	P	P
Annotation					

Element ^{82}Pb Lead--continued

Mass no.	Nuclidic mass	48WH11	50HIB1	50HIB1	52COL1
204	203.973037	1.37 (14)	1.372 (16)	1.360 (5)	1.647 (10) ^c
206	205.974455	25.15 (25)	26.24 (2)	26.29 (3)	20.84 (21) ^c
207	206.975885	21.11 (21)	20.82 (3)	20.82 (4)	23.51 (22) ^c
208	207.976641	52.38 (52)	51.57 (3)	51.53 (3)	54.00 (30) ^c
Isotope ratio 204/206	0.0545	0.0523	0.0517	0.0790	
207/206	0.8394	0.7934	0.7919	1.1281	
208/206	2.0827	1.9653	1.9601	2.5912	
Atomic weight	207.207 (5)	207.188 (6)	207.1875 (5)	207.258 (4)	
Error	P	2XSD	2XSD	P	
Annotation					

Mass no.	Nuclidic mass	52COL1	52DIB1	53ALL1	53ALL1
204	203.973037	1.0400 (66)	1.32 (2)	1.52	1.65
206	205.974455	27.04 (25)	26.67 (20)	22.54	20.84
207	206.975885	17.62 (18)	20.50 (20)	22.70	23.51
208	207.976641	54.30 (30)	51.50 (30)	53.24	54.00
Isotope ratio 204/206	0.0385	0.0495	0.0674	0.0792	
207/206	0.6516	0.7687	1.0071	1.1281	
208/206	2.0081	1.9310	2.3620	2.5912	
Atomic weight	207.217 (4)	207.185 (3)	207.2373 (4)	207.2580 (4)	
Error	P	P	NS	NS	
Annotation					

Mass no.	Nuclidic mass	53EHR1	53EHR1	53FAR1	53FAR1
204	203.973037	1.3478 (60) ^c	1.3560 (51) ^c	1.440 (9)	1.455 (7)
206	205.974455	24.824 (64) ^c	25.158 (23) ^c	23.69 (2)	23.64 (5)
207	206.975885	21.448 (79) ^c	21.133 (60) ^c	22.54 (4)	22.61 (3)
208	207.976641	52.38 (12) ^c	52.354 (56) ^c	52.33 (4)	52.30 (7)
Isotope ratio 204/206	0.0543 (2)	0.0539 (2)	0.0608	0.0615	
207/206	0.864 (3)	0.8400 (3)	0.9515	0.9564	
208/206	2.110 (10)	2.0810 (2)	2.2089	2.2124	
Atomic weight	207.211 (1)	207.2071 (6)	207.2191 (5)	207.2188 (8)	
Error	NS	NS	SD	SD	
Annotation					

Element ^{82}Pb Lead--continued

Mass no.	Nuclidic mass	54BEG1	54BEG1	54EHR1	54EHR1	54GEI1
204	203.973037	1.33	1.32	1.3528 (52) ^c	1.3899 (72) ^c	1.3517 (80) ^c
206	205.974455	25.43	25.38	25.006 (28) ^c	24.820 (93) ^c	25.172 (52) ^c
207	206.975885	20.94	20.90	21.280 (45) ^c	21.420 (89) ^c	21.044 (58) ^c
208	207.976641	52.30	52.40	52.362 (54) ^c	52.37 (18) ^c	52.433 (99) ^c
Isotope ratio 204/206	0.0523	0.0520	0.0541 (2)	0.0560 (2)	0.0537 (3)	
207/206	0.8234	0.8235	0.851 (2)	0.8630 (2)	0.836 (2)	
208/206	2.0566	2.0646	2.094 (4)	2.110 (15)	2.083 (8)	
Atomic weight	207.2047 (4)	207.2065 (4)	207.2088 (6)	207.209 (2)	207.207 (1)	
Error	NS	NS	NS	NS	NS	3XSD
Annotation						

Mass no.	Nuclidic mass	54GEI1	54RUS1	54RUS1	55EBE1
204	203.973037	1.3846 (55) ^c	1.43	1.64	1.3237 (52) ^c
206	205.974455	24.592 (46) ^c	23.30	20.84	25.455 (28) ^c
207	206.975885	21.986 (67) ^c	22.77	23.52	20.962 (37) ^c
208	207.976641	52.037 (91) ^c	52.51	54.02	52.259 (53) ^c
Isotope ratio 204/206	0.0563 (2)	0.0614	0.0787	0.0520 (2)	
207/206	0.894 (3)	0.9773	1.1286	0.8235 (15)	
208/206	2.116 (7)	2.2536	2.5921	2.053 (4)	
Atomic weight	207.2080 (9)	207.2251 (4)	207.2584 (4)	207.2042 (6)	
Error	3XSD	NS	NS	NS	3XSD
Annotation					

Mass no.	Nuclidic mass	55EBE1	55SAK1	55SAK1	56KUL1
204	203.973037	1.6473 (42) ^c	1.37	1.34	1.21
206	205.974455	20.9707 (72) ^c	25.03	25.03	27.48
207	206.975885	23.592 (24) ^c	21.08	21.18	19.53
208	207.976641	53.790 (18) ^c	52.53	52.45	51.78
Isotope ratio 204/206	0.07855 (20)	0.0547	0.0535	0.0440	
207/206	1.1250 (15)	0.8422	0.8462	0.7107	
208/206	2.565 (6)	2.0987	2.0955	1.8843	
Atomic weight	207.2547 (2)	207.2098 (4)	207.2099 (4)	207.1825 (4)	
Error	3XSD	NS	NS	NS	
Annotation					

Element ^{82}Pb Lead--continued

Mass no.	Nuclidic mass	56KUL1	56WH11	57RUS1	57RUS1
204	203.973037	1.35	1.40 (2)	1.64	1.07
206	205.974455	25.24	25.2 (1)	20.99	24.81
207	206.975885	21.12	21.7 (1)	23.57	17.91
208	207.976641	52.29	51.7 (2)	53.80	56.21
Isotope ratio 204/206	0.0535	0.0556	0.0781	0.0431	
207/206	0.8368	0.8611	1.1229	0.7219	
208/206	2.0717	2.0516	2.5631	2.2656	
Atomic weight	207.2059 (4)	207.199 (2)	207.2548 (4)	207.2578 (4)	
Error	NS	NS	NS	NS	
Annotation					

Mass no.	Nuclidic mass	58EHR1	58EHR1	58PAL1
204	203.973037	1.396 (15) ^c	1.3569 (51) ^c	1.560 (15)
206	205.974455	24.577 (62) ^c	25.059 (21) ^c	22.50 (17)
207	206.975885	21.677 (72) ^c	21.187 (34) ^c	22.61 (17)
208	207.976641	52.35 (12) ^c	52.397 (41) ^c	53.33
Isotope ratio 204/206	0.0568 (6)	0.05415 (2)	0.0693	
207/206	0.8820 (25)	0.8455 (15)	1.0049	
208/206	2.130 (10)	2.091 (3)	2.3702	
Atomic weight	207.212 (1)	207.2086 (4)	207.237 (3)	
Error	NS	NS	NS	
Annotation				

Mass no.	Nuclidic mass	58SAK1	58SAK1	58TIL1	58TIL1
204	203.973037	1.3537 (30) ^c	1.3528 (30) ^c	1.3330 (67) ^c	1.3587 (77) ^c
206	205.974455	25.058 (74) ^c	25.054 (74) ^c	25.473 (27) ^c	25.203 (67) ^c
207	206.975885	21.118 (70) ^c	21.131 (70) ^c	21.000 (83) ^c	21.269 (99) ^c
208	207.976641	52.47 (11) ^c	52.46 (11) ^c	52.194 (56) ^c	52.17 (13) ^c
Isotope ratio 204/206	0.05402	0.05400	0.05233 (26)	0.05391 (27)	
207/206	0.84277	0.84342	0.8244 (41)	0.8439 (42)	
208/206	2.09394	2.09388	2.049 (10)	2.070 (10)	
Atomic weight	207.210 (1)	207.210 (1)	207.2031 (6)	207.204 (1)	
Error	SE	SE	P	P	
Annotation					

Element ^{82}Pb Lead--continued

Mass no.	Nuclidic mass	59FER1	60CAT1	60CAT1	60KOL1
204	203.973037	1.366 (16) ^c	1.568 (13) ^c	1.3658 (99) ^c	1.4551 (15) ^c
206	205.974455	24.99 (14) ^c	22.13 (32) ^c	24.83 (29) ^c	23.450 (11) ^c
207	206.975885	21.29 (20) ^c	23.51 (32) ^c	21.41 (25) ^c	22.615 (19) ^c
208	207.976641	52.35 (25) ^c	52.79 (42) ^c	52.40 (36) ^c	52.480 (23) ^c
Isotope ratio 204/206		0.05465 (55)	0.0708 (11)	0.0550 (7)	0.06205 (6)
207/206		0.8520 (85)	1.062 (10)	0.862 (8)	0.9644 (9)
208/206		2.0950 (210)	2.383 (20)	2.110 (14)	2.2380 (18)
Atomic weight		207.208 (3)	207.235 (5)	207.211 (5)	207.2226 (2)
Error		NS	NS	NS	SD
Annotation					

Mass no.	Nuclidic mass	62DOE1	62DOE1	63RIC1	63RIC1
204	203.973037	1.46	1.37	1.425 (2)	1.458 (2)
206	205.974455	23.42	24.89	24.56 (1)	23.48 (1)
207	206.975885	22.62	21.45	22.36 (2)	22.54 (2)
208	207.976641	52.50	52.29	51.65 (2)	52.52 (2)
Isotope ratio 204/206		0.0622	0.0549	0.05802	0.06208
207/206		0.9658	0.8621	0.91042	0.95997
208/206		2.2421	2.1011	2.10301	2.23680
Atomic weight		207.2437 (4)	207.2088 (4)	207.2041 (2)	207.2189 (2)
Error		NS	NS	SE	SD
Annotation					

Mass no.	Nuclidic mass	66TAT1	66TAT1	68CAT1	69GRA1
204	203.973037	1.346	1.318	1.4245 (12)	1.39
206	205.974455	24.795	25.797	24.1447 (57)	25.10
207	206.975885	21.103	20.771	22.0827 (27)	21.36
208	207.976641	52.756	52.114	52.3481 (86)	52.15
Isotope ratio 204/206		0.0543	0.0511	0.059042 (37)	0.0553
207/206		0.8511	0.8052	0.91464 (33)	0.8508
208/206		2.1277	2.0202	2.1681 (6)	2.0774
Atomic weight		207.21512 (4)	207.19950 (4)	207.2152 (1)	207.2047 (4)
Error		NS	NS	3XSD	NS
Annotation				C, B	

Element $_{82}^{\text{Pb}}$ Lead--continued

Mass no.	Nuclidic mass	75BAR3	78YAM1	78YAM1	83ICA1
204	203.973037	1.3439 (7)	1.600	1.617	1.4 (1)
206	205.974455	25.353 (13)	24.771	25.194	24.1 (1)
207	206.975885	21.075 (13)	21.263	21.163	22.1 (1)
208	207.976641	52.228 (26)	52.366	52.026	52.4 (1)
Isotope ratio 204/206	0.05300	0.0646	0.0642	0.0581	
207/206	0.83127	0.8584	0.8400	0.9170	
208/206	2.0600	2.114	2.085	2.1743	
Atomic weight	207.2043 (3)	207.20383 (4)	207.19568 (4)	207.217 (4)	
Error	2XSD	NS	NS	NS	
Annotation					

Element $_{83}^{\text{Bi}}$ Bismuth

1961	208.980	1969	208.9806	1971	208.9804
Mass no.	Nuclidic mass		38NIE1		63LEI1
209	208.980388	100		100	100
Isotope ratio					
Atomic weight		208.98039 (3)	208.98039 (3)	208.98039 (3)	
Error		NS		NS	
Annotation					

Element $_{90}^{\text{Th}}$ Thorium

1961	232.038	1969	232.0381
Mass no.	Nuclidic mass		36DEM1
232	232.038053805	100	100
Isotope ratio			
Atomic weight		232.03805 (2)	232.03805 (2)
Error		NS	
Annotation			

Element ^{92}U Uranium

1961 238.03 1969 238.029 1979 238.0289

Mass no.	Nuclidic mass	39NIE1	46CHA1	46FOX1	46FOX1
234	234.040947400	0.00591 (60) ^c	0.005040 (30) ^c	0.00555 (17) ^c	0.00555 (17) ^c
235	235.043925247	0.7148 (71) ^c	0.7131 (60) ^c	0.7246 (36) ^c	0.7193 (16) ^c
238	238.050785782	99.2793 (71) ^c	99.2818 (60) ^c	99.2699 (37) ^c	99.2751 (16) ^c
Isotope ratio 234/238 235/238		0.000060 (6) 0.007200 (72)	0.000051 (3) 0.007183	0.000056 (2) 0.007299 (36)	0.000056 (2) 0.007246 (16)
Atomic weight		238.0291 (2)	238.0291 (2)	238.0288 (1)	238.02893 (5)
Error		P	P	NS	NS
Annotation					

Mass no.	Nuclidic mass	56LOU1	56WH11
234	234.040947400	0.0057 (2) ^c	0.0056 (1)
235	235.043925247	0.7204 (7) ^c	0.718 (5)
238	238.050785782	99.2739 (7) ^c	99.276 (5)
Isotope ratio 234/238 235/238		0.000058 (2) 0.007257 (7)	0.000056 0.007232
Atomic weight		238.02890 (2)	238.0289 (2)
Error		SD	NS
Annotation			

Mass no.	Nuclidic mass	69SMI1 76COW1	69SMII 76COW1	83ICAI
234	234.040947400	0.005448 (2)	0.00508 (2)	0.0055 (5)
235	235.043925247	0.7200 (1)	0.7196 (1)	0.7200 (12)
238	238.050785782	99.2745 (10)	99.2753 (10)	99.2745 (15)
Isotope ratio 234/238 235/238		0.000055 0.007253	0.000051 0.007249	0.0000554 0.0072526
Atomic weight		238.02879 (3)	238.02890 (3)	238.0289 (1)
Error		SD	SD	
Annotation		B	B	

4. References

- 20AST1 F.W. Aston, Phil. Mag. **40**, 628 (1920). The Mass Spectra of Chemical Elements.
- 32AST1 F.W. Aston, Proc. Roy. Soc. A**134**, 571 (1932). The Isotopic Constitution and Atomic Weights of Cesium, Strontium, Lithium, Rubidium, Barium, Scandium and Thallium.
- 34VAU1 A.L. Vaughan, J.H. Williams, and J.T. Tate, Phys. Rev. **46**, 327 (1934). Isotopic Abundance Ratios of C, N, A, Ne and He.
- 35BRE1 A.K. Brewer, Phys. Rev. **48**, 640 (1935). Further Evidence for the Existence of 40K.
- 35NIE1 A.O. Nier, Phys. Rev. **48**, 283 (1935). Evidence for the Existence of an Isotope of Potassium of Mass 40.
- 36BLE1 J.P. Blewett, Phys. Rev. **49**, 900 (1936). Mass Spectrograph Analysis of Bromine.
- 36DEM1 A.J. Dempster, Nature **136**, 120 (1936). Atomic Masses of Uranium and Thorium.
- 36HAL1 N.F. Hall and T.O. Jones, J. Am. Chem. Soc. **58**, 1915 (1936). A Redetermination of the Protium-Deuterium Ratio in Water.
- 36NIE1 A.O. Nier and E.E. Hanson, Phys. Rev. **50**, 722 (1936). Mass Spectrographic Analysis of the Ions Produced in HCl Under Electron Impact.
- 37BRO1 A.R. Brosi and W.D. Harkins, Phys. Rev. **52**, 472 (1937). The Abundance Ratio of the Isotopes in Natural or Isotopically Separated Carbon.
- 37NIE1 A.O. Nier, Phys. Rev. **52**, 885 (1937). The Isotopic Constitution of Osmium.
- 37NIE2 A.O. Nier, Phys. Rev. **52**, 933 (1937). A Mass Spectrographic Study of the Isotopes of Hg, Xe, Kr, Be, I, As, and Cs.
- 38BRE1 A.K. Brewer, J. Am. Chem. Soc. **60**, 691 (1938). A Mass Spectrographic Determination of the Isotope Abundance and of the Atomic Weight of Rubidium.
- 38MOR1 N. Morita and T. Titani, Bull. Chem. Soc. Japan **13**, 419 (1938). Wiederbestimmung des Deuterium-Protium-Verhältnisses in gewöhnlichem Wasser.
- 38NIE1 A.O. Nier, Phys. Rev. **54**, 275 (1938). The Isotopic Constitution of Strontium, Barium, Bismuth, Thallium and Mercury.
- 38NIE2 A.O. Nier, Phys. Rev. **53**, 282 (1938). The Isotopic Constitution of Calcium, Titanium, Sulfur and Argon.
- 38NIE3 A.O. Nier, J. Am. Chem. Soc. **60**, 1571 (1938). Variations in the Relative Abundances of the Isotopes of Common Lead from Various Sources.
- 38VOS1 J.T. Voskuyl, Thesis, Harvard Univ. (1938).
- 39MAT1 J. Mattauch and H. Lichtblau, Z. Phys. Chem. **42**, 288 (1939). Die Isotopenzusammensetzung und das Atomgewicht von Molybdänen.
- 39MAT2 J. Mattauch and H. Lichtblau, Z. Phys. **111**, 514 (1939). Ein bemerkenswertes Isotop des Cassiopeiums.
- 39NIE1 A.O. Nier, Phys. Rev. **55**, 150 (1939). The Isotopic Constitution of Uranium and the Half-Lives of the Uranium Isotopes.
- 39NIE2 A.O. Nier, Phys. Rev. **55**, 1143 (1939). The Isotopic Constitution of Iron and Chromium.
- 39NIE3 A.O. Nier and E.A. Gulbranson, J. Am. Chem. Soc. **61**, 697 (1939). Variations in the Relative Abundance of the Carbon Isotopes.
- 39SWA1 J.A. Swartout and M. Dole, J. Am. Chem. Soc. **61**, 2025 (1939). The Protium-Deuterium Ratio and the Atomic Weight of Hydrogen.
- 40VAL1 G.E. Valley, Phys. Rev. **57**, 1058 (1940). Abundance of Molybdenum Isotopes.
- 41MUR1 B.F. Murphrey, Phys. Rev. **59**, 320 (1941). Relative Abundances of the Oxygen Isotopes.
- 41MUR2 B.F. Murphrey and A.O. Nier, Phys. Rev. **59**, 771 (1941). Variations in the Relative Abundance of the Carbon Isotopes.
- 41NIE1 A.O. Nier, R.W. Thompson, and B.F. Murphrey, Phys. Rev. **60**, 112 (1941). The Isotopic Constitution of Lead and the Measurement of Geological Time.
- 41STR1 H.A. Straus, Phys. Rev. **59**, 430 (1941). A New Mass Spectrograph and the Isotopic Constitution of Nickel.
- 41VAL1 G.E. Valley and H.H. Anderson, Phys. Rev. **59**, 113 (1941). The Relative Abundance of the Stable Isotopes of Terrestrial and Meteoritic Iron.
- 41VAL2 G.E. Valley, Phys. Rev. **59**, 836 (1941). The Stable Isotopes of Nickel.
- 41WAH1 W. Wahl, Naturwiss. **29**, 536 (1941). Die Isotopenzusammensetzung und das Atomgewicht von Ytterbium.
- 41WAH2 W. Wahl, Finska Kemists Afhandl. Medd. **50**, 10 (1941). Eine approximative Bestimmung der Isotopenzusammensetzung von Erbium.
- 43MAT1 J. Mattauch and H. Ewald, Naturwiss. **31**, 487 (1943). Ueber eine neue Methode zur Messung relativer Häufigkeiten von Isotopen. Die Isotopenverteilung und das Atomgewicht von Hafnium.
- 44EWA1 H. Ewald, Z. Phys. **122**, 487 (1944). Die relativen Isotopenhäufigkeiten und die Atomgewichte von Kupfer und Ruthenium.
- 44EWA2 H. Ewald, Z. Phys. **122**, 686 (1944). Photometrische Bestimmung seltener Isotope. Relative Isotopenhäufigkeiten und Atomgewicht von Nickel.
- 44THO1 H.G. Thode and S.R. Smith, Nat. Res. Coun. Can. Atom. Energ. Proj. Rpt. Mc-57 (1944). The Natural Abundances of the Oxygen Isotopes.
- 46ALD1 L.T. Aldrich and A.O. Nier, Phys. Rev. **70**, 983 (1946). The Abundance of ^3He in Atmospheric and Well Helium.
- 46CHA1 O. Chamberlain, D. Williams, and P. Yuster, Phys. Rev. **70**, 580 (1946). Half-Life of Uranium-234.
- 46FOX1 M. Fox and B. Rustad, USAEC Rept. A-3828 (1946). Abundance Ratios of the Uranium Isotopes.
- 46ING1 M.G. Inghram, Phys. Rev. **70**, 653 (1946). The Isotopic Constitution of Tungsten, Silicon, and Boron.
- 46NEY1 E.P. Ney and J.H. McQueen, Phys. Rev. **69**, 41 (1946). A Mass-Spectrographic Study of the Isotopes of Silicon.
- 46WIL1 D. Williams and P. Yuster, Phys. Rev. **69**, 556 (1946). Isotopic Constitution of Tellurium, Silicon, Tungsten, Molybdenum, and Bromine.
- 47BRO1 H. Brown and M.G. Inghram, Phys. Rev. **72**, 3471 (1947). The Isotopic Composition of Meteoric Copper.
- 47DIB1 V.H. Dibeler, F.L. Mohler, and R.M. Reese, J. Res. Natl. Bur. Stand. **38**, 617 (1947). Mass-Spectrometer Study of the Rare Gases.
- 47DUC1 H.E. Duckworth and B.G. Hogg, Phys. Rev. **71**, 212 (1947). Relative Abundance of the Copper Isotopes and the Suitability of the Photometric Method for Detecting Small Variations in Isotopic Abundance.
- 47FAI1 H.A. Fairbank, C.T. Lane, L.T. Aldrich, and A.O. Nier, Phys. Rev. **71**, 911 (1947). The Concentration of ^3He in the Liquid and Vapor Phases of ^4He .
- 47HES1 D.C. Hess, M.G. Inghram, and R.J. Hayden, ANL-4082, 6 (1947) Abundance of Isotopes and Enrichment Tests.
- 47ING1 M.G. Inghram, D.C. Hess Jr., and R.J. Hayden, Phys. Rev. **71**, 561 (1947). Neutron Cross Sections for the Mercury Isotopes.
- 47ING2 M.G. Inghram, R.J. Hayden, and D.C. Hess Jr., Phys. Rev. **72**, 967 (1947). The Isotopic Constitution of Lanthanum and Cerium.
- 47ING3 M.G. Inghram, D.C. Hess, and R.J. Hayden, ANL-4012 (1947). Mass Spectroscopy and Crystal Structure Abundance of Isotopes.
- 47LOU1 M. Lounsbury, S. Epstein, and H.G. Thode, Phys. Rev. **72**, 517 (1947). The Isotopic Composition of Normal Krypton and Xenon.

- 47VAL1 G.E. Valley and H.H. Anderson, J. Am. Chem. Soc. **69**, 1871 (1947). A Comparison of the Abundance Ratios of the Isotopes of Terrestrial and of Meteoritic Iron.
- 48ALD1 L.T. Aldrich and A.O. Nier, Phys. Rev. **74**, 1590 (1948). The Occurrence of ^{3}He in Natural Sources of Helium.
- 48EWA1 H. Ewald, Angew. Chem. **160**, 126 (1948). Photometrische Bestimmung der relativen Isotopenhäufigkeiten mit dem Massenspektrographen.
- 48HES1 D.C. Hess Jr., Phys. Rev. **74**, 773 (1948). The Isotopic Constitution of Europium, Gadolinium, and Terbium.
- 48HES2 D.C. Hess Jr., M.G. Inghram, and R.J. Hayden, Phys. Rev. **74**, 1531 (1948). The Relative Abundance of the Zinc Isotopes.
- 48HIB1 R.F. Hibbs and J.W. Redmond, Y-290, AECU-24 (1949). Natural Abundance Measurements On Potassium and Magnesium.
- 48HIN1 H. Hintenberger, J. Mattauch, and W. Seelmann-Eggebert, Z. Naturforsch. **3a**, 413 (1948). Ueber die Häufigkeiten der Isotope des Zircons.
- 48ING1 M.G. Inghram, D.C. Hess Jr., and R.J. Hayden, Phys. Rev. **73**, 180 (1948). Isotopic Composition of Samarium.
- 48ING2 M.G. Inghram, D.C. Hess Jr., and R.J. Hayden, Phys. Rev. **74**, 98 (1948). The Isotopic Constitution of Praseodymium and Neodymium.
- 48ING3 M.G. Inghram, D.C. Hess Jr., H.S. Brown, and E. Goldberg, Phys. Rev. **74**, 3431 (1948). On the Isotopic Composition of Meteoritic and Terrestrial Gallium.
- 48ING4 M.G. Inghram and D.C. Hess, ANL-4120, 7 (1948). Abundance of Isotopes.
- 48LEL1 W.T. Leland and A.O. Nier, Phys. Rev. **73**, 1206 (1948). The Relative Abundances of the Zinc and Cadmium Isotopes.
- 48MAT1 J. Mattauch, and H. Scheld, Z. Naturforsch. **3a**, 105 (1948). Die Isotopenzusammensetzung und die Atomgewichte von Neodym, Samarium und Wolfram.
- 48PAU1 W. Paul, Z. Phys. **124**, 244 (1948). Ein Massenspektrometer zur Bestimmung von Isotopen Mischungsverhältnissen.
- 48RAN1 K. Rankama, Bull. Geol. Soc. Am. **59**, 389 (1948). New Evidence of the Origin of Pre-Cambrian Carbon.
- 48RAN2 K. Rankama, J. Geol. **56**, 199 (1948). A Note on the Original Isotopic Composition of Terrestrial Carbon.
- 48THO1 H.G. Thode, J. Macnamara, F.P. Lossing, and C.B. Collins, J. Am. Chem. Soc. **70**, 3008 (1948). Natural Variations in the Isotopic Content of Boron and its Chemical Atomic Weight.
- 48WHI1 J.R. White, and A.E. Cameron, Phys. Rev. **74**, 991 (1948). The Natural Abundance of the Isotopes of Stable Elements.
- 49HAY1 R.J. Hayden, D.C. Hess Jr., and M.G. Inghram, Phys. Rev. **75**, 322 (1949). The Isotopic Constitution of Ytterbium.
- 49HER1 J.M. Herndon, and R.F. Hibbs, Y-508 (Oct. 28, 1949). Assay Laboratory Department Process Improvement Report, July-Sept 1949.
- 49HES1 D.C. Hess Jr., and M.G. Inghram, Phys. Rev. **76**, 1717 (1949). On the Occurrence of Vanadium 50 in Nature.
- 49HIB1 R.F. Hibbs, J.W. Redmond, H.R. Gwinn, and W.D. Harman, Phys. Rev. **75**, 533 (1949). Natural Abundance Measurements on Germanium.
- 49HIB2 R.F. Hibbs, AECU-556 (1949). Mass Spectrometric Measurements of Natural Isotopic Spectra.
- 49HIN1 H. Hintenberger, Z. Naturforsch. **4a**, 76 (1949). Ueber die Häufigkeiten der Isotope des Zirkons.
- 49ING1 M.G. Inghram, R.J. Hayden, and D.C. Hess Jr., Phys. Rev. **75**, 693 (1949). The Isotopic Constitution of Dysprosium.
- 49LEL1 W.T. Leland, Phys. Rev. **76**, 992 (1949). On the Abundance of ^{129}I , ^{118}Te and ^{190}Pt .
- 49LEL2 W.T. Leland, Phys. Rev. **76**, 1722 (1949). A Naturally Occurring Odd-Odd Isotope of Vanadium.
- 49TRO1 A. Trofimov, Doklady Akad. Nauk. SSSR **66**, 181 (1949). Isotopic Composition of Sulfur in Meteorites and Terrestrial Objects.
- 50BEC1 E.W. Becker and W. Vogell, Z. Naturforsch. **5a**, 174 (1950). Die natürliche Häufigkeit von ^{13}C und ^{18}O und die Isotopenverschiebung im Lösungsgleichgewicht Blausäure/Eisessig.
- 50HAY1 R.J. Hayden, D.C. Hess Jr. and M.G. Inghram, Phys. Rev. **77**, 299 (1950). The Isotopic Constitution of Erbium and Lutetium.
- 50HER1 J.M. Herndon and R.F. Hibbs, Y-604 (1950). Assay Laboratory Department-Process Improvement Report, Jan-Mar 1950.
- 50HIB1 R.F. Hibbs, Y-646 (1950). Assay Laboratory Department Process Improvement Report.
- 50LAG1 C.R. Lagergren and M.E. Kettner, Phys. Rev. **80**, 102 (1950). Isotopic Constitution of Thulium.
- 50LEL1 W.T. Leland, Phys. Rev. **77**, 634 (1950). The Isotopic Composition of Scandium, Gadolinium, Dysprosium, Holmium, Erbium, and Ytterbium.
- 50MAC1 J. Macnamara, and H.G. Thode, Phys. Rev. **78**, 307 (1950). Comparison of the Isotopic Constitution of Terrestrial and Meteoritic Sulfur.
- 50NIE1 A.O. Nier, Phys. Rev. **77**, 789 (1950). A Redetermination of the Relative Abundances of the Isotopes of Carbon, Nitrogen, Oxygen, Argon, and Potassium.
- 50NIE2 A.O. Nier, Phys. Rev. **79**, 450 (1950). A Redetermination of the Relative Abundances of the Isotopes of Neon, Krypton, Rubidium, Xenon, and Mercury.
- 50OSB1 O. Osberghaus, Z. Phys. **128**, 366 (1950). Die Isotopenhäufigkeit des Bors. Massenspektrometrische Untersuchung der Elektronenstossprodukte von BF_3 und BCl_3 .
- 50SCH1 A.O. Schaeffer, J. Chem. Phys. **18**, 1681 (1950). The Effect of Mass Discrimination on Isotopic Abundance Measurements. Relative Abundances of Krypton Isotopes.
- 50TRO1 A.V. Trofimov, Doklady Akad. Nauk. SSSR **72**, 663 (1950). Carbon Isotopic Ratio in Meteorites.
- 51GRA1 R.P. Graham, J. Macnamara, I.H. Crocker, and R.B. MacFarlane, Can. J. Chem. **29**, 89 (1951). The Isotopic Constitution of Germanium.
- 51KIR1 I. Kirschenbaum, Physical Properties and Analyses of Heavy Water, McGraw Hill Book Co, Inc. (1951).
- 51MAR1 K.E. Mars, J. Geol. **51**, 131 (1951). A Preliminary Investigation of the Relative Abundance of the Carbon Isotopes in Swedish Rocks.
- 51SOM1 H. Sommer and J.A. Hippel, Phys. Rev. **83**, 229 (1951). A New Method for the Measurement of the Isotopic Abundance of Solids.
- 51WIC1 F.E. Wickman and H. Von Ubisch, Geochim. Cosmochim. Acta **1**, 119 (1951). Two notes on the isotopic constitution of carbon in minerals.
- 51WIC2 F.E. Wickman, R. Blix, and H. Von Ubisch, J. Geol. **59**, 142 (1951). On the Variations in the Relative Abundance of the Carbon Isotopes in Carbonate Minerals.
- 52COL1 C.B. Collins, R.M. Farquhar, and R.D. Russell, Phys. Rev. **88**, 1275 (1952). Variations in the Relative Abundances of the Isotopes of Common Lead.
- 52DIB1 V.H. Dibeler, J. Res. Natl. Bur. Stand. **49**, 235 (1952). Mass Spectra of the Tetramethyl Compounds of Carbon, Silicon, Germanium, Tin, and Lead.
- 52MAT1 H.J. Mattraw and C.F. Pachucki, AECU-1903 (1952). Isotopic Abundances of Titanium and Nickel.
- 52NOR1 J.F. Norton and P.D. Zemany, J. Chem. Phys. **20**, 525 (1952). The Atomic Weight of Silicon.
- 52REU1 C. Reitersward, Arkiv Fysik **4** 203 (1952). The isotopic abundance of ^{40}K .

- 52WIC1 F.E. Wickman, Geochim. Cosmochim. Acta 2, 243 (1952). Variations in the relative abundance of the carbon isotopes in plants.
- 53ALD1 L.T. Aldrich, L.F. Herzog, W.K. Holyk, F.B. Whiting, and L.H. Ahrens, Phys. Rev. 89, 631 (1953). Variations in Isotopic Abundances of Strontium.
- 53ALL1 D.W. Allan, R.M. Farquhar, and R.D. Russell, Science 118, 486 (1953). A Note on the Lead Isotope Method of Age Determination.
- 53ANT1 S. Antkiv and V.H. Dibeler, J. Chem. Phys. 21, 1890 (1953). Mass Spectrum of Gallium Vapor.
- 53CRA1 H. Craig, Geochim. Cosmochim. Acta 3 53 (1953). The geochemistry of the stable carbon isotopes.
- 53DAN1 W. Dansgaard, Geochim. Cosmochim. Acta 3, 253 (1953). Comparative measurements of standards for carbon isotopes.
- 53EHR1 H.F. Ehrenberg, Z. Phys. 134, 317 (1953). Isotopenanalysen an Blei aus Mineralen.
- 53FAR1 R.M. Farquhar, G.H. Palmer, and K.L. Aitken, Nature 172, 860 (1953). A Comparision of Lead Isotope Analysis Techniques.
- 53FRI1 L. Friedman and A.P. Irsa, J. Am. Chem. Soc. 75, 5741 (1953). Ruthenium Isotope Abundances.
- 53HER1 L.F. Herzog, L.T. Aldrich, W.K. Holyk, F.B. Whiting, and L.H. Ahrens, Trans. Am. Geophys. Union 34, 461 (1953). Variations in Strontium Isotopic Abundance in Minerals. Part II: Radiogenic ^{87}Sr in Biotite, Feldspar, and Celestite.
- 53REY1 J.H. Reynolds, Phys. Rev. 90, 1047 (1953). The Isotopic Constitution of Silicon, Germanium, and Hafnium.
- 53SIT1 J.R. Sites, G. Consolazio, and R. Baldock, Phys. Rev. 92, 1096 (1953). Isotopic Abundance of Palladium.
- 53WAL1 W.H. Walker and H.G. Thode, Phys. Rev. 90, 447 (1953). Relative Abundances and Neutron Capture Cross-Sections of the Neodymium Isotopes.
- 53WIC1 F.E. Wickman, Geochim. Cosmochim. Acta 3, 244 (1953). Wird das Haeufigkeitsverhaeltnis der Kohlenstoffisotopen bei der Inkohlung Veraendert?
- 54BAL1 R. Baldock, ORNL 1719 (1954). ORNL Status and Progress Report, April 1954.
- 54BEG1 F. Begemann, J.G. Eiss, F.G. Houtermans, and W. Buser, Il Nuovo Cim. 11, 663 (1954) Isotopenzusammensetzung und Radioaktivitaet von rezentem Vesuvblei.
- 54CLA1 G.R. Clarke, W.H. Denton, and P. Reynolds, Nature 174, 469 (1954). Determination of the Absolute Concentration of Deuterium in Thames River Water.
- 54EHR1 H.F. Ehrenberg and G. Horlitz, Z. Naturforsch. 9a, 951 (1954) Weitere Isotopen-Analysen an Bleierzen.
- 54GEI1 J. Geiss, Z. Naturforsch. 9a, 218 (1954). Isotopenanalysen an "gewoehnlichem Blei".
- 54HER1 L.F. Herzog, NYO-3934-2 (1954). Variations in Isotopic Abundances of Strontium, Calcium, and Argon and Related Topics, Annual Progress Report for 1953-54.
- 54HOG1 J.E. Hogg, Can. J. Chem. 32, 1039 (1954). The Mass Spectrum of Titanium Tetrachloride.
- 54RUS1 R.D. Russell, R.M. Farquhar, G.L. Cumming, and J. Tuze Wilson, Trans. Am. Geophys. Union 35, 301 (1954). Dating Galenas by Means of Their Isotopic Constitutions.
- 55BOY1 A.W. Boyd, F. Brown, and M. Lounsbury, Can. J. Phys. 33, 35 (1955). Mass Spectrometric Study of Natural and Neutron-Irradiated Chlorine.
- 55CAM1 A.F. Cameron, J. Am. Chem. Soc. 77, 2731 (1955). Variation in the Natural Abundance of the Lithium Isotopes.
- 55CAM2 A.E. Cameron, and E.L. Lippert, Science 121, 136 (1955). Isotopic Composition of Bromine in Nature.
- 55DIB1 V.H. Dibeler, Anal. Chem. 27, 1958 (1955). Isotope Analysis Using Dimethylmercury.
- 55EBE1 P. Eberhardt, J. Geiss, and F.G. Houtermans, Z. Phys. 141, 91 (1955). Isotopenverhaeltnisse von "gewoehnlichem Blei" und ihre Deutung.
- 55LAN1 S. Landergren, Geochim. Cosmochim. Acta 7, 240 (1955). A note on the isotope ratio $^{12}\text{C}/^{13}\text{C}$ in metamorphosed alum shale.
- 55OWE1 H.R. Owen, and O.A. Schaeffer, J. Am. Chem. Soc. 77, 898 (1955). The Isotope Abundances of Chlorine from Various Sources.
- 55SAK1 H. Sakai, M. Honda, and E. Minami, Bull. Chem. Soc. Japan 28, 533 (1955). Isotopic Composition of Common Lead in Japan.
- 55WHI1 F.A. White, T.L. Collins Jr., and F.M. Rourke, Phys. Rev. 97, 566 (1955). New Naturally Occurring Isotope of Tantalum.
- 56AKI1 P.A. Akishin, G.M. Panchenkova, N.N. Vasilev, and O.T. Nikitin, Zhur. Fiz. Khim. SSSR 30, 1387 (1956). Isotopic Analysis of Ca, Sr, and Ba with the Aid of Aluminum Ion.
- 56BRA1 P. Bradt, F.L. Mohler, and V.H. Dibeler, J. Res. Nat. Bur. Stand. 57, 223 (1956). Mass Spectrum of Sulfur Vapor.
- 56HES1 D.C. Hess, ANL-5420, 150 (1956). A Search for Rare Naturally Occurring Isotopes.
- 56KUL1 J.L. Kulp, W.U. Ault, and H.W. Feely, Econ. Geol. 51, 139 (1956). Sulfur Isotope Abundances in Sulfide Minerals.
- 56LOU1 M. Lounsbury, Can. J. Chem. 34, 259 (1956). The Natural Abundances of the Uranium Isotopes.
- 56ORD1 K. Ordzhonikidze, and V. Shiutse, Sov. Phys. Jetp 2, 396 (1956). Investigation of the Isotopic Constitution of Lithium.
- 56PAN1 G.M. Panchenkova, P.A. Akishin, N.N. Vasilev, O.T. Nikitin, and S.D. Moiseev, Zhur. Fiz. Khim. 30, 1380 (1956). Isotopic Analysis of Alkali Elements with the Aid of Synthetic Alumino Silicate Ion Sources.
- 56REU1 C. Reutersward, Arkiv Fysik 11, 1 (1956). On the isotopic composition of potassium.
- 56SHI1 V. Shiutse, Sov. Phys. Jetp 2, 402 (1956). An Investigation of the Isotopic Constitution of Boron.
- 56WHI1 F.A. White, T.L. Collins Jr., and F.M. Rourke, Phys. Rev. 101, 1786 (1956). Search for Possible Naturally Occurring Isotopes of Low Abundance.
- 57AIT1 K.L. Aitken, D.J. Littler, E.E. Lockett, and G.H. Palmer, J. Nucl. Energy 4, 33 (1957). The Pile-Neutron Absorption Cross-Section of ^{149}Sm .
- 57COL1 T.L. Collins, F.M. Rourke, and F.A. White, Phys. Rev. 105, 196 (1957). Mass Spectrometric Investigation of the Rare Earth Elements for the Existence of New Stable Isotopes.
- 57CRA1 H. Craig, Geochim. Cosmochim. Acta 12, 133 (1957). Isotopic standards for carbon and oxygen and correction factors for mass-spectrometric analysis of carbon dioxide.
- 57GAV1 S. Gavelin, Geochim. Cosmochim. Acta 12, 297 (1957). Variations in isotopic composition of carbon from metamorphic rocks in northern Sweden and their geological significance.
- 57HES1 D.C. Hess, R.R. Marshall, and H.C. Urey, Science 126, 1291 (1957). Surface Ionization of Silver; Silver in Meteorites.
- 57RUS1 R.D. Russell, and R.M. Farquhar, Mining Eng. 9, 556 (1957). Isotopic Constitutions and Origins of Lead Ores.
- 58BEN1 P.G. Bentley, and A.N. Hamer, Nature 182, 1156 (1958). Boron-10 Abundance in Nature.
- 58DRA1 H.W. Drawin, Nukleonik 1, 109 (1958). Isotopenhaeufigkeitsbestimmungen von Titan und Zirkon.
- 58EHR1 H.F. Ehrenberg, and H.J. Murtz, Z. Naturforsch. 13a, 854 (1958). Isotopen-Zusammensetzung einiger Bleiglanze.
- 58JUN1 G. Junk, and H.J. Svec, Geochim. Cosmochim. Acta 14, 234 (1958). The absolute abundance of the nitrogen isotopes in the atmosphere and compressed gas from various sources.
- 58OMU1 I. Omura, and N. Morito, J. Phys. Soc. Japan 13, 659 (1958). On the Measurement of Isotope Abundance of Lithium with Mass Spectrometer.

58PAL1	G.H. Palmer, J. Nucl. Energy 7 , 1 (1958). The Thermal-Emission Ion Source in Solid Source Mass Spectrometry.	62HAR1	G. Harms, C. Blanc, L. Espagno, and D. Blanc, Comptes Rendus 255 , 3203 (1962). Variations de la composition isotopique du potassium de différentes origines.
58PIN1	W.H. Pinson Jr., L.F. Herzog, H.W. Fairbairn, and R.F. Cormier, Geochim. Cosmochim. Acta 14 , 331 (1958) Sr/Rb age study of tektites.	62MUR1	V.R. Murthy, Geochim. Cosmochim. Acta 26 , 481 (1962). The isotopic composition of silver in iron meteorites.
58SAK1	H. Sakai, and K. Sato, Geochim. Cosmochim. Acta 15 , 1 (1958). Isotopic composition of the common lead of Japan.	62PUP1	J. Pucezin, M. Ceric, and D. Larasevic, Bull. Boris Kidrich Inst. Nucl. Sci. 13 , 77 (1962). The Isotopic Analysis of Lithium on a Mass Spectrometer.
58TIL1	G.R. Tilton, Geochim. Cosmochim. Acta 14 , 323 (1958). Isotopic composition of lead from tektites.	62RID1	B.F. Rider, J.P. Peterson Jr., and C.P. Ruiz, GEAP-4008 (1962). The Half-Life and Gamma Ray Abundance of ¹³⁷ Cs.
58WAL1	E.C. Walker, F. Cuttita, and F.E. Senftle, Geochim. Cosmochim. Acta 15 , 183 (1958). Some natural variations in the relative abundance of copper isotopes.	62SHI1	W.R. Shields, F.L. Garner, and V.H. Dibeler, J. Res. Natl. Bur. Stand. 66a , 1 (1962). Absolute Isotopic Abundance of Terrestrial Silver.
58WAN1	H. Wanke, and H. Hintenberger, Z. Naturforsch. 13a , 895 (1958). Helium und Neon als Reaktionsprodukte der Hohenstrahlung in Eisenmeteoriten.	62SHI2	W.R. Shields, T.J. Murphy, E.L. Garner, and V.H. Dibeler, J. Am. Chem. Soc. 84 , 1519 (1962). Absolute Isotopic Abundance Ratios and the Atomic Weight of Chlorine.
59CRO1	E.A.C. Crouch, E.R. Preece, I.G. Swainbank, and A.H. Turnbull, Nature 184 , 358 (1959). Atomic Weight of Silver.	62STA1	H. Stauffer, and M. Honda, J. Geophys. Res. 67 , 3503 (1962). Cosmic-Ray-Produced Stable Isotopes in Iron Meteorites.
59FER1	G. Ferrara, D. Ledent, and H. Stauffer, Helv. Phys. Acta 32 , 279 (1959). Blei-Isotopenverhältnisse sedimentärer Uranvorkommen in der Schweiz und in Italien.	62SVE1	H.J. Svec, G.D. Flesch, and J. Capellen, Geochim. Cosmochim. Acta 26 , 1351 (1962). The absolute abundance of the chromium isotopes in some secondary minerals.
59OKA1	J. Okamoto, M. Kakuta, N. Morito, Y. Nakajima, H. Tsuyama, and H. Onuki, Japan Analyst (Bunseki Kagaku) 8 , 445 (1959). Isotopic Analysis of Zinc with Mass Spectrometer.	62TAN2	S. Taniguchi, O. Toyama, and T. Hayakawa, Mass Spectroscopy (Shitsuryo Bunseki) 21 , 100 (1962). Relative Measurements of Lithium Isotope Ratio by Surface Ionization Ion Source II. Lithium Carbonate and Lithium Nitrate Prepared from Laboratory Standard Lithium Sulfate (Merck).
60BEN1	P.G. Bentley, J. Sci. Instr. 37 , 323 (1960). Isotope analysis of boron in boron trifluoride by mass spectrometry, and measurement of natural boron 10 concentration.	62UME1	S. Umemoto, J. Geophys. Res. 67 , 375 (1962). Isotopic Composition of Barium and Cerium in Stone Meteorites.
60CAT1	E.J. Catanzaro, and P.W. Gast, Geochim. Cosmochim. Acta 19 , 113 (1960). Isotopic composition of lead in pegmatitic feldspars.	62WHI1	F.A. White, J.C. Sheffield, and F.M. Rourke, J. Appl. Phys. 33 , 2915 (1962). Isotopic Abundance Determinations of Copper by Sputtering.
60FLE1	G.D. Flesch, H.J. Svec, and H.G. Staley, Geochim. Cosmochim. Acta 20 , 300 (1960). The absolute abundance of the chromium isotopes in chromite.	63BIE1	P.J. De Bièvre, G.H. Debus, and J. Spaepen, Reactor Sci. Technol. (J.Nucl.Energy Parts A/B) 17 , 439 (1963). Thermal Neutron Absorption Cross Section of Boron.
60HOR1	Y. Horibe, and M. Kobayakawa, Geochim. Cosmochim. Acta 20 , 273 (1960). Deuterium abundance of natural waters.	63LEI1	F.D. Leipziger, Appl. Spec. 17 , 158 (1963). Some New Upper Limits of Isotopic Abundance By Mass Spectrometry.
60KEN1	B.R.F. Kendall, Nature 186 , 225 (1960). Isotopic Composition of Potassium.	63MUR1	V.R. Murthy, Geochim. Cosmochim. Acta 27 , 1171 (1963). Elemental and isotopic abundances of molybdenum in some meteorites.
60KOL1	F. Kollar, R.D. Russell, and T.J. Ulrych, Nature 187 , 754 (1960). Precision Intercomparisons of Lead Isotope Ratios: Broken Hill and Mount Isa.	63PIL1	J. Pilot, Kernenergie 6 , 714 (1963). Ueber die massenspektrometrische Isotopenanalyse an Stickstoff aus Erdgassen und Gesteinen.
60MUR1	V.R. Murthy, Phys. Rev. Lett. 5 , 539 (1960). The Isotopic Composition of Silver in an Iron Meteorite.	63RIC1	J.R. Richards, Geochim. Cosmochim. Acta 27 , 217 (1963). Isotopic Composition of Australian Leads. Northwestern Queensland and the Northern Territory — A Reconnaissance.
60OMU1	I. Omura, N. Morito, Y. Nakajima, J. Okamoto and H. Tsuyama, Mass Spectroscopy (Shitsuryo Bunseki) "14," 56 (1960). Measurements of Natural Isotopic Abundances by Surface Ionization Method.	63SHI1	W.R. Shields, E.L. Garner, C.E. Hedge, and S.S. Goldich, J. Geophys. Res. 68 , 2331 (1963). Survey of ⁸⁵ Rb/ ⁸⁷ Rb Ratios in Minerals.
60PAL1	G.H. Palmer, NBS Technical Note 51 (1960). Isotopic Abundance Ratios Reported for Reference Samples Stocked by NBS.	63SHI2	M. Shima, J. Geophys. Res. 68 , 4289 (1963). Isotopic Composition of Germanium in Meteorites.
60SHI1	W.R. Shields, D.N. Craig, and V.H. Dibeler, J. Am. Chem. Soc. 82 , 5033 (1960). Absolute Isotopic Abundance Ratios and the Atomic Weight of Silver.	63SHI3	M. Shima, Geochim. Cosmochim. Acta 27 , 911. Geochemical study of boron isotopes.
61FIN1	H.O. Finley, E.E. Leuang Jr., NBL-170 49 (1961). Mass Spectrometric Determination of Boron Isotopic Abundance in the Near Normal Range.	63SHI4	M. Shima, and M. Honda, J. Geophys. Res. 68 , 2849 (1963). Isotopic Abundance of Meteoritic Lithium.
61GOR1	P. Goris, T.D. Morgan, and R.A. Nielsen, IDO-14549 (1961). Comparative Boron Isotopic Analysis.	63SVE1	H.J. Svec, J. Capellen, G. Flesch, G. Junk, and F. Saalfeld, IS-700, Sect.C 41 (1963). Absolute Abundance of the Vanadium Isotopes.
61MCM1	C.C. McMullen, C.B. Cragg, and H.G. Thode, Geochim. Cosmochim. Acta 23 , 147 (1961). Absolute Ratio of ¹¹ B/ ¹⁰ B in Seares Lake Borax.	64BAC1	M.M. Backus, W.H. Pinson, L.F. Herzog, and P.M. Hurley, Geochim. Cosmochim. Acta 28 , 735 (1964). Calcium isotope ratios in the Homestead and Pasamonte meteorites and a Devonian limestone.
61MEY1	S. Meyerson, Anal. Chem. 33 , 964 (1961). Natural Abundance of Chlorine Isotopes.	64CAT1	E.J. Catanzaro, T.J. Murphy, E.L. Garner, and W.R. Shields, J. Res. Natl. Bur. Stand. 68a , 593 (1964). Absolute Isotopic Abundance Ratio and the Atomic Weight of Bromine.
62CRO1	E.A.C. Crouch, and A.H. Turnbull, J. Chem. Soc. 31 , 161 (1962). The Absolute Isotopic Abundance Ratio and the Atomic Weight of Natural Silver.		
62DOE1	B.R. Doe, J. Geophys. Res. 67 , 2895 (1962). Relationships of Lead Isotopes among Granite, Pegmatites, and Sulfide Ores near Balmat, New York.		

64CHE2	J. Chenouard, Advances in Mass Spectrometry 5, 583 (1964). Determinations d'abondances isotopiques de Fer.	66TAT1	M. Tatsumoto, J. Geophys. Res. 71, 1721 (1966). Isotopic Composition of Lead in Volcanic Rocks from Hawaii, Iwo Jima, and Japan.
64CLA1	W.R. Clarke, and H.G. Thode, J. Geophys. Res. 69, 3673 (1964). The Isotopic Composition of Krypton in Meteorites.	66WAL1	J.R. Walton, and A.E. Cameron, Z. Naturforsch. 21a, 115 (1966). The Isotopic Composition of Atmospheric Neon.
64CRO1	E.A.C. Crouch, and T.A. Tuplin, Nature 202, 1282 (1964). Isotopic Composition and the Atomic Weight of Naturally Occurring Molybdenum: A Possible Reflexion of the Creation Process.	67KAN1	T. Kanazaki, S. Yokozuka, and H. Kakihana, Japan Analyst (Bunseki Kagaku) 16, 7 (1967). Determination of the Abundance Ratio of Copper Isotopes by a Surface Ionization Method.
64FLE1	G.D. Flesch, J. Capellen, and H.J. Svec, IS-921 (1964). The Abundance of the Vanadium Isotopes from Sources of Geochemical Interest.	67TAK1	N. Takematsu, S. Matsuo, and S. Sato, Geochem. J. 1, 51 (1967). Isotopic Composition of Magnesium in Upper Mantle Materials and a Meteorite.
64KOM1	T. Komori, S. Tamura, M. Ouchi, K. Gunji, H. Hashitani and H. Yoshida, Japan Analyst (Bunseki Kagaku) 13, 32 (1964). Determination of Neodymium, Samarium and Europium. (Determination of Rare Earth Elements in Nuclear Fuel Materials by Isotope Dilution Method. I.).	68AGY1	E.K. Agyei, and C.C. McMullen, Can. J. Earth. Sci. 5, 921 (1968) (see also: E.K. Agyei, Ph.D. Thesis (Physics), McMaster Univ., (1968). A Study of the Isotopic Abundance of Boron from Various Sources.
64KRA1	D. Krankowsky, and O. Muller, Geochim. Cosmochim. Acta 28, 1625 (1964). Isotopenhaeufigkeit und Konzentration des Lithiums in Steinmeteoriten.	68BEL1	H.A. Belsheim, IS-T-217 (1968). The Absolute Abundance of the Titanium Isotopes in Nature.
64SHI1	W.R. Shields, T.J. Murphy, and E.L. Garner, J. Res. Natl. Bur. Stand. 68a, 589 (1964). Absolute Isotopic Abundance Ratios and the Atomic Weight of a Reference Sample of Copper.	68CAT1	E.J. Catanzaro, T.J. Murphy, W.R. Shields, and E.L. Garner, J. Res. NBS 72a, 261 (1968). Absolute Isotopic Abundance Ratios of Common, Equal-Atom, and Radioactive Lead Isotopic Standards.
64SHI2	M. Shima, Bull. Chem. Soc. Japan 37, 284 (1964). The Isotopic Composition of Magnesium in Terrestrial Samples.	68IMA1	M. Imamura, M. Shima, and M. M. Honda, Mass Spectroscopy (Shitsuryo Bunseki) 16, 291 (1968). Determination of Cosmic-Ray-Produced Titanium and Vanadium in Iron Meteorites.
64WET1	G.W. Wetherill, J. Geophys. Res. 69, 4403 (1964). Isotopic Composition and Concentration of Molybdenum in Iron Meteorites.	68SHI1	M. Shima, M. Imamura, and M. Honda, Mass Spectroscopy (Shitsuryo Bunseki) 16, 277 (1968). Determination of Cosmic-Ray-Produced Potassium and Calcium in Iron Meteorites.
65EBE1	P. Eberhardt, O. Eugster, and K. Marti, Z. Naturforsch. 20a, 623 (1965). A Redetermination of the Isotopic Composition of Terrestrial.	69BAL1	H. Balsiger, J. Geiss, and M.E. Lipschutz, Earth Planet. Sci. Lett. 6, 117 (1969). Vanadium Isotopic Composition in Meteorite and Terrestrial Matter.
65LAE1	J.R. De Laeter and P.M. Jeffery, J. Geophys. Res. 70, 2895 (1965). The Isotopic Composition of Terrestrial and Meteoritic Tin.	69BIE1	P.J. De Bièvre, and G.H. Debus, Int. J. Mass Spectrom. Ion Physics 2, 15 (1969). Absolute Isotopic Ratio Determination of a Natural Boron Standard.
65SHI1	W.R. Shields, S.S. Goldich, E.L. Garner, and T.J. Murphy, J. Geophys. Res. 70, 479 (1965). Natural Variations in the Abundance Ratio and the Atomic Weight of Copper.	69CAT1	E.J. Catanzaro, T.J. Murphy, E.L. Garner, and W.R. Shields, J. Res. NBS 73a, 511 (1969). Absolute Isotopic Abundance Ratios and Atomic Weight of Terrestrial Rubidium.
65SHI2	M. Shima, J. Mineral. Petrol. Econ. Geol. 53, 228 (1965). The geochemical studies of stable isotopes on iron and nickel.	69BUG1	O. Eugster, F. Tera, and G.J. Wasserburg, J. Geophys. Res. 74, 3897 (1969). Isotopic Analyses of Barium in Meteorites and in Terrestrial Samples.
65SVE1	H.J. Svec, and A.R. Anderson, Jr., Geochim. Cosmochim. Acta 29, 633 (1965). The absolute abundance of the lithium isotopes in natural sources.	69GRA1	S. Graeser, and J.C. Huziker, Geochim. Int. 6, 983 (1969). Rb-Sr, and Pb Isotope Determination on Rock and Minerals of the Ivrea Zone.
66CAT1	E.J. Catanzaro, T.J. Murphy, E.L. Garner, and W.R. Shields, J. Res. NBS 70a, 453 (1966). Absolute Isotopic Abundance Ratios and the Atomic Weight of Magnesium.	69SMI1	R.F. Smith, and J.M. Jackson, Union Carbide Corporation KY-581 (1969). Variations in U-234 Concentration of Natural Uranium.
66CAT2	E.J. Catanzaro, and T.J. Murphy, J. Geophys. Res. 71, 1271 (1966). Magnesium Isotope Ratios in Natural Samples.	70CAT1	E.J. Catanzaro, C.E. Champion, E.L. Garner, G. Marinenko, K.M. Sappenfield, and W.R. Shields, Standard Reference Materials: Boric Acid; Isotopic and Assay Standard Reference Materials NBS Special Publication 260-17, US Printing Office, (1970).
66FLE1	G.D. Flesch, J. Capellen and H.V. Svec, Advances in Mass Spectrometry 3, 571 (1966). The Abundance of the Vanadium Isotopes from Sources of Geochemical Interest.	70Eug1	O. Eugster, F. Tera, D.S. Burnett, and G.J. Wasserburg, J. Geophys. Res. 75, 2735 (1970). Isotopic Composition of Gadolinium and Neutron-Capture Effects in Some Meteorites.
66KOM1	T. Komori, H. Yoshida, K. Gunji, K. Toida, and S. Tamura, Japan Analyst (Bunseki Kagaku) 15, 589 (1966). Determination of Cerium, Gadolinium, Dysprosium, Erbium, and Ytterbium. (Determination of Rare Earth Elements in Nuclear Fuel Materials by Isotope Dilution Method. II.).	70HAG1	R. Hagemann, G. Nief, and E. Roth, Tellus 22, 712 (1970). Absolute isotopic scale for deuterium analysis of natural waters. Absolute D/H ratio for SMOW.
66MAS1	M. Shima, and M. Honda, Mass Spectroscopy (Shitsuryo Bunseki) "14", 23 (1966). Isotopic Measurement of Chromium Induced by Cosmic Rays in Iron Meteorites.	70MAM1	B.A. Mamyrin, G.S. Anufriev, I.L. Kamenskiy, and I.N. Tolstikhin, Geokhimiya "6", 721 (1970). Determination of the Isotopic Composition of Atmospheric Helium.
66SHI1	W.R. Shields, T.J. Murphy, E.J. Catanzaro, and E.L. Garner, J. Res. NBS 70a, 193 (1966). Absolute Isotopic Abundance Ratios and the Atomic Weight of a Reference Sample of Chromium.	70PEL1	I.Z. Pelly, M.E. Lipschutz, and H. Balsiger, Geochim. Cosmochim. Acta 34, 1033 (1970). Vanadium isotopic composition and contents in chondrites.
66SHI2	M. Shima and M. Honda, Geochem. J. 1, 27 (1966). Distribution and Isotopic Composition of Lithium in Stone Meteorites.	70SCH1	D.N. Schramm, F. Tera, and G.J. Wasserburg, Earth Planet. Sci. Lett. 10, 44 (1970). The Isotopic Abundance of ^{26}Mg and Limits on ^{26}Al in the Early Solar System.

- 70TAK1 N. Takaoka, Mass Spectroscopy (Shitsuryo Bunseki) **18**, 888 (1970). Preliminary Results on Chemical Separation of Ruthenium from Molybdenum Minerals by Distillation and Isotope Analysis of Ruthenium by Surface Ionization.
- 71COL1 M.L. Coleman, Earth Planet. Sci. Lett. **12**, 399 (1971). Potassium-Calcium Dates from Pegmatitic Micas.
- 71GEN1 R. Gensho, and M. Honda, Mass Spectroscopy (Shitsuryo Bunseki) **21**, 134 (1971). Measurements of the Isotopic Ratio of Boron.
- 71MEL1 C.E. Melton, W. Massey, and B.N. Abels, Z. Naturforsch. **26a**, 1241 (1971). The Isotopic Abundance of Neon, Argon and Krypton.
- 71POD1 F.A. Podosek, J.C. Huneke, D.S. Burnett, and G.J. Wasserburg, Earth Planet. Sci. Lett. **10**, 199 (1971). Isotopic Composition of Xenon and Krypton in the Lunar Soil and in the Solar Wind.
- 72LAE1 J.R. De Laeter, Geochim. Cosmochim. Acta **36**, 735 (1972). The isotopic composition and elemental abundance of gallium in meteorites and in terrestrial samples.
- 72LOV1 A.J. Loveless, S. Yanagita, H. Mabuchi, M. Ozima, and R.D. Russell, Geochim. Cosmochim. Acta **36**, 685 (1972). Isotopic Ratios of Gd, Sm and Eu in "Abee" Enstatite Chondrite.
- 72MAS1 A. Masuda, S. Yanagita, H. Mabuchi, K. Notsu, and M. Ojima, LUNAR SCIENCE-III, Revised Abstracts of Papers Presented at the Third Lunar Science Conference Houston 10-13 January 1972, C. Watkins (ed), Lunar Science Institute, Houston (1972). Lanthanum 138/139 Ratio in Lunar Sample 14163.
- 72MOO1 L.J. Moore, and L.A. Machlan, Anal. Chem. **44**, 2291 (1972). High Accuracy Determination of Calcium in Blood Serum by Isotope Dilution Mass Spectrometry.
- 72ROS1 K.J.R. Rosman, Geochim. Cosmochim. Acta **36**, 801 (1972). A survey of the isotopic and elemental abundance of zinc.
- 72STE1 C.M. Stevens, Int. J. Mass Spectrom. Ion Phys. **8**, 251 (1972). The Isotopic Abundance of Molybdenum in Terrestrial Minerals.
- 73BAR1 I.L. Barnes, E.L. Garner, J.W. Gramlich, L.A. Machlan, J.R. Moody, L.J. Moore, T.J. Murphy and W.R. Shields, Proc. 4th Lunar Sci. Conf.; Suppl.4, Geochim. Cosmochim. Acta **2**, 1197 (1973). Isotopic abundance ratios and concentrations of selected elements in some Apollo-15 and Apollo-16 samples.
- 73FLE1 G.D. Flesch, A.R. Anderson, and H.J. Svec, Int. J. Mass Spectrom. Ion Phys. **12**, 265 (1973). A Secondary Isotopic Standard for $^{6}\text{Li}/\text{Li}$ Determinations.
- 73GRA1 J.W. Gramlich, T.J. Murphy, E.L. Garner, and W.R. Shields, J. Res. NBS **77a**, 691 (1973). Absolute Isotopic Abundance Ratio and Atomic Weight of a Reference Sample of Rhodium.
- 73LAE1 J.R. De Laeter, and R. Date, Int. J. Mass Spectrom. Ion Phys. **12**, 455 (1973). The Isotopic Composition of Barium.
- 73NOM1 M. Nomura, M. Okamoto, and H. Kakihana, Mass Spectroscopy (Shitsuryo Bunseki) **21**, 277 (1973). Determination of Boron Isotope Ratio by the Surface Ionization Method.
- 73TAM1 S. Tamura, Mass Spectroscopy (Shitsuryo Bunseki) **21**, 283 (1973). Determination of Traces of Boron in Heat-Resisting Alloy Standards by Isotope Dilution Mass Spectrometry.
- 73WAL1 J.R. Walton, A.E. Cameron, R.L. Walker, and T.L. Hebble, Int. J. Mass Spectrom. Ion Phys. **12**, 439 (1973). Determination of the Abundance of Krypton in the Earth's Atmosphere by Isotope Dilution Mass Spectrometry.
- 73LEE1 T. Lee, and D.A. Papanastassiou, Geophys. Res. Lett. **1**, 225 (1974). Mg Isotopic Anomalies in the Allende Meteorite and Correlation with O and Sr Effects.
- 74MOO1 L.J. Moore, L.A. Machlan, W.R. Shields, and E.L. Garner, Anal. Chem. **46**, 1082 (1974). Internal Normalization Techniques for High Accuracy Isotope Dilution Analyses - Application to Molybdenum and Nickel in Standard Reference Materials.
- 74SHI2 M. Shima, and H.G. Thode, AECL-5014 110 (1974). Palladium Isotope Abundances.
- 75DAR2 I.L. Barnes, L.J. Moore, L.A. Machlan, T.J. Murphy, and W.R. Shields, J. Res. NBS **79a**, 727 (1975). Absolute Isotopic Abundance Ratios and the Atomic Weight of a Reference Sample of Silicon.
- 75BAR3 I.L. Barnes, W.R. Shields, T.J. Murphy, and R.H. Brill, Adv. Chem. **138**, Archaeological Chemistry, Am. Chem. Soc. (1975). Isotopic Analysis of Laurion Lead Ores.
- 75GAR1 E.L. Garner, T.J. Murphy, J.W. Gramlich, P.J. Paulsen, and I.L. Barnes, J. Res. NBS **79a**, 713 (1975). Absolute Isotopic Abundance Ratios and the Atomic Weight of a Reference Sample of Potassium.
- 75IMA1 K. Imamura, M. Shima and M. Honda, Earth Planet. Sci. Lett. **26**, 54 (1975). Cosmic-Ray Produced ^{40}K and ^{50}V in the Metal Phase of Chondrites.
- 75LUG1 G.W. Lugmair, N.B. Scheinin, and K. Marti, Proc. Lunar Sci. Conf., 6th, Geochim. Cosmochim. Acta Suppl. **62**, 1419 (1975). Sm-Nd Age and History of Apollo 17 Basalt 73075: Evidence for Early Differentiation of the Lunar Exterior.
- 75MUR1 M. Murozumi, and Y. Abe, Japan Analyst (Bunseki Kagaku) **24**, 337 (1975). Isotope Dilution Mass Spectrometry of Copper in Sea Water.
- 75ROS1 K.J.R. Rosman, and J.R. De Laeter, Int. J. Mass Spectrom. Ion Phys. **16**, 385 (1975). The Isotopic Composition of Cadmium in Terrestrial Minerals.
- 75TAM1 S. Tamura, Mass Spectroscopy (Shitsuryo Bunseki) **23**, 49 (1975). Isotopic Analysis of Molybdenum by Surface Ionization Mass Spectrometry and a Carbonization Technique.
- 75YAN1 S. Yanagita, Ph.D. Thesis, Univ. of Tokyo (1975).
- 76BAE1 P. Baertsch, Earth Planet. Sci. Lett. **31**, 341 (1976). Absolute ^{18}O Content of Standard Mean Ocean Water.
- 76CHO1 T.J. Chow, Science **193**, 57 (1976). Barium in Southern California Coastal Waters: A Potential Indicator of Marine Drilling Contamination.
- 76CLA1 W.B. Clarke, W.J. Jenkins, and Z. Top, Int. J. Appl. Radiat. Isotopes **27**, 515 (1976). Determination of Tritium by Mass Spectrometric Measurement of ^{3}He .
- 76COW1 G.A. Cowan, Sci. Am. **235**, 36 (1976). A Natural Fission Reactor.
- 76DEV1 C. Devillers T. Lecomte, M. Lucas, and R. Hagemann, Proc. 7th Int. Mass Spectrom. Conf. Florence 553 (1976). Mass Spectrometric Investigations on Ruthenium Isotopic Abundances.
- 76LAE1 J.R. De Laeter, and K.J.R. Rosman, Int. J. Mass Spectrom. Ion Phys. **21**, 403 (1976). The Atomic Weight of Gallium.
- 76MCC1 M.T. McCulloch, J.R. De Laeter, and K.J.R. Rosman, Earth Planet. Sci. Lett. **28**, 308 (1976). The Isotopic Composition and Elemental Abundance of Lutetium in Meteorites and Terrestrial Samples and the ^{176}Lu Cosmochronometer.
- 76MUR1 M. Murozumi, S. Nakamura, and K. Ito, Japan Analyst (Bunseki Kagaku) **25**, 706 (1976). Isotope Dilution Mass Spectrometry of Copper in Sea Water (I).
- 76NAK1 N. Nakamura, M. Tatsumoto, P.D. Nunes, D.M. Unruh, A.P. Schwab, and T.R. Wildeman, Proc. Lunar Sci. Conf. 7th 2309 (1976). 4.4 B.Y.-old Clast in Boulder 7, Apollo 17: A Comprehensive Chronological Study by U-Pb, Rb-Sr and Sm-Nd Methods.
- 77BRO1 H.L. Brown, C. Biltz, and M. Anbar, Int. J. Mass Spectrom. Ion Phys. **25**, 167 (1977). A Precision Isotope Ratio Mass Spectrometer for the Analysis of $^{6}\text{Li}/\text{Li}$.

77LEE1	T. Lee, D.A. Papanastassiou, and G.J. Wasserburg, Geochim. Cosmochim. Acta 41 , 1473 (1977). Mg and Cu Isotopic Study of Individual Microscopic Crystals from the Allende Meteorite by the Direct Loading Technique.	81MER1	N. Mermelengas, K.J.R. Rosman, and J.R. De Laeter, Int. J. Mass Spectrom. Ion Phys. 37 , 1 (1981). The Isotopic Composition of Palladium in Meteorites and Terrestrial Samples.
77MCC1	M.T. McCulloch, K.J.R. Rosman, and J.R. De Laeter, Geochim. Cosmochim. Acta 41 , 1703 (1977). The Isotopic and Elemental Abundance of Ytterbium in Meteorites and Terrestrial Samples.	81MIN1	J.F. Minster, L.Ph. Ricard, Int. J. Mass Spectrom. Ion Phys. 37 , 259 (1981). The Isotopic Composition of Zirconium.
78KEL1	W.R. Kelly, and G.J. Wasserburg, Geophys. Res. Lett. 5 , 1079 (1978). Evidence for the Existence of Palladium-107 in the Early Solar System.	81MUR1	M. Murozumi, S. Nakamura, and K. Suga, J. Chem. Soc. Japan "3", 385 (1981). Isotope Dilution Surface Ionization Mass Spectrometry of Silver in Environmental Materials.
78MUR1	M. Murozumi, S. Nakamura, T. Kato, T. Igarashi, and H. Tsubota, J. Chem. Soc. Japan "2", 226 (1978). Determination of Cadmium by Isotope Dilution-Surface Emission Mass Spectrometry.	81NIE1	S. Niemeyer, and G.W. Lugmair, Earth Planet. Sci. Lett. 53 , 211 (1981). Ubiquitous Isotopic Anomalies in Ti from Normal Allende Inclusions.
78MUR2	M. Murozumi, S. Nakamura, and T. Igarashi, J. Chem. Soc. Japan "11", 1515 (1978). Isotope Dilution-Surface Ionization Mass Spectrometry of Thallium.	82MOO1	L.J. Moore, T.J. Murphy, I.L. Barnes, and P.J. Paulsen, J. Res. NBS 87 , 1 (1982). Absolute Isotopic Abundance Ratios and Atomic Weight of a Reference Sample of Strontium.
78RUS1	W.A. Russell, D.A. Papanastassiou, and T.A. Tombrello, Geochim. Cosmochim. Acta 42 , 1073 (1978). Ca isotope fractionation on the Earth and other solar system materials.	82POW1	L.J. Powell, T.J. Murphy, and J.W. Gramlich, J. Res. NBS 87 , 9 (1982). The Absolute Isotopic Abundance and Atomic Weight of a Reference Sample of Silver.
78SHI1	M. Shima, C.E. Rees, and H.G. Thode, Can. J. Phys. 56 , 1333 (1978). The isotopic composition and atomic weight of palladium.	82SHI1	T. Shimamura, and G.W. Lugmair, Abstracts of the Lunar Planetary Science Conference XIII 722 (1982). Ni Isotopic Compositions in Terrestrial and Meteoritic Samples.
78SHI2	M. Shima, Int. J. Mass Spectrom. Ion Phys. 28 , 129 (1978). The Isotopic Composition and the Atomic Weight of Zirconium.	83DEV1	C. Devillers, T. Lecomte, and R. Hagemann, Int. J. Mass Spectrom. Ion Phys. 50 , 205 (1983). Absolute Isotope Abundances of Tin.
78SMI1	C.L. Smith, K.J.R. Rosman, and J.R. De Laeter, Int. J. Mass Spectrom. Ion Phys. 28 , 7 (1978). The Isotopic Composition of Tellurium.	83MIC1	E. Michiels, and P. De Bièvre, Int. J. Mass Spectrom. Ion Phys. 49 , 265 (1983). Absolute Isotopic Composition and the Atomic Weight of a Natural Sample of Lithium.
78YAM1	K. Yamasaki, M. Murozumi, S. Nakamura, M. Hinata and M. Yuasa, J. Chem. Soc. Japan "8", 1112 (1978). Lead Isotope Ratios in Japanese Galena Ores and Archaeological Objects.	83NOM1	M. Nomura, K. Kogure, and M. Okamoto, Int. J. Mass Spectrom. Ion Phys. 50 , 219 (1983). Isotopic Abundance Ratios and Atomic Weight of Zirconium.
79ESA1	T.M. Esat, D.E. Brownlee, D.A. Papanastassiou, and G.J. Wasserburg, Science 206 , 190 (1979). Magnesium Isotopic Composition of Interplanetary Dust Particles.	83PAT1	P.J. Patchett, Geochim. Cosmochim. Acta 47 , 81 (1983). Importance of the Lu-Hf Isotopic System in Studies of Planetary Chronology and Chemical Evolution.
79HEY1	H.R. Haydegger, J.J. Foster, and W. Compston, Nature 278 , 704 (1979). Evidence of a new Isotopic Anomaly from Titanium Isotopic Ratios in Meteoric Materials.		
79KAN1	T. Kanzaki, M. Yoshida, M. Nomura, H. Kakihana, and T. Ozawa, Geochim. Cosmochim. Acta 43 , 1859 (1979). Boron Isotopic Composition of Fumarolic Condensates and Sulfates from Satsuma, Iwo-jima, Japan.	1961	Report of the International Commission on Atomic Weights (1961), J. Am. Chem. Soc. 84 , 4175 (1962).
80DUN1	L.P. Dunstan, J.W. Gramlich, I.L. Barnes, and W.C. Purdy, J. Res. NBS 85 , 1 (1980). Absolute Isotopic Abundance and the Atomic Weight of a Reference Sample of Thallium.	1963	No changes were made in 1963.
80IMA1	M. Imamura, M. Shima, and M. Honda, Z. Naturforsch. 35a , 267 (1980). Radial Distribution of Spallogenic K, Ca, Ti, V and Mn Isotopes in Iron Meteorites.	1965	The changes recommended in 1965 are contained in the 1967 Report cited below.
80MOR1	Ph. Morand, C.J. Allegre, and J. Audouze, Meteoritics 15 , 334 (1980). Search for Nickel Isotopic Anomaly of Meteorites.	1967	International Commission on Atomic Weights, Final Version of the Report 28.9.1967, International Union of Pure and Applied Chemistry, Compt. Rend. XXIV Conference Prague, 4 to 10 September 1967, pages 130 - 141.
80NIE1	F.R. Niederer, D.A. Papanastassiou, and G.J. Wasserburg, Astrophys. J. 240 , 173 (1980). Endemic Isotopic Anomalies in Titanium.	1969	Atomic Weights of the Elements 1969, Pure Appl. Chem. 21 , 93 (1970).
80ROS1	K.J.R. Rosman, I.L. Barnes, L.J. Moore, and J.W. Gramlich, Geochim. J. 14 , 269 (1980). Isotope Composition of Cd, Ca and Mg in the Brownfield Chondrite.	1971	Atomic Weights of the Elements 1971, Pure Appl. Chem. 30 , 637 (1972).
81HOL1	P. Holliger, and C. Devillers, Earth Plant. Sci. Lett. 52 , 76 (1981). Contribution à l'Etude de la Température dans les Réacteurs Fossiles d'Oklo par la Mesure du Rapport Isotopique du Lutetium.	1973	Atomic Weights of the Elements 1973, Pure Appl. Chem. 37 , 591 (1974).
		1975	Atomic Weights of the Elements 1975, Pure Appl. Chem. 47 , 75 (1976).
		1977	Atomic Weights of the Elements 1977, Pure Appl. Chem. 51 , 405 (1979).
		1979	Atomic Weights of the Elements 1979, Pure Appl. Chem. 52 , 2349 (1980).
		1981	Atomic Weights of the Elements 1981 and Isotopic Composition of the Elements 1981, Pure Appl. Chem. 55 , 1101 and 1119 (1983).
		1983	Atomic Weights of the Elements 1983 and Isotopic Composition of the Elements 1983, Pure Appl. Chem. 56 , 653 and 675 (1984).

4.1. Commission Reports

Report of the International Commission on Atomic Weights (1961), J. Am. Chem. Soc. **84**, 4175 (1962). No changes were made in 1963. The changes recommended in 1965 are contained in the 1967 Report cited below. International Commission on Atomic Weights, Final Version of the Report 28.9.1967, International Union of Pure and Applied Chemistry, Compt. Rend. XXIV Conference Prague, 4 to 10 September 1967, pages 130 - 141. Atomic Weights of the Elements 1969, Pure Appl. Chem. **21**, 93 (1970). Atomic Weights of the Elements 1971, Pure Appl. Chem. **30**, 637 (1972). Atomic Weights of the Elements 1973, Pure Appl. Chem. **37**, 591 (1974). Atomic Weights of the Elements 1975, Pure Appl. Chem. **47**, 75 (1976). Atomic Weights of the Elements 1977, Pure Appl. Chem. **51**, 405 (1979). Atomic Weights of the Elements 1979, Pure Appl. Chem. **52**, 2349 (1980). Atomic Weights of the Elements 1981 and Isotopic Composition of the Elements 1981, Pure Appl. Chem. **55**, 1101 and 1119 (1983). Atomic Weights of the Elements 1983 and Isotopic Composition of the Elements 1983, Pure Appl. Chem. **56**, 653 and 675 (1984).